

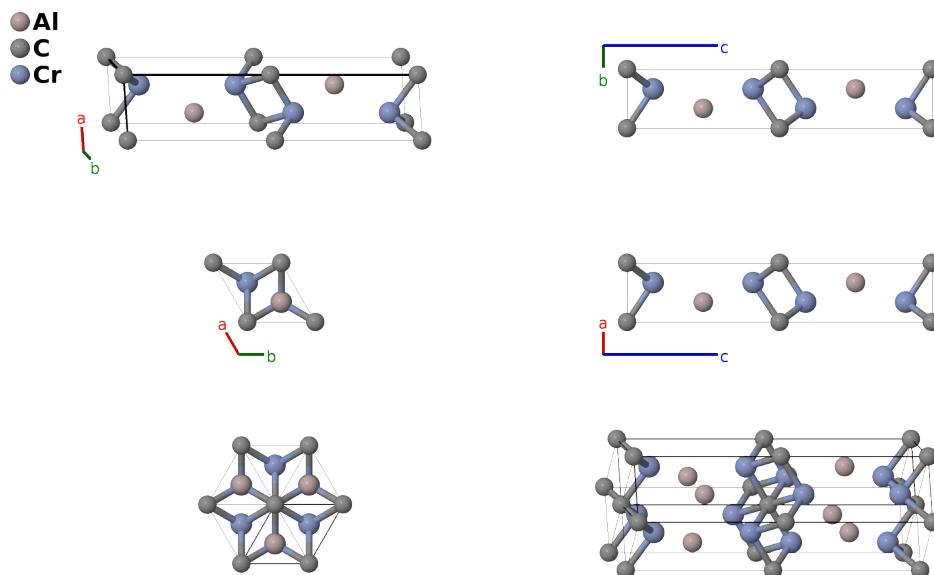
AlCr₂ Structure: ABC2_hP8_194_c_a_f-007

This structure originally had the label `ABC2_hP8_194_d_a_f`. Calls to that address will be redirected here.

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<https://aflow.org/p/NEW2>

https://aflow.org/p/ABC2_hP8_194_c_a_f-007



Prototype	AlCr ₂
AFLOW prototype label	ABC2_hP8_194_c_a_f-007
ICSD	42918
Pearson symbol	hP8
Space group number	194
Space group symbol	<i>P6₃/mmc</i>
AFLOW prototype command	<code>aflow --proto=ABC2_hP8_194_c_a_f-007 --params=a, c/a, z₃</code>

Other compounds with this structure

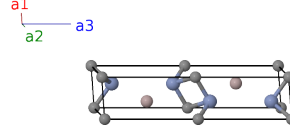
AlCNb₂, AlCTa₂, AlCTi₂, AlCV₂, AlCuO₂, AlNTi₂, AsCV₂, CCdT₂, CGaCr₂, CGaMo₂, CGaNb₂, CGaTi₂, CGaY₂, CGeCr₂, CGeTi₂, CGeV₂, CInHf₂, CInNb₂, CInTi₂, CInZr₂, CPNb₂, CPV₂, CPbHf₂, CPbTi₂, CPbZr₂, CSTi₂, CSZr₂, CSnHf₂, CSnNb₂, CSnTi₂, CSnZr₂, CTiHf₂, CTiTi₂, GaNCr₂

- Note that all of the atoms sit on close-packed $\langle 0001 \rangle$ planes. The stacking sequence may be written

Atom	Cr	C	Cr	Al	Cr	C	Cr	Al
Position	B	A	C	B	C	A	B	C

Hexagonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	0	$=$	0	(2a)	C I
\mathbf{B}_2	$\frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \hat{\mathbf{z}}$	(2a)	C I
\mathbf{B}_3	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(2c)	Al I
\mathbf{B}_4	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(2c)	Al I
\mathbf{B}_5	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(4f)	Cr I
\mathbf{B}_6	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_3 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_3 + \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	Cr I
\mathbf{B}_7	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(4f)	Cr I
\mathbf{B}_8	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 - (z_3 - \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} - c(z_3 - \frac{1}{2}) \hat{\mathbf{z}}$	(4f)	Cr I

References

- [1] W. Jeitschko, H. Nowotny, and F. Benesovsky, *Kohlenstoffhaltige ternäre Verbindungen (H-Phase)*, *Monatsh. Chem. Verw.* Tl. **94**, 672–676 (1963), doi:10.1007/BF00913068.

Found in

- [1] P. Villars and L. Calvert, *Pearson's Handbook of Crystallographic Data for Intermetallic Phases* (ASM International, Materials Park, OH, 1991), 2nd edn.