

β -BaPbO₃ Perovskite Structure:

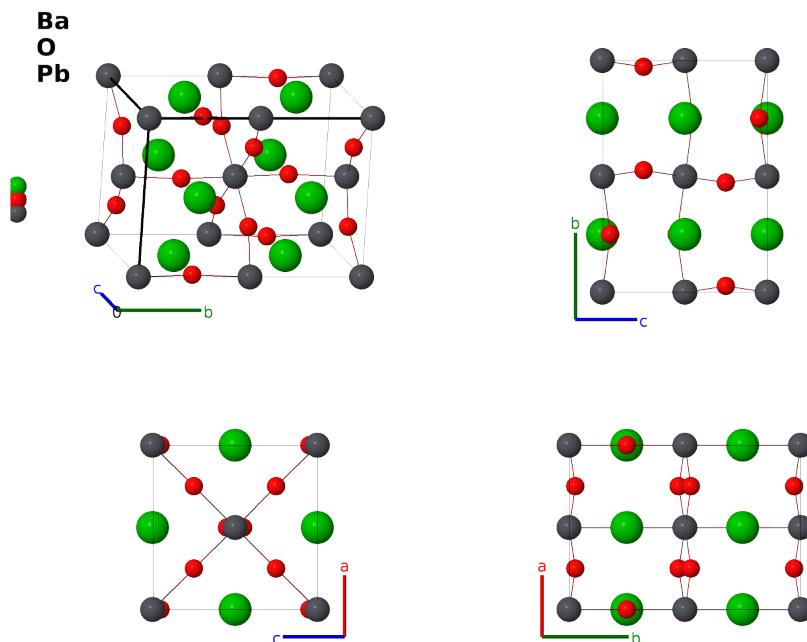
AB3C_oI20_74_e_eg_a-001

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<https://aflow.org/prototype-encyclopedia/Z961>

https://aflow.org/prototype-encyclopedia/AB3C_oI20_74_e_eg_a-001



Prototype	BaO ₃ Pb
AFLOW prototype label	AB3C_oI20_74_e_eg_a-001
ICSD	78681
CCDC	1638954
Pearson symbol	oI20
Space group number	74
Space group symbol	<i>Imma</i>
AFLOW prototype command	<code>aflow --proto=AB3C_oI20_74_e_eg_a-001 --params=a,b/a,c/a,z₂,z₃,y₄</code>

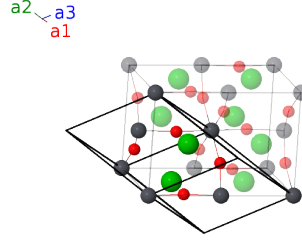
Other compounds with this structure

AlCeO₃, BaCeO₃, BaPrO₃, BaTbO₃, BaThO₃, EuNbO₃, PuAlO₃, SrMoO₃, SrSnO₃, SrTcO₃, SrZrO₃, Ba(Ce_{1-x}In_x)O₃, Ba(Ce_{1-x}Zr_x)O₃, Ba(Pb_{1-x}Bi_x)O₃, Ba(Pb_{1-x}Tl_x)O₃, Ba(Sb_{1-x}Tl_x)O₃, (Ba_{1-x}K_x)BiO₃, (Bi_{1-x}La_x)FeO₃, (La_{1-x}Ba_x)MnO₃, (Na_{1-x}Pr_x)TiO₃, (Nd_{1-x}Sr_x)MnO₃, (Pr_{1-x}Ba_x)MnO₃, (Sr_{1-x}Ba_x)SnO₃

- As with most perovskites, BaPbO₃ has a number of phase transitions. The temperature-driven transitions are summarized in (Fu, 1995), (Moussa, 2001), and (Hester, 2002). We will follow our usual convention and label the phases by Greek letters, increasing with temperature.
 - The low temperature structure is monoclinic α -BaPbO₃.
 - At some point above 26 K (Ritter, 1989) we find orthorhombic β -BaPbO₃ (this structure).
 - At 548 K it transforms into the tetragonal γ -BaPbO₃ in the γ -SrZrO₃ structure.
 - Finally, above 723-725 K δ -BaPbO₃ is in the cubic perovskite ($E2_1$) structure.
- All of the metallic sites may be alloyed, we have only listed a small number of possibilities. The oxygen sites may be partially filled, or partially replaced by nitrogen or fluorine.
- (Fu, 1995) gave the structural information in the $Ibmm$ setting of space group #74. We used AFLOW to transform this to the standard $Imma$ setting. This requires change of the axes so that $(xyz) \rightarrow (yzx)$.

Body-centered Orthorhombic primitive vectors

$$\begin{aligned}
 \mathbf{a}_1 &= -\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}b\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\
 \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}b\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\
 \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}b\hat{\mathbf{y}} - \frac{1}{2}c\hat{\mathbf{z}}
 \end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= 0$	$=$	0	(4a)	Pb I
$\mathbf{B}_2$	$= \frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{2}b\hat{\mathbf{y}}$	(4a)	Pb I
$\mathbf{B}_3$	$= (z_2 + \frac{1}{4})\mathbf{a}_1 + z_2\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	$=$	$\frac{1}{4}b\hat{\mathbf{y}} + cz_2\hat{\mathbf{z}}$	(4e)	Ba I
$\mathbf{B}_4$	$= -(z_2 - \frac{3}{4})\mathbf{a}_1 - z_2\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	$=$	$\frac{3}{4}b\hat{\mathbf{y}} - cz_2\hat{\mathbf{z}}$	(4e)	Ba I
$\mathbf{B}_5$	$= (z_3 + \frac{1}{4})\mathbf{a}_1 + z_3\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	$=$	$\frac{1}{4}b\hat{\mathbf{y}} + cz_3\hat{\mathbf{z}}$	(4e)	O I
$\mathbf{B}_6$	$= -(z_3 - \frac{3}{4})\mathbf{a}_1 - z_3\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	$=$	$\frac{3}{4}b\hat{\mathbf{y}} - cz_3\hat{\mathbf{z}}$	(4e)	O I
$\mathbf{B}_7$	$= (y_4 + \frac{1}{4})\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + (y_4 + \frac{1}{4})\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{\mathbf{x}} + by_4\hat{\mathbf{y}} + \frac{1}{4}c\hat{\mathbf{z}}$	(8g)	O II
$\mathbf{B}_8$	$= -(y_4 - \frac{3}{4})\mathbf{a}_1 - (y_4 - \frac{1}{4})\mathbf{a}_3$	$=$	$-\frac{1}{4}a\hat{\mathbf{x}} - b(y_4 - \frac{1}{2})\hat{\mathbf{y}} + \frac{1}{4}c\hat{\mathbf{z}}$	(8g)	O II
$\mathbf{B}_9$	$= -(y_4 - \frac{3}{4})\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 - (y_4 - \frac{3}{4})\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{\mathbf{x}} - b(y_4 - \frac{1}{2})\hat{\mathbf{y}} + \frac{1}{4}c\hat{\mathbf{z}}$	(8g)	O II
$\mathbf{B}_{10}$	$= (y_4 + \frac{1}{4})\mathbf{a}_1 + (y_4 + \frac{3}{4})\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{\mathbf{x}} + b(y_4 + \frac{1}{2})\hat{\mathbf{y}} - \frac{1}{4}c\hat{\mathbf{z}}$	(8g)	O II

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Found in

- [1] J. R. Hester, C. J. Howard, B. J. Kennedy, and R. Macquart, *High-Temperature Structural Studies of $SrPbO_3$ and $BaPbO_3$* , Aust. J. Chem. **55**, 543–545 (2008), doi:10.1071/CH02023.

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