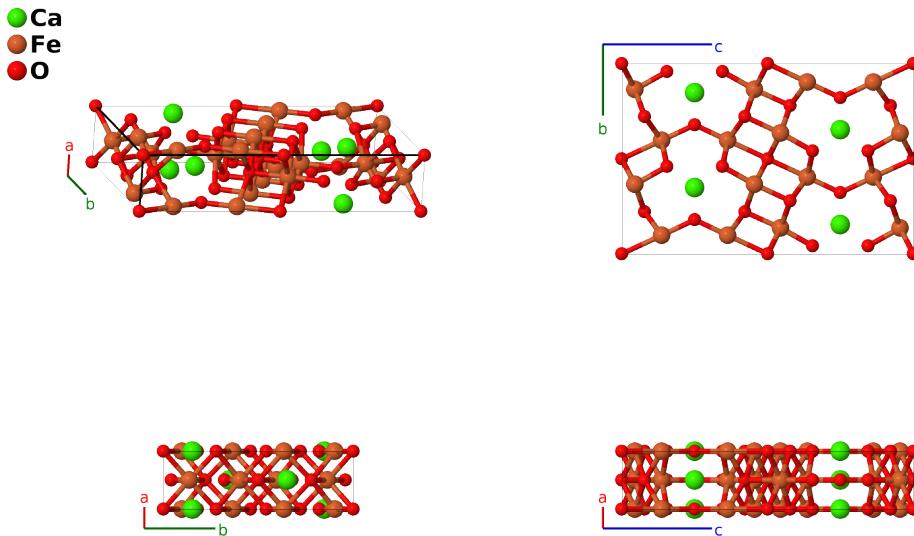


# CaFe<sub>4</sub>O<sub>6</sub> Structure: AB4C6\_oC44\_63\_c\_2f\_ac2f-001

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<https://aflow.org/p/8HU9>

[https://aflow.org/p/AB4C6\\_oC44\\_63\\_c\\_2f\\_ac2f-001](https://aflow.org/p/AB4C6_oC44_63_c_2f_ac2f-001)



<b>Prototype</b>	CaFe <sub>4</sub> O <sub>6</sub>
<b>AFLOW prototype label</b>	AB4C6_oC44_63_c_2f_ac2f-001
<b>ICSD</b>	16355
<b>Pearson symbol</b>	oC44
<b>Space group number</b>	63
<b>Space group symbol</b>	<i>Cmcm</i>
<b>AFLOW prototype command</b>	<code>aflow --proto=AB4C6_oC44_63_c_2f_ac2f-001 --params=a, b/a, c/a, y<sub>2</sub>, y<sub>3</sub>, y<sub>4</sub>, z<sub>5</sub>, z<sub>6</sub>, z<sub>7</sub>, z<sub>7</sub></code>

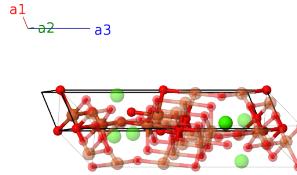
## Other compounds with this structure

Fe<sub>5</sub>O<sub>6</sub>

- We have shifted the origin so that the O-I atom is at the origin, Wyckoff position (4a), rather than (0 0 1/2), Wyckoff position (4b) as found in (Evrard, 1980).

## Base-centered Orthorhombic primitive vectors

$$\begin{aligned}
 \mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}b\hat{\mathbf{y}} \\
 \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}b\hat{\mathbf{y}} \\
 \mathbf{a}_3 &= c\hat{\mathbf{z}}
 \end{aligned}$$



## Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	= 0	= 0	(4a)	O I
$\mathbf{B}_2$	= $\frac{1}{2}\mathbf{a}_3$	= $\frac{1}{2}c\hat{\mathbf{z}}$	(4a)	O I
$\mathbf{B}_3$	= $-y_2\mathbf{a}_1 + y_2\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	= $by_2\hat{\mathbf{y}} + \frac{1}{4}c\hat{\mathbf{z}}$	(4c)	Ca I
$\mathbf{B}_4$	= $y_2\mathbf{a}_1 - y_2\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	= $-by_2\hat{\mathbf{y}} + \frac{3}{4}c\hat{\mathbf{z}}$	(4c)	Ca I
$\mathbf{B}_5$	= $-y_3\mathbf{a}_1 + y_3\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	= $by_3\hat{\mathbf{y}} + \frac{1}{4}c\hat{\mathbf{z}}$	(4c)	O II
$\mathbf{B}_6$	= $y_3\mathbf{a}_1 - y_3\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	= $-by_3\hat{\mathbf{y}} + \frac{3}{4}c\hat{\mathbf{z}}$	(4c)	O II
$\mathbf{B}_7$	= $-y_4\mathbf{a}_1 + y_4\mathbf{a}_2 + z_4\mathbf{a}_3$	= $by_4\hat{\mathbf{y}} + cz_4\hat{\mathbf{z}}$	(8f)	Fe I
$\mathbf{B}_8$	= $y_4\mathbf{a}_1 - y_4\mathbf{a}_2 + (z_4 + \frac{1}{2})\mathbf{a}_3$	= $-by_4\hat{\mathbf{y}} + c(z_4 + \frac{1}{2})\hat{\mathbf{z}}$	(8f)	Fe I
$\mathbf{B}_9$	= $-y_4\mathbf{a}_1 + y_4\mathbf{a}_2 - (z_4 - \frac{1}{2})\mathbf{a}_3$	= $by_4\hat{\mathbf{y}} - c(z_4 - \frac{1}{2})\hat{\mathbf{z}}$	(8f)	Fe I
$\mathbf{B}_{10}$	= $y_4\mathbf{a}_1 - y_4\mathbf{a}_2 - z_4\mathbf{a}_3$	= $-by_4\hat{\mathbf{y}} - cz_4\hat{\mathbf{z}}$	(8f)	Fe I
$\mathbf{B}_{11}$	= $-y_5\mathbf{a}_1 + y_5\mathbf{a}_2 + z_5\mathbf{a}_3$	= $by_5\hat{\mathbf{y}} + cz_5\hat{\mathbf{z}}$	(8f)	Fe II
$\mathbf{B}_{12}$	= $y_5\mathbf{a}_1 - y_5\mathbf{a}_2 + (z_5 + \frac{1}{2})\mathbf{a}_3$	= $-by_5\hat{\mathbf{y}} + c(z_5 + \frac{1}{2})\hat{\mathbf{z}}$	(8f)	Fe II
$\mathbf{B}_{13}$	= $-y_5\mathbf{a}_1 + y_5\mathbf{a}_2 - (z_5 - \frac{1}{2})\mathbf{a}_3$	= $by_5\hat{\mathbf{y}} - c(z_5 - \frac{1}{2})\hat{\mathbf{z}}$	(8f)	Fe II
$\mathbf{B}_{14}$	= $y_5\mathbf{a}_1 - y_5\mathbf{a}_2 - z_5\mathbf{a}_3$	= $-by_5\hat{\mathbf{y}} - cz_5\hat{\mathbf{z}}$	(8f)	Fe II
$\mathbf{B}_{15}$	= $-y_6\mathbf{a}_1 + y_6\mathbf{a}_2 + z_6\mathbf{a}_3$	= $by_6\hat{\mathbf{y}} + cz_6\hat{\mathbf{z}}$	(8f)	O III
$\mathbf{B}_{16}$	= $y_6\mathbf{a}_1 - y_6\mathbf{a}_2 + (z_6 + \frac{1}{2})\mathbf{a}_3$	= $-by_6\hat{\mathbf{y}} + c(z_6 + \frac{1}{2})\hat{\mathbf{z}}$	(8f)	O III
$\mathbf{B}_{17}$	= $-y_6\mathbf{a}_1 + y_6\mathbf{a}_2 - (z_6 - \frac{1}{2})\mathbf{a}_3$	= $by_6\hat{\mathbf{y}} - c(z_6 - \frac{1}{2})\hat{\mathbf{z}}$	(8f)	O III
$\mathbf{B}_{18}$	= $y_6\mathbf{a}_1 - y_6\mathbf{a}_2 - z_6\mathbf{a}_3$	= $-by_6\hat{\mathbf{y}} - cz_6\hat{\mathbf{z}}$	(8f)	O III
$\mathbf{B}_{19}$	= $-y_7\mathbf{a}_1 + y_7\mathbf{a}_2 + z_7\mathbf{a}_3$	= $by_7\hat{\mathbf{y}} + cz_7\hat{\mathbf{z}}$	(8f)	O IV
$\mathbf{B}_{20}$	= $y_7\mathbf{a}_1 - y_7\mathbf{a}_2 + (z_7 + \frac{1}{2})\mathbf{a}_3$	= $-by_7\hat{\mathbf{y}} + c(z_7 + \frac{1}{2})\hat{\mathbf{z}}$	(8f)	O IV
$\mathbf{B}_{21}$	= $-y_7\mathbf{a}_1 + y_7\mathbf{a}_2 - (z_7 - \frac{1}{2})\mathbf{a}_3$	= $by_7\hat{\mathbf{y}} - c(z_7 - \frac{1}{2})\hat{\mathbf{z}}$	(8f)	O IV
$\mathbf{B}_{22}$	= $y_7\mathbf{a}_1 - y_7\mathbf{a}_2 - z_7\mathbf{a}_3$	= $-by_7\hat{\mathbf{y}} - cz_7\hat{\mathbf{z}}$	(8f)	O IV

## References

- [1] O. Evrard, B. Malaman, F. Jeannot, A. Courtois, H. Alebouyeh, and R. Gerardin, *Mise en évidence de CaFe<sub>4</sub>O<sub>6</sub> et détermination des structures cristallines des ferrites de calcium CaFe<sub>2+n</sub>O<sub>4+n</sub> (n = 1, 2, 3): nouvel exemple d'intercroissance*, J. Solid State Chem. **35**, 112–119 (1980), doi:10.1016/0022-4596(80)90471-5.