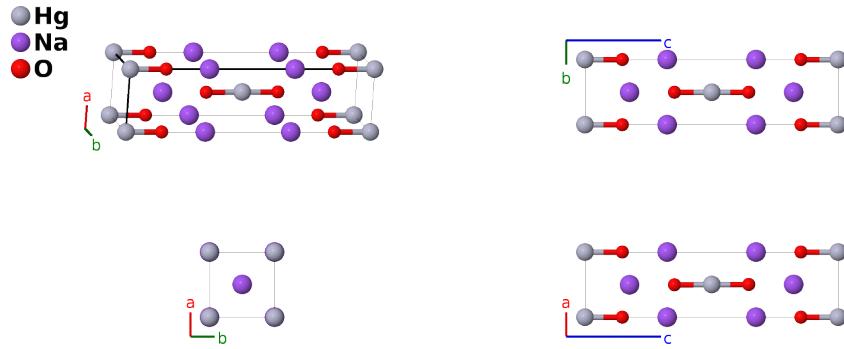


# Na<sub>2</sub>HgO<sub>2</sub> Structure: AB2C2\_tI10\_139\_a\_e-e-001

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<https://aflow.org/p/6LGV>

[https://aflow.org/p/AB2C2\\_tI10\\_139\\_a\\_e-e-001](https://aflow.org/p/AB2C2_tI10_139_a_e-e-001)



Prototype	HgNa <sub>2</sub> O <sub>2</sub>
AFLOW prototype label	AB2C2_tI10_139_a_e-e-001
ICSD	25511
Pearson symbol	tI10
Space group number	139
Space group symbol	<i>I</i> 4/ <i>mmm</i>
AFLOW prototype command	aflow --proto=AB2C2_tI10_139_a_e-e-001 --params= <i>a</i> , <i>c/a</i> , <i>z</i> <sub>2</sub> , <i>z</i> <sub>3</sub>

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**Other compounds with this structure**  
Cs<sub>2</sub>HgO<sub>2</sub>, Li<sub>2</sub>HgO<sub>2</sub>, Rb<sub>2</sub>HgO<sub>2</sub>, U<sub>2</sub>IrC<sub>2</sub>, U<sub>2</sub>RuC<sub>2</sub>

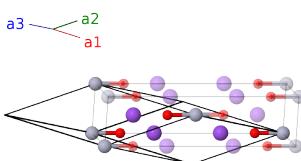
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- (Hoppe, 1964) give the space group as *I*422 #97, but the Wyckoff positions they use are consistent with *I*4/*mmm* #139, so we use the higher symmetry space group.

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**Body-centered Tetragonal primitive vectors**

$$\begin{aligned}\mathbf{a}_1 &= -\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} - \frac{1}{2}c\hat{\mathbf{z}}\end{aligned}$$




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**Basis vectors**

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1 =$	0	=	0	(2a)	Hg I
$\mathbf{B}_2 =$	$z_2 \mathbf{a}_1 + z_2 \mathbf{a}_2$	=	$cz_2 \hat{\mathbf{z}}$	(4e)	Na I
$\mathbf{B}_3 =$	$-z_2 \mathbf{a}_1 - z_2 \mathbf{a}_2$	=	$-cz_2 \hat{\mathbf{z}}$	(4e)	Na I
$\mathbf{B}_4 =$	$z_3 \mathbf{a}_1 + z_3 \mathbf{a}_2$	=	$cz_3 \hat{\mathbf{z}}$	(4e)	O I
$\mathbf{B}_5 =$	$-z_3 \mathbf{a}_1 - z_3 \mathbf{a}_2$	=	$-cz_3 \hat{\mathbf{z}}$	(4e)	O I

## References

- [1] R. Hoppe and H.-J. Rohrborn, *Oxomercurate(II) der Alkalimetalle,  $M_2HgO_2$* , Zeitschrift für anorganische und allgemeine Chemie **329**, 110–122 (1964), doi:10.1002/zaac.19643290115.