

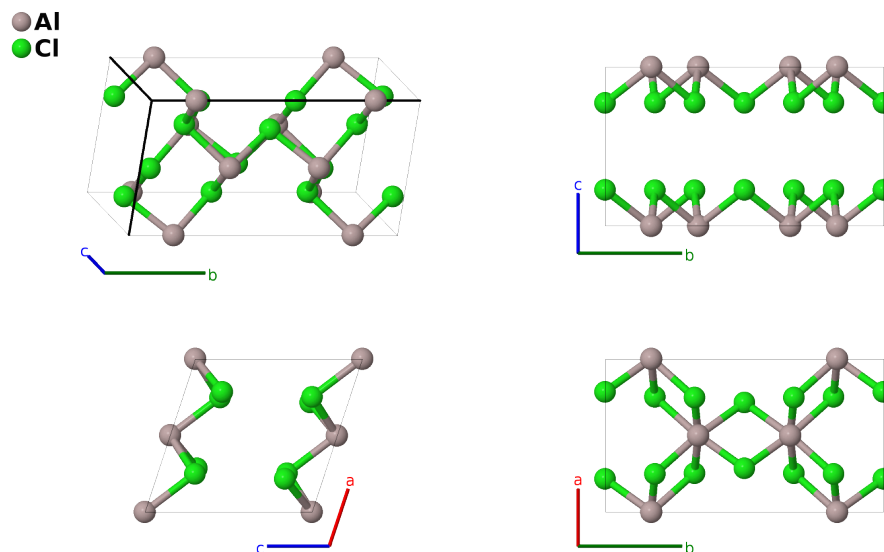
# AlCl<sub>3</sub> Structure: AB3\_mC16\_12\_g\_ij-001

This structure originally had the label AB3\_mC16\_12\_g\_ij. Calls to that address will be redirected here.

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<https://aflow.org/p/2SQ0>

[https://aflow.org/p/AB3\\_mC16\\_12\\_g\\_ij-001](https://aflow.org/p/AB3_mC16_12_g_ij-001)



Prototype	AlCl <sub>3</sub>
AFLOW prototype label	AB3_mC16_12_g_ij-001
ICSD	39566
Pearson symbol	mC16
Space group number	12
Space group symbol	$C2/m$
AFLOW prototype command	<code>aflow --proto=AB3_mC16_12_g_ij-001 --params=a, b/a, c/a, <math>\beta</math>, <math>y_1</math>, <math>x_2</math>, <math>z_2</math>, <math>x_3</math>, <math>y_3</math>, <math>z_3</math></code>

## Other compounds with this structure

DyCl<sub>3</sub>, ErCl<sub>3</sub>, HoCl<sub>3</sub>, InCl<sub>3</sub>, LuCl<sub>3</sub>,  $\alpha$ -RuCl<sub>3</sub>, TlCl<sub>3</sub>, TmCl<sub>3</sub>, YbCl<sub>3</sub>

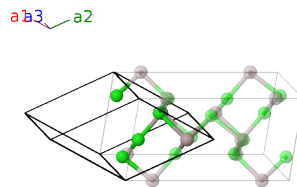
- We originally listed this as the  $D0_{15}$  structure, but this does not follow the historical record, so we have modified this page and added previous determinations of the structure as follows:
- (Laschkarew, 1930) determined that AlCl<sub>3</sub> was in space group  $P\bar{3}m1$  #164. (Hermann, 1937) designated this as the  $D0_{13}$  structure.
- (Ketelaar, 1935) determined that the structure was only pseudo-hexagonal, and place it in space group  $C2$  #5. (Gottfried, 1937) designated this as the  $D0_{15}$  structure. Neither of these works referenced the previous work.

- (Ketelaar, 1947) determined that the space group was actually  $C2/m$  #12. Here we present this using the refinement of (Trojanov, 1992).

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### Base-centered Monoclinic primitive vectors

$$\begin{aligned} \mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{1}{2}b \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \cos \beta \hat{\mathbf{x}} + c \sin \beta \hat{\mathbf{z}} \end{aligned}$$




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### Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= -y_1 \mathbf{a}_1 + y_1 \mathbf{a}_2$	$=$	$by_1 \hat{\mathbf{y}}$	(4g)	Al I
$\mathbf{B}_2$	$= y_1 \mathbf{a}_1 - y_1 \mathbf{a}_2$	$=$	$-by_1 \hat{\mathbf{y}}$	(4g)	Al I
$\mathbf{B}_3$	$= x_2 \mathbf{a}_1 + x_2 \mathbf{a}_2 + z_2 \mathbf{a}_3$	$=$	$(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} + cz_2 \sin \beta \hat{\mathbf{z}}$	(4i)	Cl I
$\mathbf{B}_4$	$= -x_2 \mathbf{a}_1 - x_2 \mathbf{a}_2 - z_2 \mathbf{a}_3$	$=$	$-(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} - cz_2 \sin \beta \hat{\mathbf{z}}$	(4i)	Cl I
$\mathbf{B}_5$	$= (x_3 - y_3) \mathbf{a}_1 + (x_3 + y_3) \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} + by_3 \hat{\mathbf{y}} + cz_3 \sin \beta \hat{\mathbf{z}}$	(8j)	Cl II
$\mathbf{B}_6$	$= -(x_3 + y_3) \mathbf{a}_1 - (x_3 - y_3) \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$-(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} + by_3 \hat{\mathbf{y}} - cz_3 \sin \beta \hat{\mathbf{z}}$	(8j)	Cl II
$\mathbf{B}_7$	$= -(x_3 - y_3) \mathbf{a}_1 - (x_3 + y_3) \mathbf{a}_2 - z_3 \mathbf{a}_3$	$=$	$-(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} - by_3 \hat{\mathbf{y}} - cz_3 \sin \beta \hat{\mathbf{z}}$	(8j)	Cl II
$\mathbf{B}_8$	$= (x_3 + y_3) \mathbf{a}_1 + (x_3 - y_3) \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$(ax_3 + cz_3 \cos \beta) \hat{\mathbf{x}} - by_3 \hat{\mathbf{y}} + cz_3 \sin \beta \hat{\mathbf{z}}$	(8j)	Cl II

### References

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- [2] W. E. Laschkarew, *Zur Struktur  $AlCl_3$* , Z. Anorganische und Allgemeine Chemie **193**, 270–276 (1930), doi:10.1002/zaac.19301930123.
- [3] C. Hermann, O. Lohrmann, and H. Philipp, eds., *Strukturbericht Band II 1928-1932* (Akademische Verlagsgesellschaft M. B. H., Leipzig, 1937).
- [4] J. A. A. Ketelaar, *Die Kristallstruktur der Aluminiumhalogenide II*, Z. Krystallogr. **90**, 237–255 (1935), doi:10.1524/zkri.1935.90.1.237.
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### Found in

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