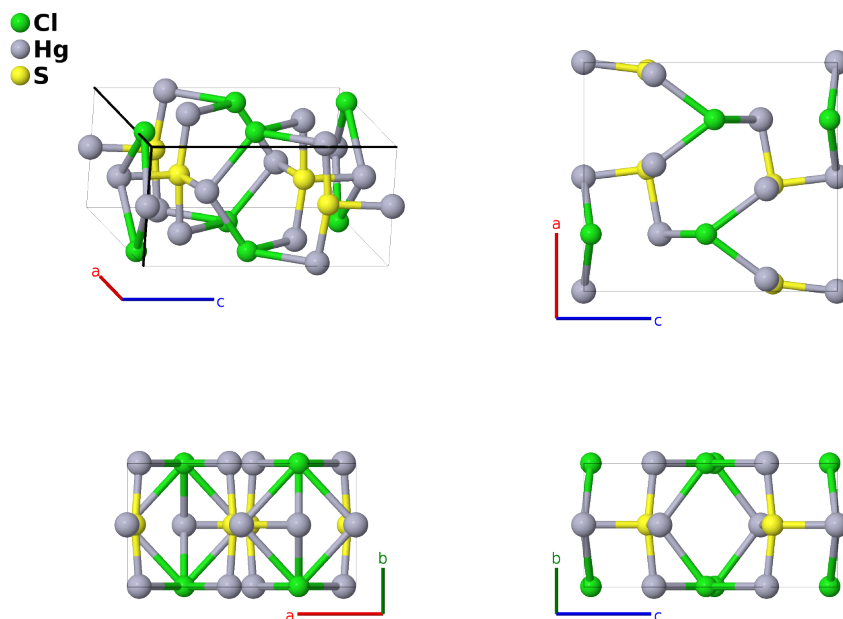


Kenhsuite (γ -Hg₃S₂Cl₂) Structure: AB2C_oP16_51_2e_bfi_j-001

Cite this page as: H. Eckert, S. Divilov, A. Zettel, M. J. Mehl, D. Hicks, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*. In preparation.

<https://afLOW.org/p/LQ5X>

https://afLOW.org/p/AB2C_oP16_51_2e_bfi_j-001



Prototype	Cl ₂ Hg ₃ S ₂
AFLOW prototype label	AB2C_oP16_51_2e_bfi_j-001
Mineral name	kenhsuite
ICSD	29252
Pearson symbol	oP16
Space group number	51
Space group symbol	<i>Pmma</i>
AFLOW prototype command	<code>afLOW --proto=AB2C_oP16_51_2e_bfi_j-001 --params=a, b/a, c/a, z₂, z₃, z₄, x₅, z₅, x₆, z₆</code>

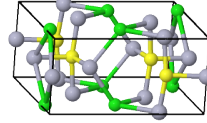
- Hg₃Cl₂S₂ is found in three forms (Carlson, 1967):
 - Corderoite (α -Hg₃Cl₂S₂), the cubic ground state.
 - β -Hg₃Cl₂S₂, which appears above 340°C, another cubic phase with a much larger unit cell.
 - Kenhsuite (γ -Hg₃Cl₂S₂), which on average has an orthorhombic lattice. This state is apparently metastable. (this structure)

- (Đurovič, 1961) found that Kenhsuite was composed of blocks of periodic structure in space group $C2/m$ #12, with lattice constants $(a', b', c') = (2a, 2b, c)$, where (a, b, c) are the lattice constants given here. This page shows what Đurovič refers to as the “composite structure,” essentially averaging all of the periodic structures. The data for this structure was originally given in the $Pbmm$ setting of space group #51. We used FINDSYM to transform it to the standard $Pmma$ setting, which involved a rotation and a translation.
- The Hg-III (4i) site is only half occupied, giving the observed stoichiometry.

