

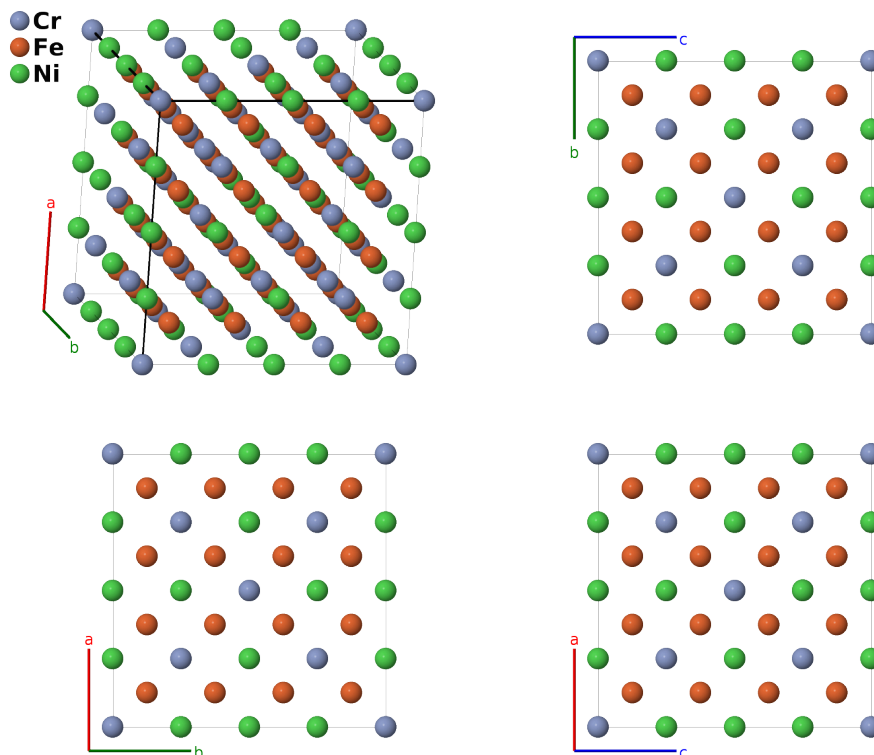
Model of Ferrite Structure (cF128): A9B16C7_cF128_225_acd_2f_be-001

This structure originally had the label A9B16C7_cF128_225_acd_2f_be. Calls to that address will be redirected here.

Cite this page as: M. J. Mehl, D. Hicks, C. Toher, O. Levy, R. M. Hanson, G. Hart, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 1*, Comput. Mater. Sci. **136**, S1-828 (2017). doi: 10.1016/j.commatsci.2017.01.017

<https://aflow.org/p/FENM>

https://aflow.org/p/A9B16C7_cF128_225_acd_2f_be-001



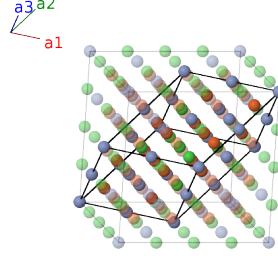
Prototype	$\text{Cr}_9\text{Fe}_{16}\text{Ni}_7$
AFLOW prototype label	A9B16C7_cF128_225_acd_2f_be-001
ICSD	none
Pearson symbol	cF128
Space group number	225
Space group symbol	$Fm\bar{3}m$
AFLOW prototype command	<code>aflow --proto=A9B16C7_cF128_225_acd_2f_be-001 --params=a, x5, x6, x7</code>

- Ferritic steels are alloys of iron and other metals with an averaged body-centered cubic structure. This model represents one approximation for a ferritic steel. It is not meant to represent a real steel, and the selection of atom types for each Wyckoff position is arbitrary.

- If we set $x_5 = 1/4$, $x_6 = 1/8$, and $x_7 = 3/8$ and replace the nickel atoms by chromium, this structure reverts to CsCl ($B2$) with $a_{B2} = 1/4a$.
- If we replace both the nickel and chromium atoms by iron the structure becomes a body-centered cubic lattice ($A2$) again with $a_{A2} = 1/4a$.

