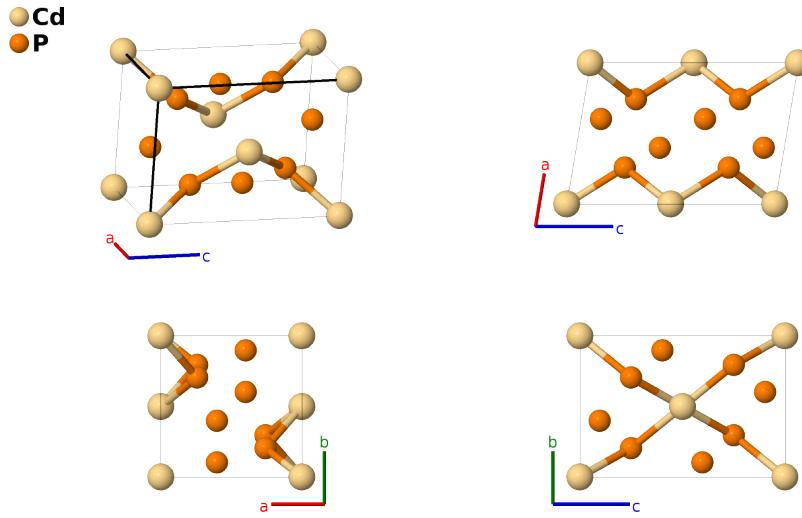


# CdP<sub>4</sub> Structure: AB4\_mP10\_14\_a\_2e-001

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<https://aflow.org/p/L3ZV>

[https://aflow.org/p/AB4\\_mP10\\_14\\_a\\_2e-001](https://aflow.org/p/AB4_mP10_14_a_2e-001)



<b>Prototype</b>	CdP <sub>4</sub>
<b>AFLOW prototype label</b>	AB4_mP10_14_a_2e-001
<b>ICSD</b>	25605
<b>Pearson symbol</b>	mP10
<b>Space group number</b>	14
<b>Space group symbol</b>	$P2_1/c$
<b>AFLOW prototype command</b>	<code>aflow --proto=AB4_mP10_14_a_2e-001 --params=a, b/a, c/a, β, x2, y2, z2, x3, y3, z3</code>

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## Other compounds with this structure

MgP<sub>4</sub>, OsP<sub>4</sub>, RuP<sub>4</sub>

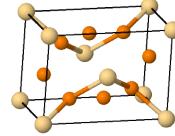
- 
- (Krebs, 1956) give distances in kX. We follow (Wood, 1947) and convert this to Ångstroms by multiplying their distances by 1.00202.
  - This is the low temperature form OsP<sub>4</sub> and RuP<sub>4</sub> (Flörke, 1982). For the high temperature form see the CrP<sub>4</sub> structure.

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## Simple Monoclinic primitive vectors

$$\begin{aligned}
 \mathbf{a}_1 &= a \hat{\mathbf{x}} \\
 \mathbf{a}_2 &= b \hat{\mathbf{y}} \\
 \mathbf{a}_3 &= c \cos \beta \hat{\mathbf{x}} + c \sin \beta \hat{\mathbf{z}}
 \end{aligned}$$

$\mathbf{a}_1$   $\mathbf{a}_2$



## Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	0	0	(2a)	Cd I
$\mathbf{B}_2$	$\frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$\frac{1}{2} c \cos \beta \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}} + \frac{1}{2} c \sin \beta \hat{\mathbf{z}}$	(2a)	Cd I
$\mathbf{B}_3$	$x_2 \mathbf{a}_1 + y_2 \mathbf{a}_2 + z_2 \mathbf{a}_3$	$(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} + by_2 \hat{\mathbf{y}} + cz_2 \sin \beta \hat{\mathbf{z}}$	(4e)	P I
$\mathbf{B}_4$	$-x_2 \mathbf{a}_1 + (y_2 + \frac{1}{2}) \mathbf{a}_2 - (z_2 - \frac{1}{2}) \mathbf{a}_3$	$-(ax_2 + c(z_2 - \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} + b(y_2 + \frac{1}{2}) \hat{\mathbf{y}} - c(z_2 - \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	P I
$\mathbf{B}_5$	$-x_2 \mathbf{a}_1 - y_2 \mathbf{a}_2 - z_2 \mathbf{a}_3$	$-(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} - by_2 \hat{\mathbf{y}} - cz_2 \sin \beta \hat{\mathbf{z}}$	(4e)	P I
$\mathbf{B}_6$	$x_2 \mathbf{a}_1 - (y_2 - \frac{1}{2}) \mathbf{a}_2 + (z_2 + \frac{1}{2}) \mathbf{a}_3$	$(ax_2 + c(z_2 + \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} - b(y_2 - \frac{1}{2}) \hat{\mathbf{y}} + c(z_2 + \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	P I
$\mathbf{B}_7$	$x_3 \mathbf{a}_1 + y_3 \mathbf{a}_2 + z_3 \mathbf{a}_3$	$(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} + by_3 \hat{\mathbf{y}} + cz_3 \sin \beta \hat{\mathbf{z}}$	(4e)	P II
$\mathbf{B}_8$	$-x_3 \mathbf{a}_1 + (y_3 + \frac{1}{2}) \mathbf{a}_2 - (z_3 - \frac{1}{2}) \mathbf{a}_3$	$-(ax_3 + c(z_3 - \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} + b(y_3 + \frac{1}{2}) \hat{\mathbf{y}} - c(z_3 - \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	P II
$\mathbf{B}_9$	$-x_3 \mathbf{a}_1 - y_3 \mathbf{a}_2 - z_3 \mathbf{a}_3$	$-(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} - by_3 \hat{\mathbf{y}} - cz_3 \sin \beta \hat{\mathbf{z}}$	(4e)	P II
$\mathbf{B}_{10}$	$x_3 \mathbf{a}_1 - (y_3 - \frac{1}{2}) \mathbf{a}_2 + (z_3 + \frac{1}{2}) \mathbf{a}_3$	$(ax_3 + c(z_3 + \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} - b(y_3 - \frac{1}{2}) \hat{\mathbf{y}} + c(z_3 + \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	P II

## References

- [1] H. Krebs, K.-H. Müller, and G. Zürn, *Über kristallisierte Metallpolyphosphide. I. Darstellung und Struktur des CdP<sub>4</sub>*, Z. Anorganische und Allgemeine Chemie **285**, 15–28 (1956), doi:10.1016/0022-5088(82)90210-7.
- [2] E. A. Wood, *The Conversion Factor for kX Units to Angström Units*, J. App. Phys. **18**, 929–930 (1947), doi:10.1063/1.1697570.

## Found in

- [1] U. Flörke and W. Jeitscho, *Preparation and properties of new modifications of RuP<sub>4</sub> and OsP<sub>4</sub> with CdP<sub>4</sub>-type structure*, J. Less-Common Met. **86**, 247–253 (1982), doi:10.1016/0022-5088(82)90210-7.