

LaLi₃Sb₂ Structure:

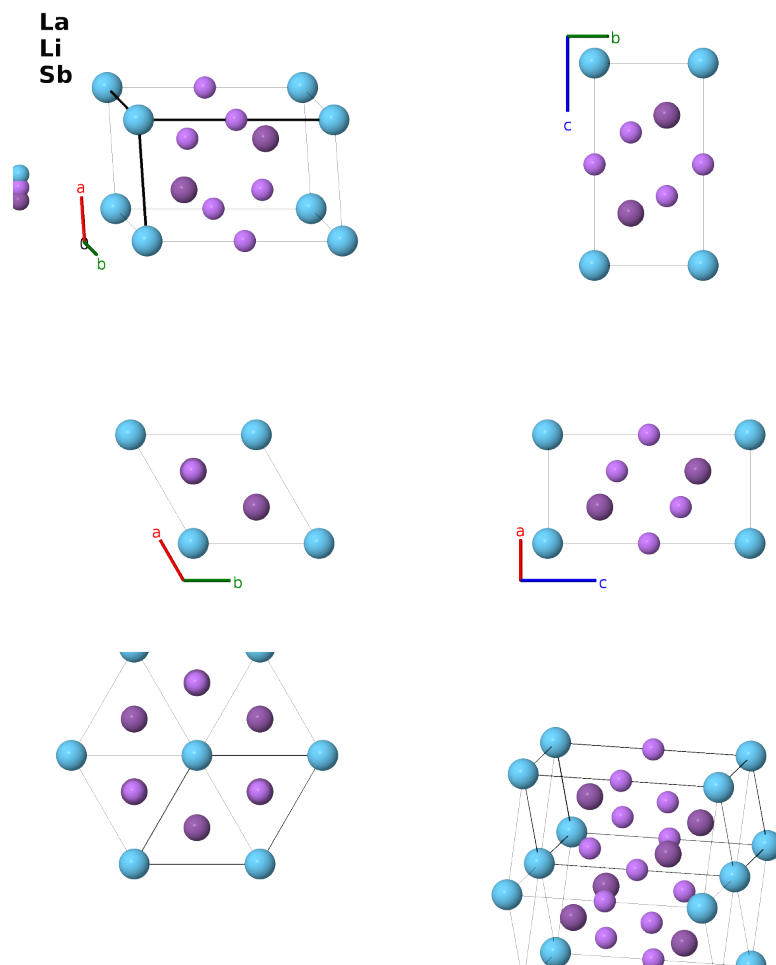
AB3C2_hP6_164_a_bd_d-001

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<https://aflow.org/prototype-encyclopedia/J1XS>

https://aflow.org/prototype-encyclopedia/AB3C2_hP6_164_a_bd_d-001



Prototype	LaLi ₃ Sb ₂
AFLOW prototype label	AB3C2_hP6_164_a_bd_d-001
ICSD	49625
CCDC	1614794
Pearson symbol	hP6
Space group number	164
Space group symbol	$P\bar{3}m1$

AFLOW prototype command `aflow --proto=AB3C2_hP6_164_a_bd_d-001`
 `--params=a, c/a, z3, z4`

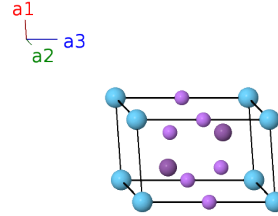
Other compounds with this structure

CeLi₃Bi₂, CeLi₃Sb₂, DyLi₃Sb₂, GdLi₃Bi₂, HoLi₃Sb₂, LaLi₃As₂, LaLi₃Bi₂, LaLi₃P₂, NdLi₃As₂, NdLi₃Bi₂, NdLi₃Sb₂, PrLi₃Bi₂, PrLi₃Sb₂, SmLi₃Bi₂, SmLi₃Sb₂, TbLi₃Bi₂, YLi₃Bi₂, YLi₃Sb₂

- (Grund, 1984) refer to this as a “filled” LiAl₂Si₂ structure. That compound is isostructural with Ce₂O₂S. Removing the (1b) atom reduces this structure to the Ce₂O₂S prototype.
 - The quaternary form of this structure, with the (1b) site containing a different species, is TbLiCu₂P₂.
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Trigonal (Hexagonal) primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_3 &= c\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$=$	0	$=$	0	(1a) La I
$\mathbf{B}_2$	$=$	$\frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{2}c\hat{\mathbf{z}}$	(1b) Li I
$\mathbf{B}_3$	$=$	$\frac{1}{3}\mathbf{a}_1 + \frac{2}{3}\mathbf{a}_2 + z_3\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + cz_3\hat{\mathbf{z}}$	(2d) Li II
$\mathbf{B}_4$	$=$	$\frac{2}{3}\mathbf{a}_1 + \frac{1}{3}\mathbf{a}_2 - z_3\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} - cz_3\hat{\mathbf{z}}$	(2d) Li II
$\mathbf{B}_5$	$=$	$\frac{1}{3}\mathbf{a}_1 + \frac{2}{3}\mathbf{a}_2 + z_4\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + cz_4\hat{\mathbf{z}}$	(2d) Sb I
$\mathbf{B}_6$	$=$	$\frac{2}{3}\mathbf{a}_1 + \frac{1}{3}\mathbf{a}_2 - z_4\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} - cz_4\hat{\mathbf{z}}$	(2d) Sb I

References

- [1] I. Grund, H.-U. Schuster, and P. Müller, *Ternäre Verbindungen von Lithium mit Yttrium, Lanthan bzw. Neodym und 5b-Elementen im “aufgefüllten” CaAl₂Si₂-Typ*, Z. Anorganische und Allgemeine Chemie **515**, 151–158 (1984), doi:10.1107/S2053229615016393.

Found in

- [1] J. Prakash, M. C. Schäfer, and S. Bobev, *Synthesis and structure determination of seven ternary bismuthides: crystal chemistry of the RELi₃Bi₂ family (RE = La-Nd, Sm, Gd, and Tb)*, Acta Crystallogr. Sect. C **71**, 894–899 (2015), doi:10.1107/S2053229615016393.

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