Face-centered Cubic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{y} + \frac{1}{2}a\hat{z} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{x} + \frac{1}{2}a\hat{z} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{x} + \frac{1}{2}a\hat{y}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	$= 0$	$=$	0	(4a)	Cr I
\mathbf{B}_2	$= \frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{x} + \frac{1}{2}a\hat{y} + \frac{1}{2}a\hat{z}$	(4b)	Ni I
\mathbf{B}_3	$= \frac{1}{4}\mathbf{a}_1 + \frac{1}{4}\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{x} + \frac{1}{4}a\hat{y} + \frac{1}{4}a\hat{z}$	(8c)	Cr II
\mathbf{B}_4	$= \frac{3}{4}\mathbf{a}_1 + \frac{3}{4}\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	$=$	$\frac{3}{4}a\hat{x} + \frac{3}{4}a\hat{y} + \frac{3}{4}a\hat{z}$	(8c)	Cr II
\mathbf{B}_5	$= \frac{1}{2}\mathbf{a}_1$	$=$	$\frac{1}{4}a\hat{y} + \frac{1}{4}a\hat{z}$	(24d)	Cr III
\mathbf{B}_6	$= \frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{2}a\hat{x} + \frac{1}{4}a\hat{y} + \frac{1}{4}a\hat{z}$	(24d)	Cr III
\mathbf{B}_7	$= \frac{1}{2}\mathbf{a}_2$	$=$	$\frac{1}{4}a\hat{x} + \frac{1}{4}a\hat{z}$	(24d)	Cr III
\mathbf{B}_8	$= \frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{x} + \frac{1}{2}a\hat{y} + \frac{1}{4}a\hat{z}$	(24d)	Cr III
\mathbf{B}_9	$= \frac{1}{2}\mathbf{a}_3$	$=$	$\frac{1}{4}a\hat{x} + \frac{1}{4}a\hat{y}$	(24d)	Cr III
\mathbf{B}_{10}	$= \frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2$	$=$	$\frac{1}{4}a\hat{x} + \frac{1}{4}a\hat{y} + \frac{1}{2}a\hat{z}$	(24d)	Cr III
\mathbf{B}_{11}	$= -x_5\mathbf{a}_1 + x_5\mathbf{a}_2 + x_5\mathbf{a}_3$	$=$	$ax_5\hat{x}$	(24e)	Ni II
\mathbf{B}_{12}	$= x_5\mathbf{a}_1 - x_5\mathbf{a}_2 - x_5\mathbf{a}_3$	$=$	$-ax_5\hat{x}$	(24e)	Ni II
\mathbf{B}_{13}	$= x_5\mathbf{a}_1 - x_5\mathbf{a}_2 + x_5\mathbf{a}_3$	$=$	$ax_5\hat{y}$	(24e)	Ni II
\mathbf{B}_{14}	$= -x_5\mathbf{a}_1 + x_5\mathbf{a}_2 - x_5\mathbf{a}_3$	$=$	$-ax_5\hat{y}$	(24e)	Ni II
\mathbf{B}_{15}	$= x_5\mathbf{a}_1 + x_5\mathbf{a}_2 - x_5\mathbf{a}_3$	$=$	$ax_5\hat{z}$	(24e)	Ni II
\mathbf{B}_{16}	$= -x_5\mathbf{a}_1 - x_5\mathbf{a}_2 + x_5\mathbf{a}_3$	$=$	$-ax_5\hat{z}$	(24e)	Ni II
\mathbf{B}_{17}	$= x_6\mathbf{a}_1 + x_6\mathbf{a}_2 + x_6\mathbf{a}_3$	$=$	$ax_6\hat{x} + ax_6\hat{y} + ax_6\hat{z}$	(32f)	Fe I
\mathbf{B}_{18}	$= x_6\mathbf{a}_1 + x_6\mathbf{a}_2 - 3x_6\mathbf{a}_3$	$=$	$-ax_6\hat{x} - ax_6\hat{y} + ax_6\hat{z}$	(32f)	Fe I
\mathbf{B}_{19}	$= x_6\mathbf{a}_1 - 3x_6\mathbf{a}_2 + x_6\mathbf{a}_3$	$=$	$-ax_6\hat{x} + ax_6\hat{y} - ax_6\hat{z}$	(32f)	Fe I
\mathbf{B}_{20}	$= -3x_6\mathbf{a}_1 + x_6\mathbf{a}_2 + x_6\mathbf{a}_3$	$=$	$ax_6\hat{x} - ax_6\hat{y} - ax_6\hat{z}$	(32f)	Fe I
\mathbf{B}_{21}	$= -x_6\mathbf{a}_1 - x_6\mathbf{a}_2 + 3x_6\mathbf{a}_3$	$=$	$ax_6\hat{x} + ax_6\hat{y} - ax_6\hat{z}$	(32f)	Fe I
\mathbf{B}_{22}	$= -x_6\mathbf{a}_1 - x_6\mathbf{a}_2 - x_6\mathbf{a}_3$	$=$	$-ax_6\hat{x} - ax_6\hat{y} - ax_6\hat{z}$	(32f)	Fe I
\mathbf{B}_{23}	$= -x_6\mathbf{a}_1 + 3x_6\mathbf{a}_2 - x_6\mathbf{a}_3$	$=$	$ax_6\hat{x} - ax_6\hat{y} + ax_6\hat{z}$	(32f)	Fe I
\mathbf{B}_{24}	$= 3x_6\mathbf{a}_1 - x_6\mathbf{a}_2 - x_6\mathbf{a}_3$	$=$	$-ax_6\hat{x} + ax_6\hat{y} + ax_6\hat{z}$	(32f)	Fe I
\mathbf{B}_{25}	$= x_7\mathbf{a}_1 + x_7\mathbf{a}_2 + x_7\mathbf{a}_3$	$=$	$ax_7\hat{x} + ax_7\hat{y} + ax_7\hat{z}$	(32f)	Fe II

