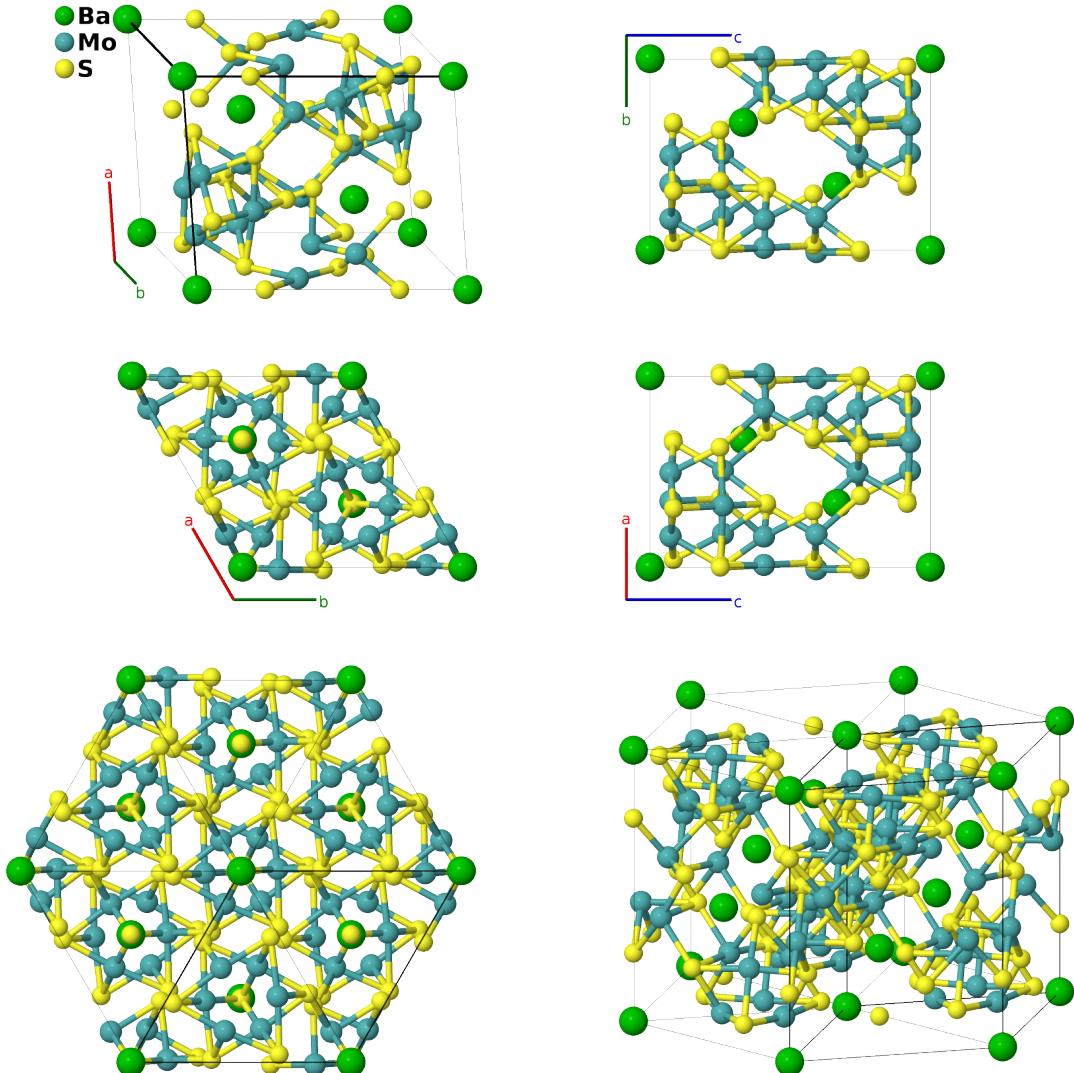


BaMo₆S₈ Structure: AB6C8_hR15_148_a_f_cf-001

Cite this page as: H. Eckert, S. Divilov, A. Zettel, M. J. Mehl, D. Hicks, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*. In preparation.

[https://afflow.org/p/CY1Y](https://aflow.org/p/CY1Y)

https://afflow.org/p/AB6C8_hR15_148_a_f_cf-001



Prototype	BaMo ₆ S ₈
AFLOW prototype label	AB6C8_hR15_148_a_f_cf-001
ICSD	none
Pearson symbol	hR15
Space group number	148
Space group symbol	$R\bar{3}$

AFLOW prototype command `aflow --proto=AB6C8_hR15_148_a_f_cf-001
--params=a, c/a, x2, x3, y3, z3, x4, y4, z4`

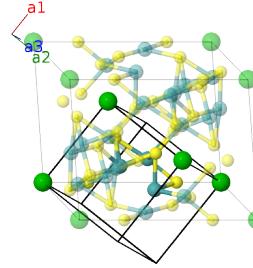
Other compounds with this structure

CaMo₆S₈, CeMo₆S₈, CeMo₆Se₈, EuMo₆S₈, LaMo₆S₈, LaMo₆Se₈, NdMo₆S₈, NdMo₆Se₈, PbMo₆S₈, PrMo₆S₈, PrMo₆Se₈, SmMo₆S₈, SmMo₆Se₈, SrMo₆S₈

- BaMo₆S₈ and other Chevrel phase compounds undergo structural transitions at low temperatures. BaMo₆S₈ transforms into a triclinic $P\bar{1}$ structure below 175K. Structural parameters for the low temperature phase can be found in (Kubel, 1990). We use the data for the rhombohedral $R\bar{3}$ structure taken at 177K.
- Hexagonal settings of this structure can be obtained with the option `--hex`.

