

Zincblende (ZnAs , $B3$) Structure:

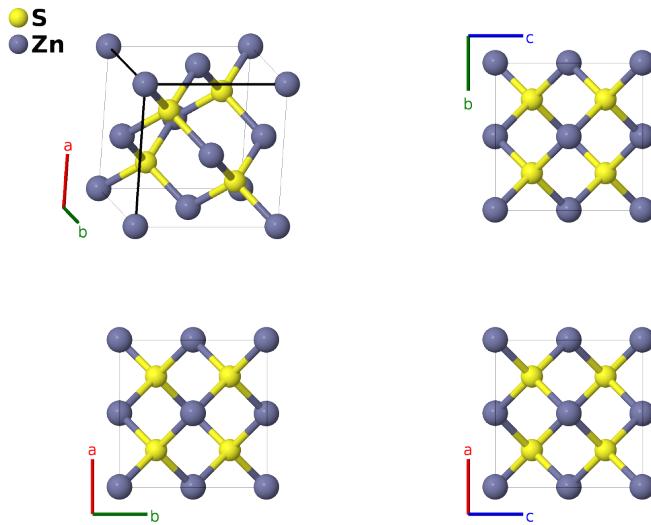
AB_cF8_216_a_c-001

This structure originally had the label AB_cF8_216_c_a. Calls to that address will be redirected here.

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<https://aflow.org/p/TL8Z>

https://aflow.org/p/AB_cF8_216_a_c-001



Prototype

SZn

AFLOW prototype label

AB_cF8_216_a_c-001

Strukturbericht designation

$B3$

Mineral name

zincblende

ICSD

651457

Pearson symbol

cF8

Space group number

216

Space group symbol

$F\bar{4}3m$

AFLOW prototype command

```
aflow --proto=AB_cF8_216_a_c-001  
--params=a
```

Other compounds with this structure

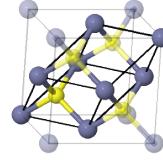
AgI, AlAs, AlP, AlSb, AsB, AsGa, AsIn, BN, BePo, BeS, BeSe, BeTe, β -CsSi, CdPo, CdS, CdSe, CdTe, CuI (marshite), FeN, GaP, GaSb, HgS, HgSe, HgTe, InP, InSb, β -MnS, β -MnSe, OZn, PoZn, SSn, SZn, TeZn, (Fe, Zn)S (sphalerite)

- This is the cubic analog of wurtzite ($B4$), i.e. the stacking of the ZnS dimers along the $<111>$ direction is ABCABC ... This is also a two-component analog of diamond ($A4$), without the inversion symmetry in the middle of the bond.
- Zincblende may also be referred to as sphalerite.

Face-centered Cubic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}}\end{aligned}$$

$\hat{\mathbf{a}}_2$
 $\hat{\mathbf{a}}_1$



Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1 =$	0	0	(4a)	S I
$\mathbf{B}_2 =$	$\frac{1}{4}\mathbf{a}_1 + \frac{1}{4}\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{y}} + \frac{1}{4}a\hat{\mathbf{z}}$	(4c)	Zn I

References

- [1] B. J. Skinner, *Unit-Cell Edges of Natural and Synthetic Sphalerites*, Am. Mineral. **46**, 1399–1411 (1961).

Found in

- [1] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).