

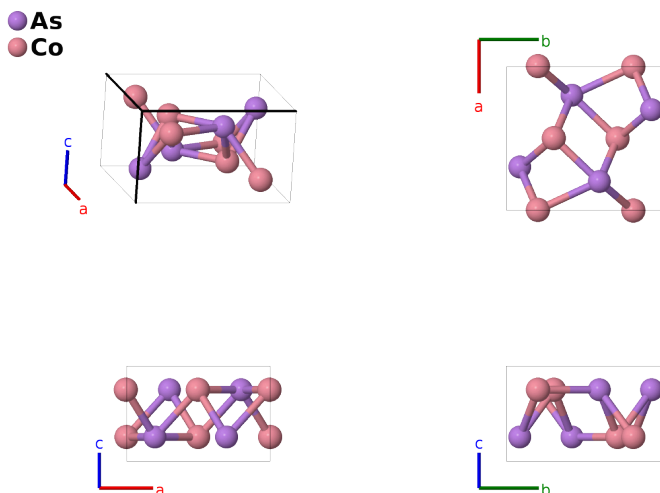
# Modderite (CoAs) Structure: AB\_oP8\_33\_a\_a-001

This structure originally had the label AB\_oP8\_33\_a.a. Calls to that address will be redirected here.

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<https://aflow.org/p/8Z7Q>

[https://aflow.org/p/AB\\_oP8\\_33\\_a\\_a-001](https://aflow.org/p/AB_oP8_33_a_a-001)



<b>Prototype</b>	AsCo
<b>AFLOW prototype label</b>	AB_oP8_33_a.a-001
<b>Mineral name</b>	modderite
<b>ICSD</b>	48027
<b>Pearson symbol</b>	oP8
<b>Space group number</b>	33
<b>Space group symbol</b>	$Pna2_1$
<b>AFLOW prototype command</b>	<pre>aflow --proto=AB_oP8_33_a_a-001       --params=a,b/a,c/a,x1,y1,z1,x2,y2,z2</pre>

## Other compounds with this structure

FeAs

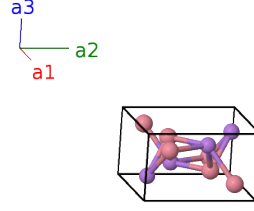
- Space group  $Pna2_1$  #33 allows an arbitrary origin for the  $z$ -axis, which is set here by taking  $z_2 = 1/4$ .
- When  $z_1 = z_2 = 1/4$ , the space group becomes  $Pnma$  #62 and the structure is equivalent to MnP (*B31*).
- (Lyman, 1984) sets  $z_1 = 0.2506$ , so this condition is almost fulfilled.
- (Lyman, 1984) lists both space groups for both CoAs and FeAs, and prefers the MnP structure for these compounds.

- AFLOW also places this structure in space group  $Pnma$ , and only predicts the lower symmetry structure if we lower the native tolerance using
- `aflow --proto=AB_oP8_33_a.a --tolerance=0.001 --params=a,b/c/a,x1,y1,z1,x2,y2,z2 .`

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### Simple Orthorhombic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$




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### Basis vectors

	Lattice coordinates	=	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$x_1 \mathbf{a}_1 + y_1 \mathbf{a}_2 + z_1 \mathbf{a}_3$	=	$ax_1 \hat{\mathbf{x}} + by_1 \hat{\mathbf{y}} + cz_1 \hat{\mathbf{z}}$	(4a)	As I
$\mathbf{B}_2$	$-x_1 \mathbf{a}_1 - y_1 \mathbf{a}_2 + (z_1 + \frac{1}{2}) \mathbf{a}_3$	=	$-ax_1 \hat{\mathbf{x}} - by_1 \hat{\mathbf{y}} + c(z_1 + \frac{1}{2}) \hat{\mathbf{z}}$	(4a)	As I
$\mathbf{B}_3$	$(x_1 + \frac{1}{2}) \mathbf{a}_1 - (y_1 - \frac{1}{2}) \mathbf{a}_2 + z_1 \mathbf{a}_3$	=	$a(x_1 + \frac{1}{2}) \hat{\mathbf{x}} - b(y_1 - \frac{1}{2}) \hat{\mathbf{y}} + cz_1 \hat{\mathbf{z}}$	(4a)	As I
$\mathbf{B}_4$	$-(x_1 - \frac{1}{2}) \mathbf{a}_1 + (y_1 + \frac{1}{2}) \mathbf{a}_2 + (z_1 + \frac{1}{2}) \mathbf{a}_3$	=	$-a(x_1 - \frac{1}{2}) \hat{\mathbf{x}} + b(y_1 + \frac{1}{2}) \hat{\mathbf{y}} + c(z_1 + \frac{1}{2}) \hat{\mathbf{z}}$	(4a)	As I
$\mathbf{B}_5$	$x_2 \mathbf{a}_1 + y_2 \mathbf{a}_2 + z_2 \mathbf{a}_3$	=	$ax_2 \hat{\mathbf{x}} + by_2 \hat{\mathbf{y}} + cz_2 \hat{\mathbf{z}}$	(4a)	Co I
$\mathbf{B}_6$	$-x_2 \mathbf{a}_1 - y_2 \mathbf{a}_2 + (z_2 + \frac{1}{2}) \mathbf{a}_3$	=	$-ax_2 \hat{\mathbf{x}} - by_2 \hat{\mathbf{y}} + c(z_2 + \frac{1}{2}) \hat{\mathbf{z}}$	(4a)	Co I
$\mathbf{B}_7$	$(x_2 + \frac{1}{2}) \mathbf{a}_1 - (y_2 - \frac{1}{2}) \mathbf{a}_2 + z_2 \mathbf{a}_3$	=	$a(x_2 + \frac{1}{2}) \hat{\mathbf{x}} - b(y_2 - \frac{1}{2}) \hat{\mathbf{y}} + cz_2 \hat{\mathbf{z}}$	(4a)	Co I
$\mathbf{B}_8$	$-(x_2 - \frac{1}{2}) \mathbf{a}_1 + (y_2 + \frac{1}{2}) \mathbf{a}_2 + (z_2 + \frac{1}{2}) \mathbf{a}_3$	=	$-a(x_2 - \frac{1}{2}) \hat{\mathbf{x}} + b(y_2 + \frac{1}{2}) \hat{\mathbf{y}} + c(z_2 + \frac{1}{2}) \hat{\mathbf{z}}$	(4a)	Co I

### References

- [1] P. S. Lyman and C. T. Prewitt, *Room- and high-pressure crystal chemistry of CoAs and FeAs*, Acta Crystallogr. Sect. B **40**, 14–20 (1984), doi:10.1107/S0108768184001695.