

Cs₇O Structure:

A7B_hP24_187_ah2j2kn_j-001

This structure originally had the label A7B_hP24_187_ai2j2kn_j. Calls to that address will be redirected here.

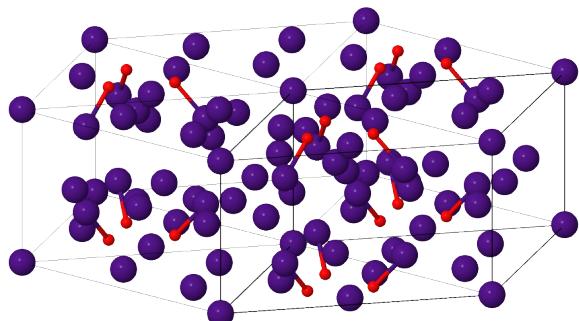
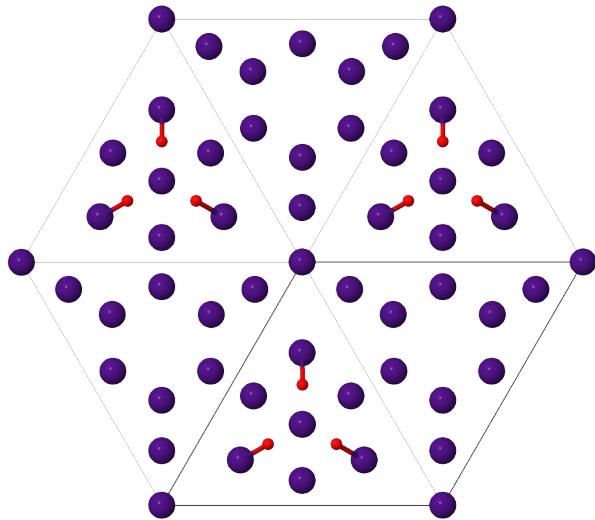
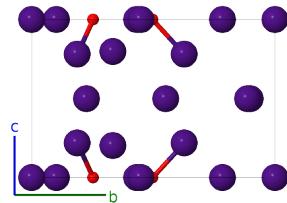
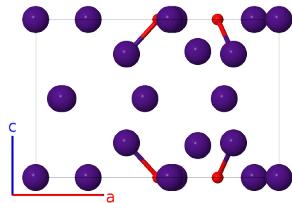
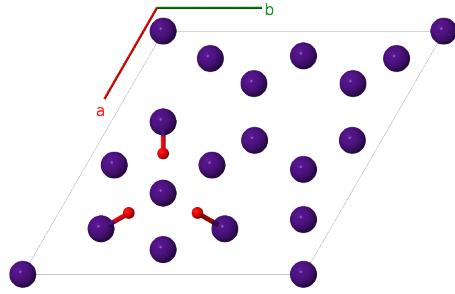
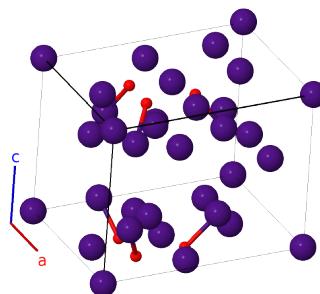
Cite this page as: D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.

<https://aflow.org/p/BZC9>

https://aflow.org/p/A7B_hP24_187_ah2j2kn_j-001

● Cs

● O



Prototype

Cs₇O

AFLOW prototype label

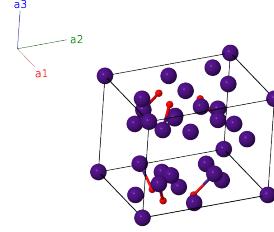
A7B_hP24_187_ah2j2kn_j-001

ICSD	111
Pearson symbol	hP24
Space group number	187
Space group symbol	$P\bar{6}m2$
AFLW prototype command	<code>aflow --proto=A7B_hP24_187_ah2j2kn_j-001 --params=a, c/a, z2, x3, x4, x5, x6, x7, x8, z8</code>

- This structure is composed of Cs_{11}O_3 molecules, similar to the building blocks of the Cs_{11}O_3 structure, interlaced with cesium atoms which have approximately the same spacing as in bcc-Cs.
- Lattice constant data was given at -150°C , while the atomic positions were given at -175°C .

Hexagonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_3 &= c\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	= 0	=	0	(1a)	Cs I
\mathbf{B}_2	= $\frac{1}{3}\mathbf{a}_1 + \frac{2}{3}\mathbf{a}_2 + z_2\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + cz_2\hat{\mathbf{z}}$	(2h)	Cs II
\mathbf{B}_3	= $\frac{1}{3}\mathbf{a}_1 + \frac{2}{3}\mathbf{a}_2 - z_2\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} - cz_2\hat{\mathbf{z}}$	(2h)	Cs II
\mathbf{B}_4	= $x_3\mathbf{a}_1 - x_3\mathbf{a}_2$	=	$-\sqrt{3}ax_3\hat{\mathbf{y}}$	(3j)	Cs III
\mathbf{B}_5	= $x_3\mathbf{a}_1 + 2x_3\mathbf{a}_2$	=	$\frac{3}{2}ax_3\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3\hat{\mathbf{y}}$	(3j)	Cs III
\mathbf{B}_6	= $-2x_3\mathbf{a}_1 - x_3\mathbf{a}_2$	=	$-\frac{3}{2}ax_3\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3\hat{\mathbf{y}}$	(3j)	Cs III
\mathbf{B}_7	= $x_4\mathbf{a}_1 - x_4\mathbf{a}_2$	=	$-\sqrt{3}ax_4\hat{\mathbf{y}}$	(3j)	Cs IV
\mathbf{B}_8	= $x_4\mathbf{a}_1 + 2x_4\mathbf{a}_2$	=	$\frac{3}{2}ax_4\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}}$	(3j)	Cs IV
\mathbf{B}_9	= $-2x_4\mathbf{a}_1 - x_4\mathbf{a}_2$	=	$-\frac{3}{2}ax_4\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}}$	(3j)	Cs IV
\mathbf{B}_{10}	= $x_5\mathbf{a}_1 - x_5\mathbf{a}_2$	=	$-\sqrt{3}ax_5\hat{\mathbf{y}}$	(3j)	O I
\mathbf{B}_{11}	= $x_5\mathbf{a}_1 + 2x_5\mathbf{a}_2$	=	$\frac{3}{2}ax_5\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_5\hat{\mathbf{y}}$	(3j)	O I
\mathbf{B}_{12}	= $-2x_5\mathbf{a}_1 - x_5\mathbf{a}_2$	=	$-\frac{3}{2}ax_5\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_5\hat{\mathbf{y}}$	(3j)	O I
\mathbf{B}_{13}	= $x_6\mathbf{a}_1 - x_6\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$-\sqrt{3}ax_6\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	Cs V
\mathbf{B}_{14}	= $x_6\mathbf{a}_1 + 2x_6\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{3}{2}ax_6\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_6\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	Cs V
\mathbf{B}_{15}	= $-2x_6\mathbf{a}_1 - x_6\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$-\frac{3}{2}ax_6\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_6\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	Cs V
\mathbf{B}_{16}	= $x_7\mathbf{a}_1 - x_7\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$-\sqrt{3}ax_7\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	Cs VI
\mathbf{B}_{17}	= $x_7\mathbf{a}_1 + 2x_7\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{3}{2}ax_7\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_7\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	Cs VI
\mathbf{B}_{18}	= $-2x_7\mathbf{a}_1 - x_7\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$-\frac{3}{2}ax_7\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_7\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	Cs VI
\mathbf{B}_{19}	= $x_8\mathbf{a}_1 - x_8\mathbf{a}_2 + z_8\mathbf{a}_3$	=	$-\sqrt{3}ax_8\hat{\mathbf{y}} + cz_8\hat{\mathbf{z}}$	(6n)	Cs VII

$$\begin{aligned}
\mathbf{B}_{20} &= x_8 \mathbf{a}_1 + 2x_8 \mathbf{a}_2 + z_8 \mathbf{a}_3 & = & \frac{3}{2}ax_8 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_8 \hat{\mathbf{y}} + cz_8 \hat{\mathbf{z}} & (6n) & \text{Cs VII} \\
\mathbf{B}_{21} &= -2x_8 \mathbf{a}_1 - x_8 \mathbf{a}_2 + z_8 \mathbf{a}_3 & = & -\frac{3}{2}ax_8 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_8 \hat{\mathbf{y}} + cz_8 \hat{\mathbf{z}} & (6n) & \text{Cs VII} \\
\mathbf{B}_{22} &= x_8 \mathbf{a}_1 - x_8 \mathbf{a}_2 - z_8 \mathbf{a}_3 & = & -\sqrt{3}ax_8 \hat{\mathbf{y}} - cz_8 \hat{\mathbf{z}} & (6n) & \text{Cs VII} \\
\mathbf{B}_{23} &= x_8 \mathbf{a}_1 + 2x_8 \mathbf{a}_2 - z_8 \mathbf{a}_3 & = & \frac{3}{2}ax_8 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_8 \hat{\mathbf{y}} - cz_8 \hat{\mathbf{z}} & (6n) & \text{Cs VII} \\
\mathbf{B}_{24} &= -2x_8 \mathbf{a}_1 - x_8 \mathbf{a}_2 - z_8 \mathbf{a}_3 & = & -\frac{3}{2}ax_8 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_8 \hat{\mathbf{y}} - cz_8 \hat{\mathbf{z}} & (6n) & \text{Cs VII}
\end{aligned}$$

References

- [1] A. Simon, *Über Alkalimetall-Suboxide. VII. Das metallreichste Cäsiumoxid—Cs₇O*, Z. Anorganische und Allgemeine Chemie **422**, 208–218 (1976), doi:10.1002/zaac.19764220303.

Found in

- [1] T. B. Massalski, H. Okamoto, P. R. Subramanian, and L. Kacprzak, eds., *Binary Alloy Phase Diagrams*, vol. 2 (ASM International, Materials Park, Ohio, USA), 2nd edn. Cd-Ce to Hf-Rb.