

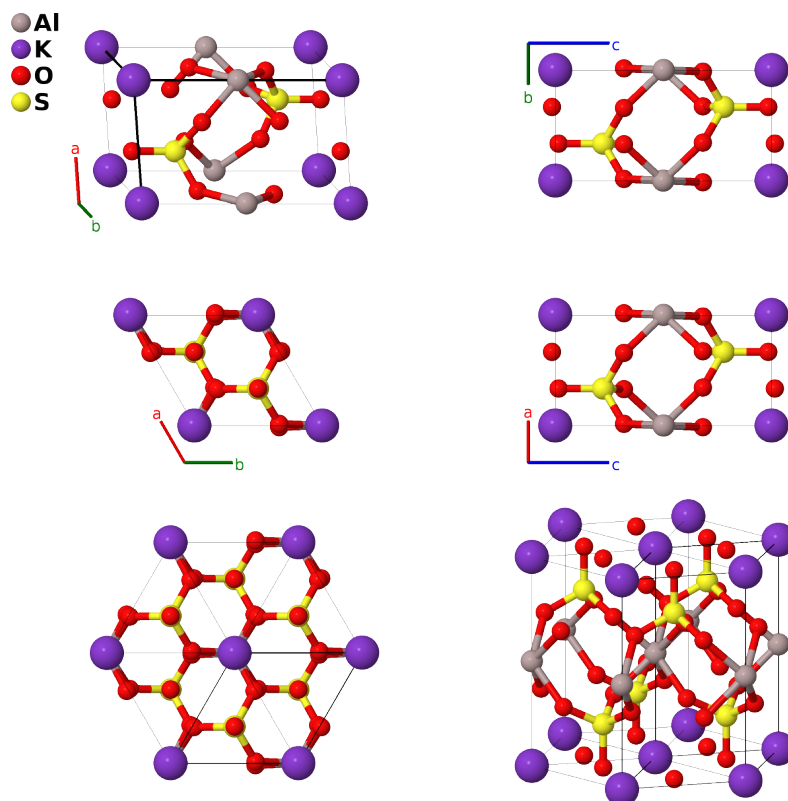
Steklite [KAl(SO₄)₂, H₃₂] Structure: ABC8D2_hP12_150_a_b_dg_d-001

This structure originally had the label ABC8D2_hP12_150_b_a_dg_d. Calls to that address will be redirected here.

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<https://afLOW.org/p/SDL2>

https://afLOW.org/p/ABC8D2_hP12_150_a_b_dg_d-001



Prototype	AlK ₂ O ₈ S ₂
AFLOW prototype label	ABC8D2_hP12_150_a_b_dg_d-001
<i>Strukturbericht</i> designation	H ₃ ₂
Mineral name	steklite
ICSD	60170
Pearson symbol	hP12
Space group number	150
Space group symbol	P321
AFLOW prototype command	afLOW --proto=ABC8D2_hP12_150_a_b_dg_d-001 --params=a, c/a, z ₃ , z ₄ , x ₅ , y ₅ , z ₅

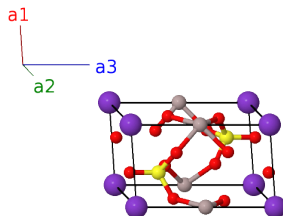
Other compounds with this structure

$\text{NH}_4(\text{Al}, \text{Fe})(\text{SO}_4)_2$ (godovikovite), $\text{KCr}(\text{SO}_4)_2$, $\text{RbCr}(\text{SO}_4)_2$

- This has been a rather difficult structure to follow through the literature. (Villars, 2016) quotes the structure given by (Manoli, 1970), but gives the space group as $P\bar{3}$ #147. (Murashko, 2013) lists the space group as both $P312$ #149 and $P321$ #150, as well as listing obviously incorrect Wyckoff positions.
- (West, 2008) states that the simple structure is in $P\bar{3}$ but that it may be doubled along the c axis and be in space group $P321$.
- After correcting Murashko's results, we find that all of these interpretations yield essentially the same structure in a given layer, and only differ as the structure is reflected through the $z = 0$ plane. As it is not clear which structure is correct, we will use the original $H3_2$ structure given by (Hermann, 1937).
- Steklite is the name of the mineral form of this compound (Murashko, 2013). (Hermann, 1937) simply calls it *Wasserfreier Alaun* (anhydrous alum). For hydrated alum, $\text{KAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$, see the $H4_{13}$ structure.