Rhombohedral primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + \frac{1}{3}c\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{\sqrt{3}}a\hat{\mathbf{y}} + \frac{1}{3}c\hat{\mathbf{z}} \\ \mathbf{a}_3 &= -\frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + \frac{1}{3}c\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	= 0	=	0	(1a)	Ba I
\mathbf{B}_2	= $x_2 \mathbf{a}_1 + x_2 \mathbf{a}_2 + x_2 \mathbf{a}_3$	=	$cx_2 \hat{\mathbf{z}}$	(2c)	S I
\mathbf{B}_3	= $-x_2 \mathbf{a}_1 - x_2 \mathbf{a}_2 - x_2 \mathbf{a}_3$	=	$-cx_2 \hat{\mathbf{z}}$	(2c)	S I
\mathbf{B}_4	= $x_3 \mathbf{a}_1 + y_3 \mathbf{a}_2 + z_3 \mathbf{a}_3$	= $\frac{1}{2}a(x_3 - z_3)\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a(x_3 - 2y_3 + z_3)\hat{\mathbf{y}} + \frac{1}{3}c(x_3 + y_3 + z_3)\hat{\mathbf{z}}$	(6f)	Mo I	
\mathbf{B}_5	= $z_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + y_3 \mathbf{a}_3$	= $-\frac{1}{2}a(y_3 - z_3)\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a(2x_3 - y_3 - z_3)\hat{\mathbf{y}} + \frac{1}{3}c(x_3 + y_3 + z_3)\hat{\mathbf{z}}$	(6f)	Mo I	
\mathbf{B}_6	= $y_3 \mathbf{a}_1 + z_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	= $-\frac{1}{2}a(x_3 - y_3)\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a(x_3 + y_3 - 2z_3)\hat{\mathbf{y}} + \frac{1}{3}c(x_3 + y_3 + z_3)\hat{\mathbf{z}}$	(6f)	Mo I	
\mathbf{B}_7	= $-x_3 \mathbf{a}_1 - y_3 \mathbf{a}_2 - z_3 \mathbf{a}_3$	= $-\frac{1}{2}a(x_3 - z_3)\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a(x_3 - 2y_3 + z_3)\hat{\mathbf{y}} - \frac{1}{3}c(x_3 + y_3 + z_3)\hat{\mathbf{z}}$	(6f)	Mo I	
\mathbf{B}_8	= $-z_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 - y_3 \mathbf{a}_3$	= $\frac{1}{2}a(y_3 - z_3)\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a(2x_3 - y_3 - z_3)\hat{\mathbf{y}} - \frac{1}{3}c(x_3 + y_3 + z_3)\hat{\mathbf{z}}$	(6f)	Mo I	
\mathbf{B}_9	= $-y_3 \mathbf{a}_1 - z_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	= $\frac{1}{2}a(x_3 - y_3)\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a(x_3 + y_3 - 2z_3)\hat{\mathbf{y}} - \frac{1}{3}c(x_3 + y_3 + z_3)\hat{\mathbf{z}}$	(6f)	Mo I	
\mathbf{B}_{10}	= $x_4 \mathbf{a}_1 + y_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	= $\frac{1}{2}a(x_4 - z_4)\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a(x_4 - 2y_4 + z_4)\hat{\mathbf{y}} + \frac{1}{3}c(x_4 + y_4 + z_4)\hat{\mathbf{z}}$	(6f)	S II	
\mathbf{B}_{11}	= $z_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + y_4 \mathbf{a}_3$	= $-\frac{1}{2}a(y_4 - z_4)\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a(2x_4 - y_4 - z_4)\hat{\mathbf{y}} + \frac{1}{3}c(x_4 + y_4 + z_4)\hat{\mathbf{z}}$	(6f)	S II	
\mathbf{B}_{12}	= $y_4 \mathbf{a}_1 + z_4 \mathbf{a}_2 + x_4 \mathbf{a}_3$	= $-\frac{1}{2}a(x_4 - y_4)\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a(x_4 + y_4 - 2z_4)\hat{\mathbf{y}} + \frac{1}{3}c(x_4 + y_4 + z_4)\hat{\mathbf{z}}$	(6f)	S II	
\mathbf{B}_{13}	= $-x_4 \mathbf{a}_1 - y_4 \mathbf{a}_2 - z_4 \mathbf{a}_3$	= $-\frac{1}{2}a(x_4 - z_4)\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a(x_4 - 2y_4 + z_4)\hat{\mathbf{y}} - \frac{1}{3}c(x_4 + y_4 + z_4)\hat{\mathbf{z}}$	(6f)	S II	

$$\begin{aligned} \mathbf{B}_{14} &= -z_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 - y_4 \mathbf{a}_3 & = & \frac{1}{2}a(y_4 - z_4) \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a(2x_4 - y_4 - z_4) \hat{\mathbf{y}} - \frac{1}{3}c(x_4 + y_4 + z_4) \hat{\mathbf{z}} & (6f) & S\text{ II} \\ \mathbf{B}_{15} &= -y_4 \mathbf{a}_1 - z_4 \mathbf{a}_2 - x_4 \mathbf{a}_3 & = & \frac{1}{2}a(x_4 - y_4) \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a(x_4 + y_4 - 2z_4) \hat{\mathbf{y}} - \frac{1}{3}c(x_4 + y_4 + z_4) \hat{\mathbf{z}} & (6f) & S\text{ II} \end{aligned}$$

References

- [1] F. Kubel and K. Yvon, *Structural phase transitions in Chevrel phases containing divalent metal cations. II. Structure refinement of triclinic EuMo₆S₈ and BaMo₆S₈ at low temperature*, Acta Crystallogr. Sect. C **46**, 181–186 (1990), doi:10.1107/S0108270189005913.

Found in

- [1] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).