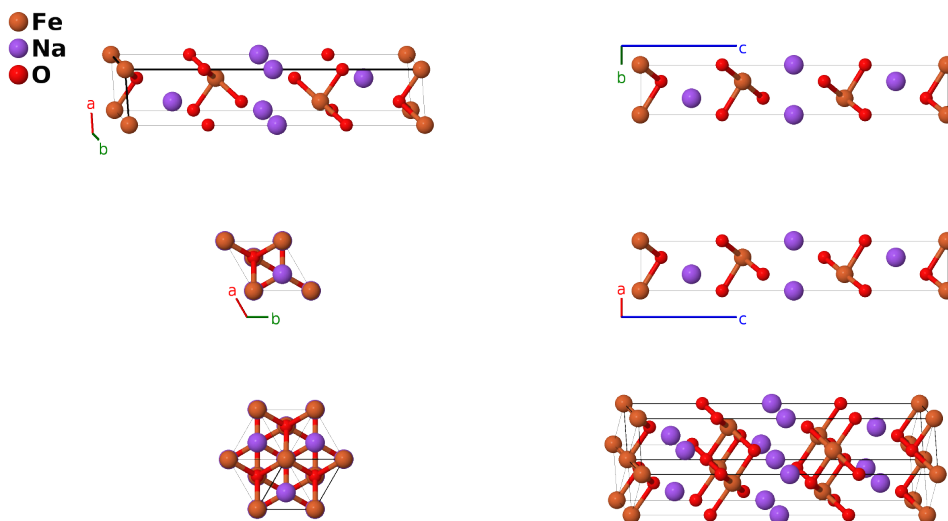


# $\alpha$ -NaFeO<sub>2</sub> Structure: ABC2\_hR4\_166\_a\_b\_c-009

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<https://afLOW.org/p/5FG5>

[https://afLOW.org/p/ABC2\\_hR4\\_166\\_a\\_b\\_c-009](https://afLOW.org/p/ABC2_hR4_166_a_b_c-009)



<b>Prototype</b>	FeNaO <sub>2</sub>
<b>AFLOW prototype label</b>	ABC2_hR4_166_a_b_c-009
<b>ICSD</b>	187705
<b>Pearson symbol</b>	hR4
<b>Space group number</b>	166
<b>Space group symbol</b>	$R\bar{3}m$
<b>AFLOW prototype command</b>	<code>afLOW --proto=ABC2_hR4_166_a_b_c-009 --params=a, c/a, x<sub>3</sub></code>

## Other compounds with this structure

CeRbO<sub>2</sub>, CeRbS<sub>2</sub>, CeRbSSe, CeRbSe<sub>2</sub>, CeRbTe<sub>2</sub>, CeRbTeSe<sub>2</sub>, CsYbSe<sub>2</sub>, KCeS<sub>2</sub>, LiCoO<sub>2</sub>, LiNi<sub>2</sub>, NaCoO<sub>2</sub>, NaYbO<sub>2</sub>, NaYbS<sub>2</sub>, NaYbSe<sub>2</sub>

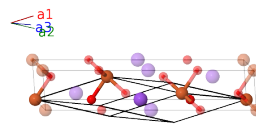
- This is also referred to as O3-NaFeO<sub>2</sub>.
- This structure is related to O2-LiCoO<sub>2</sub>, O4-LiCoO<sub>2</sub>, and OP4-LiNaCo<sub>2</sub>O<sub>4</sub>.
- $\alpha$ -NaFeO<sub>2</sub> has the same AFLOW Label, ABC2\_hR4\_166\_a\_b\_c, as rhombohedral delafossite, CuFeO<sub>2</sub> and caswellsilverite ( $F5_1$ ). The difference in the internal parameter  $z_3$  causes a large change in the bonding of the crystals, so we present them as different structures.
- The structures are generated by the same symmetry operations with different sets of parameters (`--params`) specified in their corresponding CIF files.

- Hexagonal settings of this structure can be obtained with the option `--hex`.

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### Rhombohedral primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{1}{3}c \hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{\sqrt{3}}a \hat{\mathbf{y}} + \frac{1}{3}c \hat{\mathbf{z}} \\ \mathbf{a}_3 &= -\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{1}{3}c \hat{\mathbf{z}}\end{aligned}$$




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### Basis vectors

	Lattice coordinates	=	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$0$	=	$0$	(1a)	Fe I
$\mathbf{B}_2$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	=	$\frac{1}{2}c \hat{\mathbf{z}}$	(1b)	Na I
$\mathbf{B}_3$	$x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	=	$cx_3 \hat{\mathbf{z}}$	(2c)	O I
$\mathbf{B}_4$	$-x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	=	$-cx_3 \hat{\mathbf{z}}$	(2c)	O I

### References

- [1] N. Yabuuchi, H. Yoshida, and S. Komaba, *Crystal Structures and Electrode Performance of Alpha-NaFeO<sub>2</sub> for Rechargeable Sodium Batteries*, *Electrochemistry* **80**, 716–719 (2012), doi:10.5796/electrochemistry.80.716.