Trigonal (Hexagonal) primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	0	$=$	0	(1a)	Al I
\mathbf{B}_2	$\frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \hat{\mathbf{z}}$	(1b)	K I
\mathbf{B}_3	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(2d)	O I
\mathbf{B}_4	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(2d)	O I
\mathbf{B}_5	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_4 \hat{\mathbf{z}}$	(2d)	S I
\mathbf{B}_6	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 - z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - cz_4 \hat{\mathbf{z}}$	(2d)	S I
\mathbf{B}_7	$x_5 \mathbf{a}_1 + y_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$\frac{1}{2}a(x_5 + y_5) \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a(x_5 - y_5) \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(6g)	O II
\mathbf{B}_8	$-y_5 \mathbf{a}_1 + (x_5 - y_5) \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$\frac{1}{2}a(x_5 - 2y_5) \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_5 \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(6g)	O II
\mathbf{B}_9	$-(x_5 - y_5) \mathbf{a}_1 - x_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$-\frac{1}{2}a(2x_5 - y_5) \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ay_5 \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(6g)	O II
\mathbf{B}_{10}	$y_5 \mathbf{a}_1 + x_5 \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$\frac{1}{2}a(x_5 + y_5) \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a(x_5 - y_5) \hat{\mathbf{y}} - cz_5 \hat{\mathbf{z}}$	(6g)	O II
\mathbf{B}_{11}	$(x_5 - y_5) \mathbf{a}_1 - y_5 \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$\frac{1}{2}a(x_5 - 2y_5) \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_5 \hat{\mathbf{y}} - cz_5 \hat{\mathbf{z}}$	(6g)	O II
\mathbf{B}_{12}	$-x_5 \mathbf{a}_1 - (x_5 - y_5) \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$-\frac{1}{2}a(2x_5 - y_5) \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ay_5 \hat{\mathbf{y}} - cz_5 \hat{\mathbf{z}}$	(6g)	O II

References

- [1] L. Vegard and A. Maurstad, *Die Kristallstruktur der wasserfreien Alaune $R'R''(\text{SO}_4)_2$* , Z. Kristallogr. **69**, 519–532 (1929), doi:10.1524/zkri.1929.69.1.519.
- [2] L. Vegard and A. Maurstad, *Die Kristallstruktur der wasserfreien Alaune $R'R''(\text{SO}_4)_2$* , Skrifter utgitt av det Norske Videnskaps-Akademi i Oslo pp. 1–24 (1928).

- [3] D. V. West, Q. Huang, H. W. Zandbergen, T. M. McQueen, and R. J. Cava, *Structural disorder, octahedral coordination and two-dimensional ferromagnetism in anhydrous alums*, J. Solid State Chem. **181**, 2768–2775 (2008), doi:10.1016/j.jssc.2008.07.006.
- [4] M. N. Murashko, I. V. Pekov, S. V. Krivovichev, A. P. Chernyatyeva, V. O. Yapaskurt, A. E. Zadov, and M. E. Zelensky, *Steklite, $KAl(SO_4)_2$: A finding at the Tolbachik Volcano, Kamchatka, Russia, validating its status as a mineral species and crystal structure*, Geol. Ore Deposits **55**, 594–600 (2013), doi:10.1134/S1075701513070088.
- [5] P. Villars, *$KAl(SO_4)_2$ ($KAl[SO_4]_2$ trig1) Crystal Structure* (2016). PAULING FILE in: Inorganic Solid Phases, SpringerMaterials (online database).
- [6] J. M. Manoli, P. Herpin, and G. Pannetier, *Structure cristalline du sulfate double d'aluminium et de potassium*, Bull. Soc. Chim. France pp. 98–101 (1970).

Found in

- [1] P. P. Ewald and C. Hermann, eds., *Strukturbericht 1913-1928* (Akademische Verlagsgesellschaft M. B. H., Leipzig, 1931).
- [2] C. Hermann, O. Lohrmann, and H. Philipp, eds., *Strukturbericht Band II 1928-1932* (Akademische Verlagsgesellschaft M. B. H., Leipzig, 1937).
- [3] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).