

Room Temperature GaMo₄S₈ Structure:

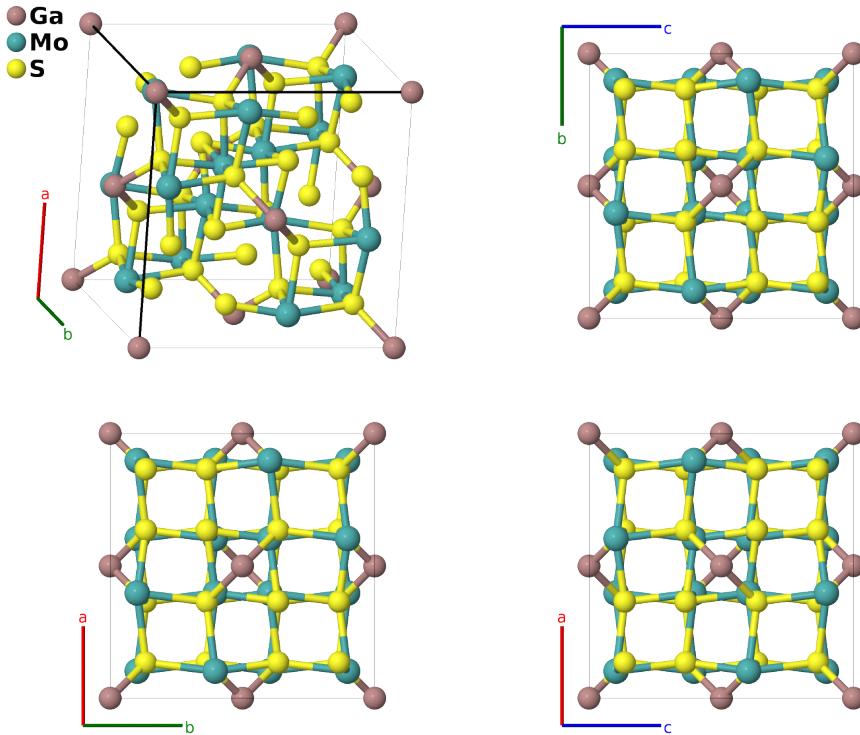
AB₄C₈_cF52_216_a_e_2e-001

This structure originally had the label `AB4C8_cF52_216_a_e_2e`. Calls to that address will be redirected here.

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<https://aflow.org/p/BEEG>

https://aflow.org/p/AB4C8_cF52_216_a_e_2e-001



Prototype GaMo₄S₈

AFLOW prototype label AB4C8_cF52_216_a_e_2e-001

ICSD 49566

Pearson symbol cF52

Space group number 216

Space group symbol $F\bar{4}3m$

AFLOW prototype command

```
aflow --proto=AB4C8_cF52_216_a_e_2e-001  
--params=a,x2,x3,x4
```

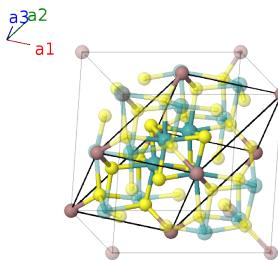
Other compounds with this structure

AlMo₄S₈, Co(Mo₂Re₂)S₈, Fe(Mo₂Re₂)S₈, GaMo₄S₄Te₄, GaMo₄S₈, GaMo₄Se₄Te₄, GaMo₄Se₈, GaMo₄Te₈, GaNb₄S₈, GaNb₄Se₈, GaNb₄Te₈, GaRe₄S₈, GaRe₄Se₈, GaRe₄Te₈, GaTa₄S₈, GaTa₄Se₈, GaTa₄Te₈, GaV₄S₈, GaV₄Se₈, GaV₄Te₈, GeMo₄S₈, GeMo₄Se₈, GeMo₄Te₈, GeNb₄S₈, GeNb₄Se₈, GeNb₄Te₈, GeRe₄S₈, GeRe₄Se₈, GeRe₄Te₈, GeTa₄S₈, GeTa₄Se₈, GeTa₄Te₈, GeV₄S₈, GeV₄Se₈, GeV₄Te₈, LaMo₄S₈, Ni(Mo₂Re₂)S₈, Zn(Mo₂Re₂)S₈

