

LiNiO₂ Structure:

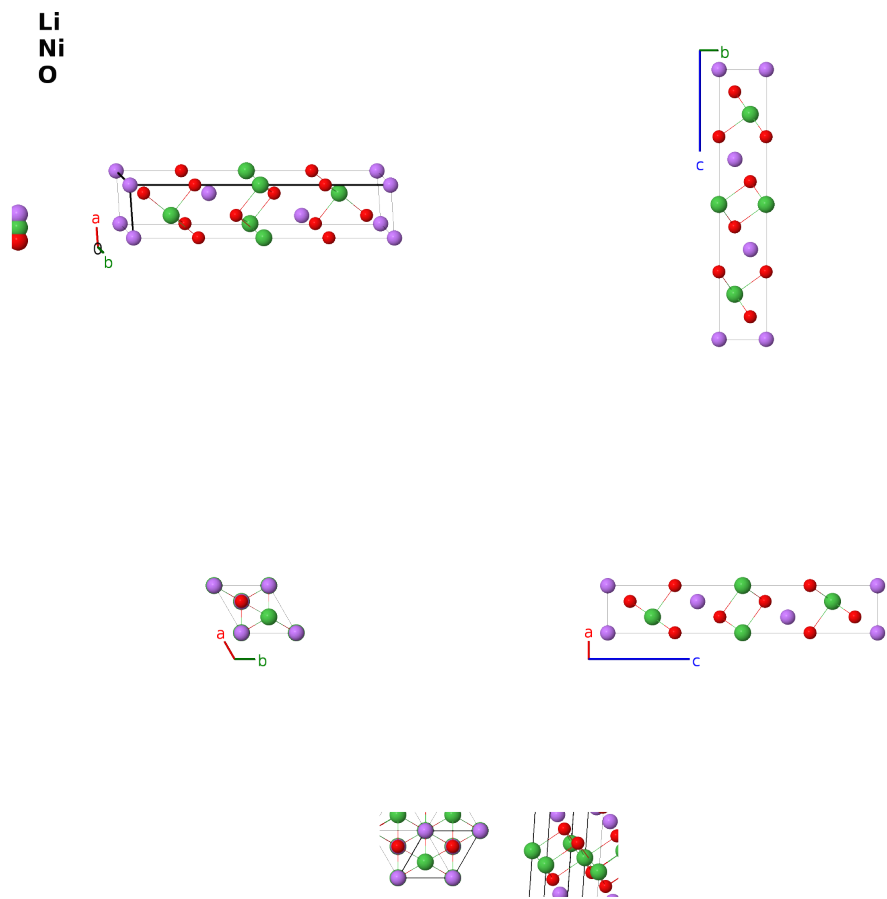
ABC2_hR4_166_a_b_c-010

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<https://aflow.org/prototype-encyclopedia/0R3A>

https://aflow.org/prototype-encyclopedia/ABC2_hR4_166_a_b_c-010



Prototype	LiNiO ₂
AFLOW prototype label	ABC2_hR4_166_a_b_c-010
ICSD	26608
CCDC	1601998
Pearson symbol	hR4
Space group number	166
Space group symbol	$R\bar{3}m$
AFLOW prototype command	<code>aflow --proto=ABC2_hR4_166_a_b_c-010</code> <code>--params=a, c/a, x₃</code>

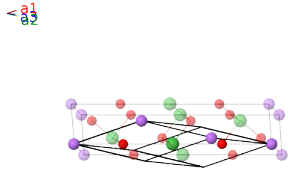
Other compounds with this structure

NaNiO₂

- α -NaFeO₂, rhombohedral delafossite, CuFeO₂, caswellsilverite (α -NaFeO₂, $F\bar{5}_1$), LiNiO₂ and Li₂Mn₃O₇ have the same prototype label, ABC2_hR4.166_a_b_c. The structures are generated by the same symmetry operations with different sets of parameters (`--params`) specified in their corresponding CIF files.
 - Hexagonal settings of this structure can be obtained with the option `--hex`.
-

Rhombohedral primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{1}{3}c \hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{\sqrt{3}}a \hat{\mathbf{y}} + \frac{1}{3}c \hat{\mathbf{z}} \\ \mathbf{a}_3 &= -\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{1}{3}c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	=	0	=	0	(1a) Li I
$\mathbf{B}_2$	=	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	=	$\frac{1}{2}c \hat{\mathbf{z}}$	(1b) Ni I
$\mathbf{B}_3$	=	$x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	=	$cx_3 \hat{\mathbf{z}}$	(2c) O I
$\mathbf{B}_4$	=	$-x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	=	$-cx_3 \hat{\mathbf{z}}$	(2c) O I

References

- [1] L. D. Dyer, B. S. B. Jr., and G. P. Smith, *Alkali Metal-Nickel Oxides of the Type MNiO₂*, J. Am. Chem. Soc. **76**, 1499–1503 (1954), doi:10.1021/ja01635a012.

Found in

- [1] M. Sofin and M. Jansen, *New Route of Preparation and Properties of NaNiO₂*, Z. Naturforsch. B **60**, 701–704 (2005), doi:10.1515/znb-2005-0615.

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