

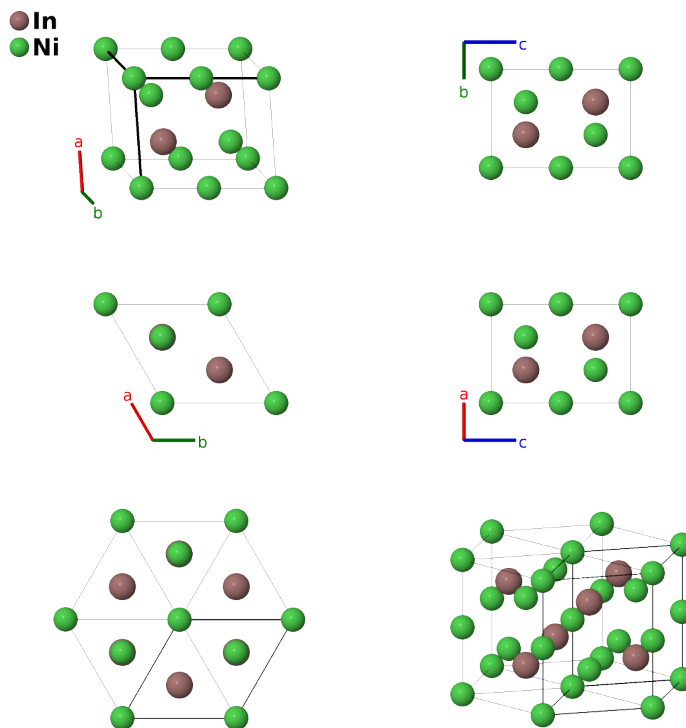
# InNi<sub>2</sub> (*B*8<sub>2</sub>) Structure: AB2\_hP6\_194\_c\_ad-001

This structure originally had the label AB2\_hP6\_194\_c\_ad. Calls to that address will be redirected here.

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<https://afLOW.org/p/X5ZH>

[https://afLOW.org/p/AB2\\_hP6\\_194\\_c\\_ad-001](https://afLOW.org/p/AB2_hP6_194_c_ad-001)



Prototype	InNi <sub>2</sub>
AFLOW prototype label	AB2_hP6_194_c_ad-001
<i>Strukturbericht</i> designation	<i>B</i> 8 <sub>2</sub>
ICSD	59436
Pearson symbol	hP6
Space group number	194
Space group symbol	<i>P</i> 6 <sub>3</sub> / <i>mmc</i>
AFLOW prototype command	<code>afLOW --proto=AB2_hP6_194_c_ad-001 --params=a,c/a</code>

## Other compounds with this structure

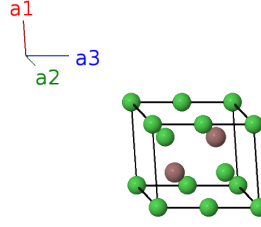
AlZr<sub>2</sub>, BiIn<sub>2</sub>, GaTi<sub>2</sub>, SnTi<sub>2</sub>, BeHfSi, BeSiZr, CoFeGe, CoNiGe, CoNiSn, FeGeMn, FeGeNi, GeMnNi

- Replacing the Ni-II atoms with In transforms the crystal into the  $C32$  (hexagonal  $\omega$ ) phase.

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### Hexagonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$




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### Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$=$	$0$	$=$	$0$	(2a) Ni I
$\mathbf{B}_2$	$=$	$\frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} c \hat{\mathbf{z}}$	(2a) Ni I
$\mathbf{B}_3$	$=$	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6} a \hat{\mathbf{y}} + \frac{1}{4} c \hat{\mathbf{z}}$	(2c) In I
$\mathbf{B}_4$	$=$	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6} a \hat{\mathbf{y}} + \frac{3}{4} c \hat{\mathbf{z}}$	(2c) In I
$\mathbf{B}_5$	$=$	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6} a \hat{\mathbf{y}} + \frac{3}{4} c \hat{\mathbf{z}}$	(2d) Ni II
$\mathbf{B}_6$	$=$	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6} a \hat{\mathbf{y}} + \frac{1}{4} c \hat{\mathbf{z}}$	(2d) Ni II

### References

- [1] M. Ellner, *Über die kristallchemischen parameter der Ni-, Co- und Fe-haltigen phasen vom NiAs-Typ*, J. Less-Common Met. **48**, 21–52 (1976), doi:10.1016/0022-5088(76)90231-9.

### Found in

- [1] P. Villars, *Ni<sub>2</sub>In Crystal Structure* (2016). PAULING FILE in: Inorganic Solid Phases, SpringerMaterials (online database), Springer, Heidelberg (ed.) SpringerMaterials.