

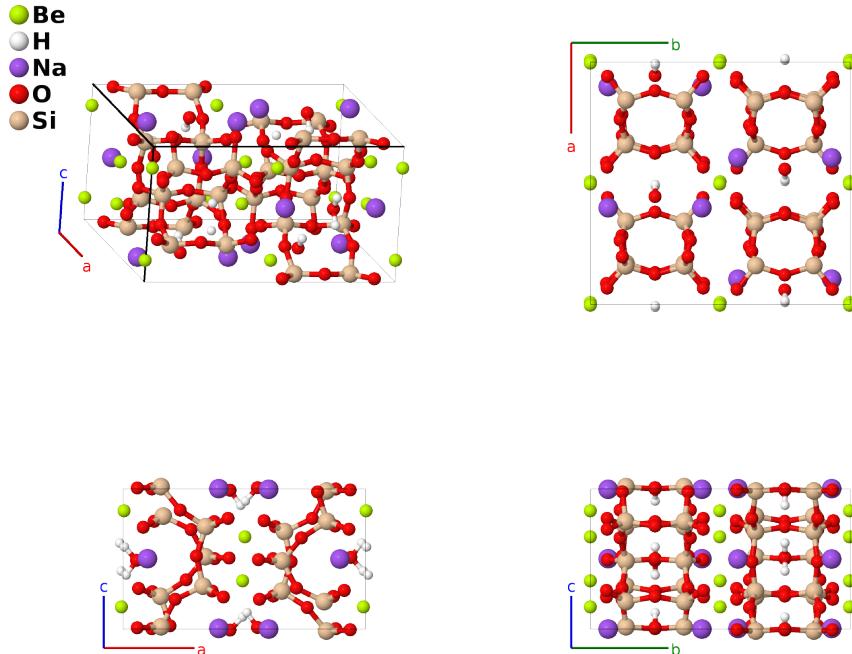
# Epididymite ( $\text{BeHNaO}_8\text{Si}_3$ , $S4_7$ ) Structure: ABCD8E3\_oP112\_62\_d\_2c\_d\_4c6d\_3d-001

This structure originally had the label ABCD8E3\_oP112\_62\_d\_2c\_d\_4c6d\_3d. Calls to that address will be redirected here.

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<https://aflow.org/p/QQ21>

[https://aflow.org/p/ABCD8E3\\_oP112\\_62\\_d\\_2c\\_d\\_4c6d\\_3d-001](https://aflow.org/p/ABCD8E3_oP112_62_d_2c_d_4c6d_3d-001)



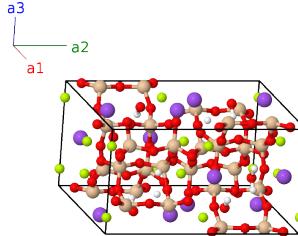
<b>Prototype</b>	$\text{BeHNaO}_8\text{Si}_3$
<b>AFLOW prototype label</b>	ABCD8E3_oP112_62_d_2c_d_4c6d_3d-001
<b>Strukturbericht designation</b>	$S4_7$
<b>Mineral name</b>	epididymite
<b>ICSD</b>	160817
<b>Pearson symbol</b>	$\text{oP}112$
<b>Space group number</b>	62
<b>Space group symbol</b>	$Pnma$
<b>AFLOW prototype command</b>	<pre>aflow --proto=ABCD8E3_oP112_62_d_2c_d_4c6d_3d-001 --params=a, b/a, c/a, x1, z1, x2, z2, x3, z3, x4, z4, x5, z5, x6, z6, x7, z7, y7, z7, x8, y8, z8, x9, y9, z9, x10, y10, z10, x11, y11, z11, x12, y12, z12, x13, y13, z13, x14, y14, z14, x15, y15, z15, x16, y16, z16, x17, y17, z17</pre>

- (Ito, 1934)'s original determination of the epididymite structure, designated as  $S4_7$  by (Gottfried, 1937), was flawed, even in the author's estimation. In particular, both the Si-O and Si-Be distances are very small, with Si-Be less than 1 Å.
- Ito and many later authors assumed that the hydrogen atoms formed OH radicals, but (Gattta, 2008) found that the hydrogen atoms were actually a part of a water molecule.
- Since the (Ito, 1934) structure is not even suitable for starting first-principles calculations and the (Gattta, 2008) structure maintains the  $Pnma$  #62 space group, we have dropped the former and designate the latter as  $S4_7$ .
- Epididymite and its diomorph, eudidymite, are two forms of hydrated sodium beryllium silicate which are stable under ambient conditions. (Gattta, 2008)

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### Simple Orthorhombic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$




