

NH₄NO₃ I (*G*₀₈) Structure:

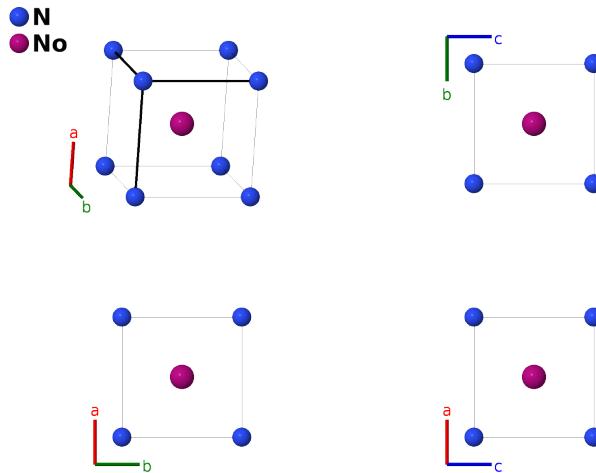
AB_cP2_221_a_b-001

This structure originally had the label AB_cP2_221_a_b.NH4.NO3. Calls to that address will be redirected here.

Cite this page as: D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.

<https://aflow.org/p/5YKM>

https://aflow.org/p/AB_cP2_221_a_b-001



Prototype (NH₄)(NO₃)

AFLOW prototype label AB_cP2_221_a_b-001

Strukturbericht designation *G*₀₈

ICSD none

Pearson symbol cP2

Space group number 221

Space group symbol *Pm* $\bar{3}m$

AFLOW prototype command `aflow --proto=AB_cP2_221_a_b-001
--params=a`

- Ammonium Nitrate exists in a variety of forms, (Hermann, 1937) depending on the temperature:

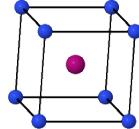
Phase	Temperature °C	Strukturbericht	Page	
I	125 – 170	<i>G</i> ₀ ₈	AB_cP2_221_a_b-001	(this structure)
II	84 – 125	<i>G</i> ₀ ₉	ABC3_tP10_100_b_a_bc	
III	32 – 84	<i>G</i> ₀ ₁₀	ABC3_oP20_62_c_c_cd-002	
IV	-18 – 32	<i>G</i> ₀ ₁₁	A4B2C3_oP18_59_ef_ab_af-001	
V	< -18		A4B2C3_tP72_77_8d_ab2c2d_6d2-001	

- In the *G*₀₈ structure, both the NH₄ and NO₃ radicals rotate freely about their centers of mass (Kracek, 1931). The two radicals sit on the same sites as atoms in the CsCl (*B*2) structure.

Simple Cubic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= a \hat{\mathbf{z}}\end{aligned}$$

a1
a3
a2



Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1 =$	0	0	(1a)	NH I
$\mathbf{B}_2 =$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$\frac{1}{2} a \hat{\mathbf{x}} + \frac{1}{2} a \hat{\mathbf{y}} + \frac{1}{2} a \hat{\mathbf{z}}$	(1b)	NO I

References

- [1] F. C. Kracek, S. B. Hendricks, and E. Posnjak, *Group Rotation in Solid Ammonium and Calcium Nitrates*, Nature **128**, 410–411 (1931), doi:10.1038/128410b0.
- [2] S. B. Hendricks, E. Posnjak, and F. C. Kracek, *Molecular Rotation in the Solid State. The Variation of the Crystal Structure of Ammonium Nitrate with Temperature*, J. Am. Chem. Soc. **54**, 2766–2786 (1932), doi:10.1021/ja01346a020.

Found in

- [1] C. Hermann, O. Lohrmann, and H. Philipp, eds., *Strukturbericht Band II 1928-1932* (Akademische Verlagsgesellschaft M. B. H., Leipzig, 1937).