

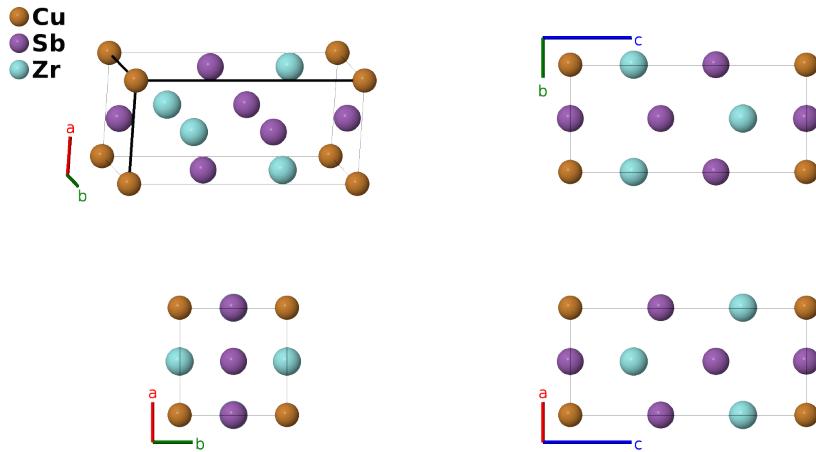
# Zr<sub>2</sub>CuSb<sub>3</sub> Structure:

## AB<sub>3</sub>C<sub>2</sub>\_tP6\_115\_a\_bg\_g-001

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<https://aflow.org/p/NQAB>

[https://aflow.org/p/AB3C2\\_tP6\\_115\\_a\\_bg\\_g-001](https://aflow.org/p/AB3C2_tP6_115_a_bg_g-001)



Prototype	CuSb <sub>3</sub> Zr <sub>2</sub>
AFLOW prototype label	AB3C2_tP6_115_a_bg_g-001
ICSD	93242
Pearson symbol	tP6
Space group number	115
Space group symbol	$P\bar{4}m2$
AFLOW prototype command	aflow --proto=AB3C2_tP6_115_a_bg_g-001 --params=a, c/a, z <sub>3</sub> , z <sub>4</sub>

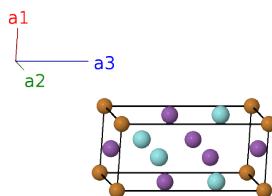
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**Other compounds with this structure**  
Hf<sub>2</sub>CuSb<sub>3</sub>, Hf<sub>2</sub>GaSb<sub>3</sub>, Ti<sub>2</sub>CuSb<sub>3</sub>, Zr<sub>2</sub>PdSb<sub>3</sub>

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Simple Tetragonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$




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Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1 =$	0	=	0	(1a)	Cu I
$\mathbf{B}_2 =$	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}}$	(1b)	Sb I
$\mathbf{B}_3 =$	$\frac{1}{2}\mathbf{a}_2 + z_3\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{y}} + cz_3\hat{\mathbf{z}}$	(2g)	Sb II
$\mathbf{B}_4 =$	$\frac{1}{2}\mathbf{a}_1 - z_3\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} - cz_3\hat{\mathbf{z}}$	(2g)	Sb II
$\mathbf{B}_5 =$	$\frac{1}{2}\mathbf{a}_2 + z_4\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{y}} + cz_4\hat{\mathbf{z}}$	(2g)	Zr I
$\mathbf{B}_6 =$	$\frac{1}{2}\mathbf{a}_1 - z_4\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} - cz_4\hat{\mathbf{z}}$	(2g)	Zr I

## References

- [1] N. Koblyuk, G. Melnyk, L. Romaka, O. I. Bodak, and D. Fruchart, *Crystal structure of Zr<sub>2</sub>CuSb<sub>3</sub> and related compounds*, J. Alloys Compd. **317-318**, 284–286 (2001), doi:10.1016/S0925-8388(00)01349-9.