Simple Orthorhombic primitive vectors



$$\begin{aligned} \mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}} \end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	$= \frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{2} b \hat{\mathbf{y}}$	(2b)	Hg I
\mathbf{B}_2	$= \frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}}$	(2b)	Hg I
\mathbf{B}_3	$= \frac{1}{4} \mathbf{a}_1 + z_2 \mathbf{a}_3$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} + cz_2 \hat{\mathbf{z}}$	(2e)	Cl I
\mathbf{B}_4	$= \frac{3}{4} \mathbf{a}_1 - z_2 \mathbf{a}_3$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} - cz_2 \hat{\mathbf{z}}$	(2e)	Cl I
\mathbf{B}_5	$= \frac{1}{4} \mathbf{a}_1 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} + cz_3 \hat{\mathbf{z}}$	(2e)	Cl II
\mathbf{B}_6	$= \frac{3}{4} \mathbf{a}_1 - z_3 \mathbf{a}_3$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} - cz_3 \hat{\mathbf{z}}$	(2e)	Cl II
\mathbf{B}_7	$= \frac{1}{4} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}} + cz_4 \hat{\mathbf{z}}$	(2f)	Hg II
\mathbf{B}_8	$= \frac{3}{4} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 - z_4 \mathbf{a}_3$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}} - cz_4 \hat{\mathbf{z}}$	(2f)	Hg II
\mathbf{B}_9	$= x_5 \mathbf{a}_1 + z_5 \mathbf{a}_3$	$=$	$ax_5 \hat{\mathbf{x}} + cz_5 \hat{\mathbf{z}}$	(4i)	Hg III
\mathbf{B}_{10}	$= -(x_5 - \frac{1}{2}) \mathbf{a}_1 + z_5 \mathbf{a}_3$	$=$	$-a(x_5 - \frac{1}{2}) \hat{\mathbf{x}} + cz_5 \hat{\mathbf{z}}$	(4i)	Hg III
\mathbf{B}_{11}	$= -x_5 \mathbf{a}_1 - z_5 \mathbf{a}_3$	$=$	$-ax_5 \hat{\mathbf{x}} - cz_5 \hat{\mathbf{z}}$	(4i)	Hg III
\mathbf{B}_{12}	$= (x_5 + \frac{1}{2}) \mathbf{a}_1 - z_5 \mathbf{a}_3$	$=$	$a(x_5 + \frac{1}{2}) \hat{\mathbf{x}} - cz_5 \hat{\mathbf{z}}$	(4i)	Hg III
\mathbf{B}_{13}	$= x_6 \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$ax_6 \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(4j)	S I
\mathbf{B}_{14}	$= -(x_6 - \frac{1}{2}) \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$-a(x_6 - \frac{1}{2}) \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(4j)	S I
\mathbf{B}_{15}	$= -x_6 \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 - z_6 \mathbf{a}_3$	$=$	$-ax_6 \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}} - cz_6 \hat{\mathbf{z}}$	(4j)	S I
\mathbf{B}_{16}	$= (x_6 + \frac{1}{2}) \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 - z_6 \mathbf{a}_3$	$=$	$a(x_6 + \frac{1}{2}) \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}} - cz_6 \hat{\mathbf{z}}$	(4j)	S I

References

- [1] S. Đurovič, *The Crystal Structure of γ -Hg₃S₂Cl₂*, Acta Crystallogr. Sect. B **24**, 1661–1670 (1961), doi:10.1107/S0567740868004814.
- [2] E. H. Carlson, *The growth of HgS and Hg₃S₂Cl₂ single crystals by a vapor phase method* **1**, 271–277 (1967), doi:10.1016/0022-0248(67)90033-4.

Found in

- [1] O. V. Bokotey, I. P. Studenyak, I. I. Nebola, and Y. V. Minets, *Theoretical study of structural features and optical properties of the $Hg_3S_2Cl_2$ polymorphs* **660**, 193–196 (2016), doi:10.1016/j.jallcom.2015.11.086.