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### Basis vectors

	Lattice coordinates	=	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$x_1 \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_1 \mathbf{a}_3$	=	$ax_1 \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} + cz_1 \hat{\mathbf{z}}$	(4c)	H I
$\mathbf{B}_2$	$-(x_1 - \frac{1}{2}) \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + (z_1 + \frac{1}{2}) \mathbf{a}_3$	=	$-a(x_1 - \frac{1}{2}) \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} + c(z_1 + \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	H I
$\mathbf{B}_3$	$-x_1 \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_1 \mathbf{a}_3$	=	$-ax_1 \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} - cz_1 \hat{\mathbf{z}}$	(4c)	H I
$\mathbf{B}_4$	$(x_1 + \frac{1}{2}) \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 - (z_1 - \frac{1}{2}) \mathbf{a}_3$	=	$a(x_1 + \frac{1}{2}) \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} - c(z_1 - \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	H I
$\mathbf{B}_5$	$x_2 \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_2 \mathbf{a}_3$	=	$ax_2 \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} + cz_2 \hat{\mathbf{z}}$	(4c)	H II
$\mathbf{B}_6$	$-(x_2 - \frac{1}{2}) \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + (z_2 + \frac{1}{2}) \mathbf{a}_3$	=	$-a(x_2 - \frac{1}{2}) \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} + c(z_2 + \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	H II
$\mathbf{B}_7$	$-x_2 \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_2 \mathbf{a}_3$	=	$-ax_2 \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} - cz_2 \hat{\mathbf{z}}$	(4c)	H II
$\mathbf{B}_8$	$(x_2 + \frac{1}{2}) \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 - (z_2 - \frac{1}{2}) \mathbf{a}_3$	=	$a(x_2 + \frac{1}{2}) \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} - c(z_2 - \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	H II
$\mathbf{B}_9$	$x_3 \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_3 \mathbf{a}_3$	=	$ax_3 \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(4c)	O I
$\mathbf{B}_{10}$	$-(x_3 - \frac{1}{2}) \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + (z_3 + \frac{1}{2}) \mathbf{a}_3$	=	$-a(x_3 - \frac{1}{2}) \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} + c(z_3 + \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	O I
$\mathbf{B}_{11}$	$-x_3 \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_3 \mathbf{a}_3$	=	$-ax_3 \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(4c)	O I
$\mathbf{B}_{12}$	$(x_3 + \frac{1}{2}) \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 - (z_3 - \frac{1}{2}) \mathbf{a}_3$	=	$a(x_3 + \frac{1}{2}) \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} - c(z_3 - \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	O I
$\mathbf{B}_{13}$	$x_4 \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_4 \mathbf{a}_3$	=	$ax_4 \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} + cz_4 \hat{\mathbf{z}}$	(4c)	O II
$\mathbf{B}_{14}$	$-(x_4 - \frac{1}{2}) \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + (z_4 + \frac{1}{2}) \mathbf{a}_3$	=	$-a(x_4 - \frac{1}{2}) \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} + c(z_4 + \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	O II
$\mathbf{B}_{15}$	$-x_4 \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_4 \mathbf{a}_3$	=	$-ax_4 \hat{\mathbf{x}} + \frac{3}{4}b \hat{\mathbf{y}} - cz_4 \hat{\mathbf{z}}$	(4c)	O II
$\mathbf{B}_{16}$	$(x_4 + \frac{1}{2}) \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 - (z_4 - \frac{1}{2}) \mathbf{a}_3$	=	$a(x_4 + \frac{1}{2}) \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} - c(z_4 - \frac{1}{2}) \hat{\mathbf{z}}$	(4c)	O II
$\mathbf{B}_{17}$	$x_5 \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_5 \mathbf{a}_3$	=	$ax_5 \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(4c)	O III







$$\begin{aligned}
 \mathbf{B}_{110} &= \left( x_{17} + \frac{1}{2} \right) \mathbf{a}_1 + y_{17} \mathbf{a}_2 - \left( z_{17} - \frac{1}{2} \right) \mathbf{a}_3 & = & a \left( x_{17} + \frac{1}{2} \right) \hat{\mathbf{x}} + b y_{17} \hat{\mathbf{y}} - c \left( z_{17} - \frac{1}{2} \right) \hat{\mathbf{z}} & (8d) & \text{Si III} \\
 \mathbf{B}_{111} &= x_{17} \mathbf{a}_1 - \left( y_{17} - \frac{1}{2} \right) \mathbf{a}_2 + z_{17} \mathbf{a}_3 & = & a x_{17} \hat{\mathbf{x}} - b \left( y_{17} - \frac{1}{2} \right) \hat{\mathbf{y}} + c z_{17} \hat{\mathbf{z}} & (8d) & \text{Si III} \\
 \mathbf{B}_{112} &= - \left( x_{17} - \frac{1}{2} \right) \mathbf{a}_1 + \left( y_{17} + \frac{1}{2} \right) \mathbf{a}_2 + \left( z_{17} + \frac{1}{2} \right) \mathbf{a}_3 & = & -a \left( x_{17} - \frac{1}{2} \right) \hat{\mathbf{x}} + b \left( y_{17} + \frac{1}{2} \right) \hat{\mathbf{y}} + c \left( z_{17} + \frac{1}{2} \right) \hat{\mathbf{z}} & (8d) & \text{Si III}
 \end{aligned}$$

## References

- [1] G. D. Gatta, N. Rotiroti, G. J. McIntyre, A. Guastoni, and F. Nestola, *New insights into the crystal chemistry of epididymite and eudidymite from Malosa, Malawi: A single-crystal neutron diffraction study*, Am. Mineral. **93**, 1158–1165 (2008), doi:10.2138/am.2008.2965.
- [2] T. Ito, *The Structure of Epididymite (HNaBeSi<sub>3</sub>O<sub>8</sub>)*, Z. Kristallogr. **88**, 142–149 (1934), doi:10.1524/zkri.1934.88.1.142.
- [3] C. Gottfried and F. Schossberger, eds., *Strukturbericht Band III 1933-1935* (Akademische Verlagsgesellschaft M. B. H., Leipzig, 1937).

## Found in

- [1] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).