

# PrNiO<sub>3</sub> Structure:

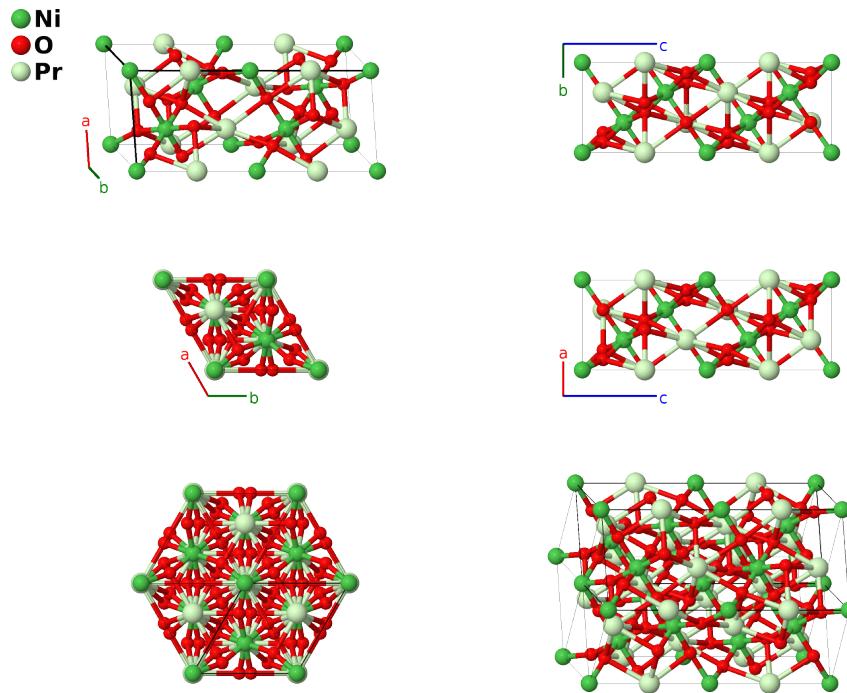
## AB3C\_hR10\_167\_b\_e\_a-001

This structure originally had the label `AB3C_hR10_167_b_e_a`. Calls to that address will be redirected here.

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<https://aflow.org/p/XMCY>

[https://aflow.org/p/AB3C\\_hR10\\_167\\_b\\_e\\_a-001](https://aflow.org/p/AB3C_hR10_167_b_e_a-001)



**Prototype** NiO<sub>3</sub>Pr

**AFLOW prototype label** AB3C\_hR10\_167\_b\_e\_a-001

**ICSD** 69186

**Pearson symbol** hR10

**Space group number** 167

**Space group symbol**  $R\bar{3}c$

**AFLOW prototype command**

```
aflow --proto=AB3C_hR10_167_b_e_a-001  
--params=a, c/a, x3
```

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### Other compounds with this structure

LaAlO<sub>3</sub>, LaNiO<sub>3</sub>

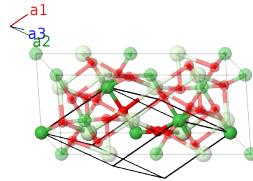
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- This structure has the same crystal structure and occupies the same Wyckoff positions as Calcite (CaCO<sub>3</sub>,  $G0_1$ , ABC3\_hR10\_167\_a\_b\_e), but  $c/a$  and  $x_3$  are different enough to warrant calling this a new structure.

- This is the high-temperature form of  $\text{PrNiO}_3$ , and we present the data collected at 500°C, below which it transforms into the orthorhombic perovskite structure (`AB3C_oP20.62_c_cd_a`).
- Hexagonal settings of this structure can be obtained with the option `--hex`.

## Rhombohedral primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + \frac{1}{3}c\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{\sqrt{3}}a\hat{\mathbf{y}} + \frac{1}{3}c\hat{\mathbf{z}} \\ \mathbf{a}_3 &= -\frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + \frac{1}{3}c\hat{\mathbf{z}}\end{aligned}$$



## Basis vectors

	Lattice coordinates	=	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$\frac{1}{4}\mathbf{a}_1 + \frac{1}{4}\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	=	$\frac{1}{4}c\hat{\mathbf{z}}$	(2a)	Pr I
$\mathbf{B}_2$	$\frac{3}{4}\mathbf{a}_1 + \frac{3}{4}\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	=	$\frac{3}{4}c\hat{\mathbf{z}}$	(2a)	Pr I
$\mathbf{B}_3$	0	=	0	(2b)	Ni I
$\mathbf{B}_4$	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{2}c\hat{\mathbf{z}}$	(2b)	Ni I
$\mathbf{B}_5$	$x_3\mathbf{a}_1 - (x_3 - \frac{1}{2})\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	=	$\frac{1}{8}a(4x_3 - 1)\hat{\mathbf{x}} - \frac{\sqrt{3}}{8}a(4x_3 - 1)\hat{\mathbf{y}} + \frac{1}{4}c\hat{\mathbf{z}}$	(6e)	O I
$\mathbf{B}_6$	$\frac{1}{4}\mathbf{a}_1 + x_3\mathbf{a}_2 - (x_3 - \frac{1}{2})\mathbf{a}_3$	=	$\frac{1}{8}a(4x_3 - 1)\hat{\mathbf{x}} + \frac{\sqrt{3}}{8}a(4x_3 - 1)\hat{\mathbf{y}} + \frac{1}{4}c\hat{\mathbf{z}}$	(6e)	O I
$\mathbf{B}_7$	$-(x_3 - \frac{1}{2})\mathbf{a}_1 + \frac{1}{4}\mathbf{a}_2 + x_3\mathbf{a}_3$	=	$-a(x_3 - \frac{1}{4})\hat{\mathbf{x}} + \frac{1}{4}c\hat{\mathbf{z}}$	(6e)	O I
$\mathbf{B}_8$	$-x_3\mathbf{a}_1 + (x_3 + \frac{1}{2})\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	=	$-\frac{1}{8}a(4x_3 + 3)\hat{\mathbf{x}} + \frac{\sqrt{3}}{24}a(12x_3 + 1)\hat{\mathbf{y}} + \frac{5}{12}c\hat{\mathbf{z}}$	(6e)	O I
$\mathbf{B}_9$	$\frac{3}{4}\mathbf{a}_1 - x_3\mathbf{a}_2 + (x_3 + \frac{1}{2})\mathbf{a}_3$	=	$-\frac{1}{8}a(4x_3 - 1)\hat{\mathbf{x}} - \frac{\sqrt{3}}{24}a(12x_3 + 5)\hat{\mathbf{y}} + \frac{5}{12}c\hat{\mathbf{z}}$	(6e)	O I
$\mathbf{B}_{10}$	$(x_3 + \frac{1}{2})\mathbf{a}_1 + \frac{3}{4}\mathbf{a}_2 - x_3\mathbf{a}_3$	=	$a(x_3 + \frac{1}{4})\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + \frac{5}{12}c\hat{\mathbf{z}}$	(6e)	O I

## References

[1] T. C. Huang, W. Parrish, H. Toraya, P. Lacorre, and J. B. Torrance, *High-Temperature Crystal Structures of Orthorhombic and Rhombohedral  $\text{PrNiO}_3$* , Mater. Res. Bull. **25**, 1091–1098 (1990), doi:10.1016/0025-5408(90)90138-R.

## Found in

[1] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).