- This is the room temperature structure. Below 45K GaMo_4S_8 transforms into a rhombohedral structure.
- (Ben Yaich, 1984) do not give the lattice constant for GaMo_4S_8 . We infer it from their interatomic distances and obtain a value of $a = 9.7294\text{\AA}$. The ICSD entry uses $a = 9.74\text{\AA}$.

Face-centered Cubic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	=	0	=	0	(4a)
\mathbf{B}_2	=	$x_2 \mathbf{a}_1 + x_2 \mathbf{a}_2 + x_2 \mathbf{a}_3$	=	$ax_2 \hat{\mathbf{x}} + ax_2 \hat{\mathbf{y}} + ax_2 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_3	=	$x_2 \mathbf{a}_1 + x_2 \mathbf{a}_2 - 3x_2 \mathbf{a}_3$	=	$-ax_2 \hat{\mathbf{x}} - ax_2 \hat{\mathbf{y}} + ax_2 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_4	=	$x_2 \mathbf{a}_1 - 3x_2 \mathbf{a}_2 + x_2 \mathbf{a}_3$	=	$-ax_2 \hat{\mathbf{x}} + ax_2 \hat{\mathbf{y}} - ax_2 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_5	=	$-3x_2 \mathbf{a}_1 + x_2 \mathbf{a}_2 + x_2 \mathbf{a}_3$	=	$ax_2 \hat{\mathbf{x}} - ax_2 \hat{\mathbf{y}} - ax_2 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_6	=	$x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	=	$ax_3 \hat{\mathbf{x}} + ax_3 \hat{\mathbf{y}} + ax_3 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_7	=	$x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 - 3x_3 \mathbf{a}_3$	=	$-ax_3 \hat{\mathbf{x}} - ax_3 \hat{\mathbf{y}} + ax_3 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_8	=	$x_3 \mathbf{a}_1 - 3x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	=	$-ax_3 \hat{\mathbf{x}} + ax_3 \hat{\mathbf{y}} - ax_3 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_9	=	$-3x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	=	$ax_3 \hat{\mathbf{x}} - ax_3 \hat{\mathbf{y}} - ax_3 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_{10}	=	$x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + x_4 \mathbf{a}_3$	=	$ax_4 \hat{\mathbf{x}} + ax_4 \hat{\mathbf{y}} + ax_4 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_{11}	=	$x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 - 3x_4 \mathbf{a}_3$	=	$-ax_4 \hat{\mathbf{x}} - ax_4 \hat{\mathbf{y}} + ax_4 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_{12}	=	$x_4 \mathbf{a}_1 - 3x_4 \mathbf{a}_2 + x_4 \mathbf{a}_3$	=	$-ax_4 \hat{\mathbf{x}} + ax_4 \hat{\mathbf{y}} - ax_4 \hat{\mathbf{z}}$	(16e)
\mathbf{B}_{13}	=	$-3x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + x_4 \mathbf{a}_3$	=	$ax_4 \hat{\mathbf{x}} - ax_4 \hat{\mathbf{y}} - ax_4 \hat{\mathbf{z}}$	(16e)

References

- [1] H. B. Yaich, J. C. Jegaden, M. P. R. Chevrel, M. Sergent, A. Berton, J. Chaussy, A. K. Rastogi, and R. Tournier, *Nouveaux chalcogénures mixtes $\text{GaMo}_4(XX)_8$ ($X = S, Se, Te$) à clusters tétraédriques Mo_4* , J. Solid State Chem. **51**, 212–217 (1984), doi:10.1016/0022-4596(84)90336-0.

Found in

- [1] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).