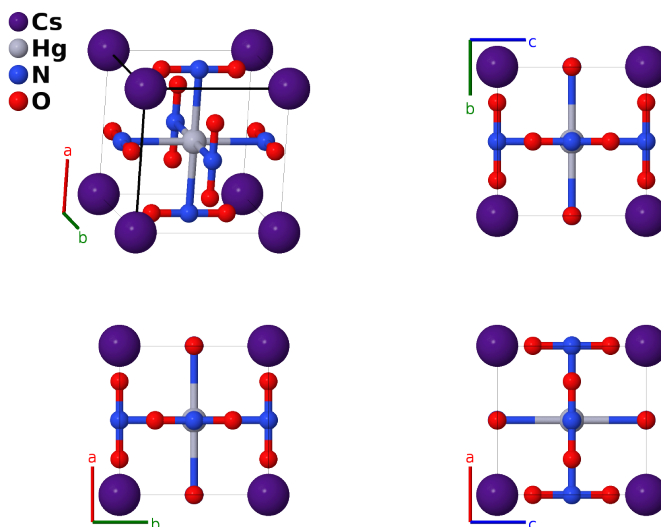


CsHgN₃O₆ Structure: ABC3D6_cP11_200_a_b_c_f-001

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<https://aflow.org/p/4JP4>

https://aflow.org/p/ABC3D6_cP11_200_a_b_c_f-001



Prototype	CsHgN ₃ O ₆
AFLOW prototype label	ABC3D6_cP11_200_a_b_c_f-001
ICSD	28645
Pearson symbol	cP11
Space group number	200
Space group symbol	$Pm\bar{3}$
AFLOW prototype command	<code>aflow --proto=ABC3D6_cP11_200_a_b_c_f-001 --params=a, x₄</code>

Other compounds with this structure

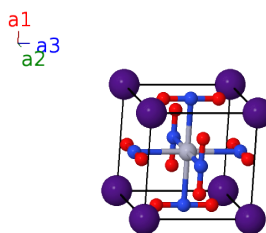
CsCdN₃O₆, KHgN₃O₆, RbCdN₃O₆, RbHgN₃O₆, TlCdN₃O₆, TlHgN₃O₆

Simple Cubic primitive vectors

$$\mathbf{a}_1 = a \hat{x}$$

$$\mathbf{a}_2 = a \hat{y}$$

$$\mathbf{a}_3 = a \hat{z}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	=	0	=	0	(1a) Cs I
\mathbf{B}_2	=	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}}$	(1b) Hg I
\mathbf{B}_3	=	$\frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}}$	(3c) N I
\mathbf{B}_4	=	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{z}}$	(3c) N I
\mathbf{B}_5	=	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}}$	(3c) N I
\mathbf{B}_6	=	$x_4\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_3$	=	$ax_4\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{z}}$	(6f) O I
\mathbf{B}_7	=	$-x_4\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_3$	=	$-ax_4\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{z}}$	(6f) O I
\mathbf{B}_8	=	$\frac{1}{2}\mathbf{a}_1 + x_4\mathbf{a}_2$	=	$\frac{1}{2}a\hat{\mathbf{x}} + ax_4\hat{\mathbf{y}}$	(6f) O I
\mathbf{B}_9	=	$\frac{1}{2}\mathbf{a}_1 - x_4\mathbf{a}_2$	=	$\frac{1}{2}a\hat{\mathbf{x}} - ax_4\hat{\mathbf{y}}$	(6f) O I
\mathbf{B}_{10}	=	$\frac{1}{2}\mathbf{a}_2 + x_4\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{y}} + ax_4\hat{\mathbf{z}}$	(6f) O I
\mathbf{B}_{11}	=	$\frac{1}{2}\mathbf{a}_2 - x_4\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{y}} - ax_4\hat{\mathbf{z}}$	(6f) O I

References

- [1] A. Ferrari and C. Colla, *Mercurinitriti di metalli monovalenti*, Gazz. Chim. Ital. **65**, 789–797 (1935).

Found in

- [1] *Crystallography Open Database*. Information card for entry 1010350.