$$\begin{aligned}
\mathbf{B}_{26} &= x_7 \mathbf{a}_1 + x_7 \mathbf{a}_2 - 3x_7 \mathbf{a}_3 &= & -ax_7 \hat{\mathbf{x}} - ax_7 \hat{\mathbf{y}} + ax_7 \hat{\mathbf{z}} & (32f) & \text{Fe II} \\
\mathbf{B}_{27} &= x_7 \mathbf{a}_1 - 3x_7 \mathbf{a}_2 + x_7 \mathbf{a}_3 &= & -ax_7 \hat{\mathbf{x}} + ax_7 \hat{\mathbf{y}} - ax_7 \hat{\mathbf{z}} & (32f) & \text{Fe II} \\
\mathbf{B}_{28} &= -3x_7 \mathbf{a}_1 + x_7 \mathbf{a}_2 + x_7 \mathbf{a}_3 &= & ax_7 \hat{\mathbf{x}} - ax_7 \hat{\mathbf{y}} - ax_7 \hat{\mathbf{z}} & (32f) & \text{Fe II} \\
\mathbf{B}_{29} &= -x_7 \mathbf{a}_1 - x_7 \mathbf{a}_2 + 3x_7 \mathbf{a}_3 &= & ax_7 \hat{\mathbf{x}} + ax_7 \hat{\mathbf{y}} - ax_7 \hat{\mathbf{z}} & (32f) & \text{Fe II} \\
\mathbf{B}_{30} &= -x_7 \mathbf{a}_1 - x_7 \mathbf{a}_2 - x_7 \mathbf{a}_3 &= & -ax_7 \hat{\mathbf{x}} - ax_7 \hat{\mathbf{y}} - ax_7 \hat{\mathbf{z}} & (32f) & \text{Fe II} \\
\mathbf{B}_{31} &= -x_7 \mathbf{a}_1 + 3x_7 \mathbf{a}_2 - x_7 \mathbf{a}_3 &= & ax_7 \hat{\mathbf{x}} - ax_7 \hat{\mathbf{y}} + ax_7 \hat{\mathbf{z}} & (32f) & \text{Fe II} \\
\mathbf{B}_{32} &= 3x_7 \mathbf{a}_1 - x_7 \mathbf{a}_2 - x_7 \mathbf{a}_3 &= & -ax_7 \hat{\mathbf{x}} + ax_7 \hat{\mathbf{y}} + ax_7 \hat{\mathbf{z}} & (32f) & \text{Fe II}
\end{aligned}$$

References

- [1] M. J. Mehl, D. Hicks, C. Toher, O. Levy, R. M. Hanson, G. Hart, and S. Curtarolo, *The AFLOW library of crystallographic prototypes: part 1*, *Comput. Mater. Sci.* **136**, S1–S828 (2017), doi:10.1016/j.commatsci.2017.01.017.