

α -BaPbO₃ Perovskite Structure:

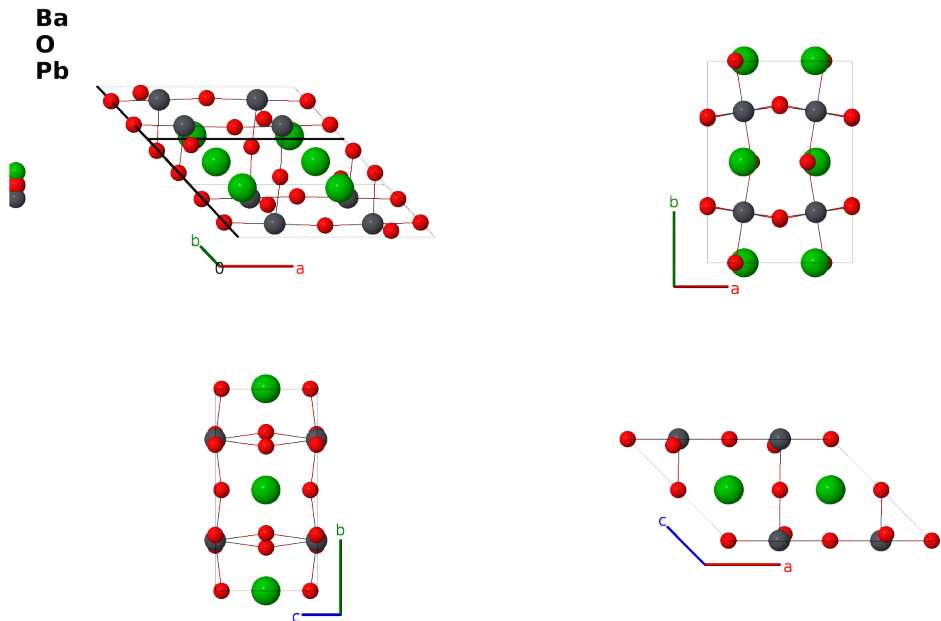
AB3C_mC20_12_i_ghi_e-001

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<https://aflow.org/prototype-encyclopedia/729H>

https://aflow.org/prototype-encyclopedia/AB3C_mC20_12_i_ghi_e-001



Prototype	BaO ₃ Pb
AFLOW prototype label	AB3C_mC20_12_i_ghi_e-001
ICSD	94313
CCDC	1653754
Pearson symbol	mC20
Space group number	12
Space group symbol	$C2/m$
AFLOW prototype command	<code>aflow --proto=AB3C_mC20_12_i_ghi_e-001</code> <code>--params=a, b/a, c/a, β, y_2, y_3, x_4, z_4, x_5, z_5</code>

Other compounds with this structure

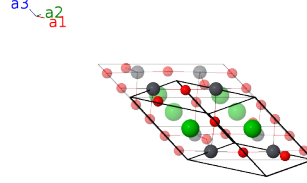
Al(Pr_{1-x}La_x)O₃, Ba(Ce_{1-x}Nd_x)O₃, Ba(Ce_{1-x}Y_x)O₃, Ba(Pb_{1-x}Bi_x)O₃, LaTa(O, N)₃, SrTa(O, N)₃

- As with most perovskites, BaPbO₃ has a number of phase transitions. The temperature-driven transitions are summarized in (Fu, 1995), (Moussa, 2001), and (Hestor, 2002). We will follow our usual convention and label the phases by Greek letters, increasing with temperature.

- The low temperature structure is monoclinic α -BaPbO₃ (this structure).
- At some point above 26 K (Ritter, 1989) we find orthorhombic β -BaPbO₃.
- At 548 K it transforms into the tetragonal γ -BaPbO₃ in the γ -SrZrO₃ structure.
- Finally, above 723-725 K δ -BaPbO₃ is in the cubic perovskite ($E2_1$) structure.
- The oxygen atoms may be replaced by nitrogen in some systems.
- We use the data taken by (Moussa, 2001) at 15 K.

Base-centered Monoclinic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{1}{2}b \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \cos \beta \hat{\mathbf{x}} + c \sin \beta \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{1}{4}b \hat{\mathbf{y}}$	(4e)	Pb I
$\mathbf{B}_2$	$= \frac{1}{2} \mathbf{a}_1$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{1}{4}b \hat{\mathbf{y}}$	(4e)	Pb I
$\mathbf{B}_3$	$= -y_2 \mathbf{a}_1 + y_2 \mathbf{a}_2$	$=$	$by_2 \hat{\mathbf{y}}$	(4g)	O I
$\mathbf{B}_4$	$= y_2 \mathbf{a}_1 - y_2 \mathbf{a}_2$	$=$	$-by_2 \hat{\mathbf{y}}$	(4g)	O I
$\mathbf{B}_5$	$= -y_3 \mathbf{a}_1 + y_3 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \cos \beta \hat{\mathbf{x}} + by_3 \hat{\mathbf{y}} + \frac{1}{2}c \sin \beta \hat{\mathbf{z}}$	(4h)	O II
$\mathbf{B}_6$	$= y_3 \mathbf{a}_1 - y_3 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \cos \beta \hat{\mathbf{x}} - by_3 \hat{\mathbf{y}} + \frac{1}{2}c \sin \beta \hat{\mathbf{z}}$	(4h)	O II
$\mathbf{B}_7$	$= x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$(ax_4 + cz_4 \cos \beta) \hat{\mathbf{x}} + cz_4 \sin \beta \hat{\mathbf{z}}$	(4i)	Ba I
$\mathbf{B}_8$	$= -x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 - z_4 \mathbf{a}_3$	$=$	$-(ax_4 + cz_4 \cos \beta) \hat{\mathbf{x}} - cz_4 \sin \beta \hat{\mathbf{z}}$	(4i)	Ba I
$\mathbf{B}_9$	$= x_5 \mathbf{a}_1 + x_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$(ax_5 + cz_5 \cos \beta) \hat{\mathbf{x}} + cz_5 \sin \beta \hat{\mathbf{z}}$	(4i)	O III
$\mathbf{B}_{10}$	$= -x_5 \mathbf{a}_1 - x_5 \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$-(ax_5 + cz_5 \cos \beta) \hat{\mathbf{x}} - cz_5 \sin \beta \hat{\mathbf{z}}$	(4i)	O III

References

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- [2] W. T. Fu and D. J. W. Ijdo, *A comparative study on the structure of APbO₃ (A=Ba,Sr)*, Solid State Commun. **95**, 581–585 (1995), doi:10.1016/0038-1098(95)00334-7.
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Found in

[1] *AFLOW Database*, <https://aflow.org>. BaPbO₃, ICSD entry 94313.

First cited in

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