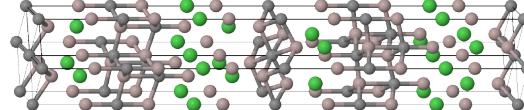
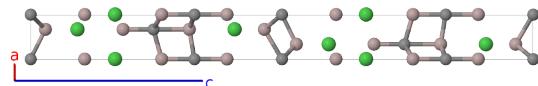
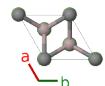
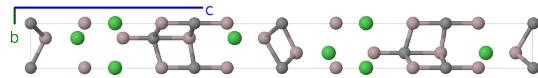
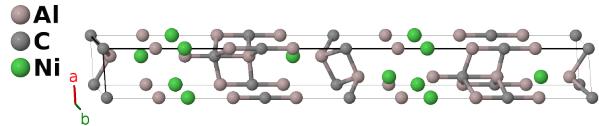


Al₇C₃N₃ Structure: A7B3C3_hP26_186_3a4b_2ab_a2b-001

Cite this page as: H. Eckert, S. Divilov, A. Zettel, M. J. Mehl, D. Hicks, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*. In preparation.

<https://aflow.org/p/JALT>

https://aflow.org/p/A7B3C3.hP26_186_3a4b_2ab_a2b-001



Prototype Al₇C₃N₃

AFLOW prototype label A7B3C3_hP26_186_3a4b_2ab_a2b-001

ICSD 14400

Pearson symbol hP26

Space group number 186

Space group symbol P6₃mc

AFLOW prototype command `aflow --proto=A7B3C3_hP26_186_3a4b_2ab_a2b-001 --params=a, c/a, z1, z2, z3, z4, z5, z6, z7, z8, z9, z10, z11, z12, z13`

Other compounds with this structure

(Al_{7-x}Si_x)(O_yC_zN_{6-y-z})

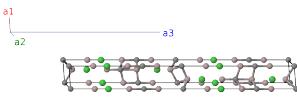
- (Jeffrey, 1963) place the structure in space group P6₃mc #186, but give the structure in a pseudo-hexagonal cell in space group Cmc2₁ #36. With the help of FINDSYM we converted this to the hexagonal structure.
- (Jeffrey, 1966) uses the P6₃mc space group exclusively, and we list the ICSD entry associated with that paper.

Hexagonal primitive vectors

$$\mathbf{a}_1 = \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a\hat{\mathbf{y}}$$

$$\mathbf{a}_2 = \frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a\hat{\mathbf{y}}$$

$$\mathbf{a}_3 = c\hat{\mathbf{z}}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
B₁ =	$z_1 \mathbf{a}_3$	=	$cz_1 \hat{\mathbf{z}}$	(2a)	Al I
B₂ =	$(z_1 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_1 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Al I
B₃ =	$z_2 \mathbf{a}_3$	=	$cz_2 \hat{\mathbf{z}}$	(2a)	Al II
B₄ =	$(z_2 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_2 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Al II
B₅ =	$z_3 \mathbf{a}_3$	=	$cz_3 \hat{\mathbf{z}}$	(2a)	Al III
B₆ =	$(z_3 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_3 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Al III
B₇ =	$z_4 \mathbf{a}_3$	=	$cz_4 \hat{\mathbf{z}}$	(2a)	C I
B₈ =	$(z_4 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_4 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	C I
B₉ =	$z_5 \mathbf{a}_3$	=	$cz_5 \hat{\mathbf{z}}$	(2a)	C II
B₁₀ =	$(z_5 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_5 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	C II
B₁₁ =	$z_6 \mathbf{a}_3$	=	$cz_6 \hat{\mathbf{z}}$	(2a)	Ni I
B₁₂ =	$(z_6 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_6 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Ni I
B₁₃ =	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_7 \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_7 \hat{\mathbf{z}}$	(2b)	Al IV
B₁₄ =	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_7 + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_7 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al IV
B₁₅ =	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_8 \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_8 \hat{\mathbf{z}}$	(2b)	Al V
B₁₆ =	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_8 + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_8 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al V
B₁₇ =	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_9 \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_9 \hat{\mathbf{z}}$	(2b)	Al VI
B₁₈ =	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_9 + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_9 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al VI
B₁₉ =	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_{10} \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_{10} \hat{\mathbf{z}}$	(2b)	Al VII
B₂₀ =	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_{10} + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_{10} + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al VII
B₂₁ =	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_{11} \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_{11} \hat{\mathbf{z}}$	(2b)	C III
B₂₂ =	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_{11} + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_{11} + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	C III
B₂₃ =	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_{12} \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_{12} \hat{\mathbf{z}}$	(2b)	Ni II
B₂₄ =	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_{12} + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_{12} + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Ni II
B₂₅ =	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_{13} \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_{13} \hat{\mathbf{z}}$	(2b)	Ni III
B₂₆ =	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_{13} + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_{13} + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Ni III

References

- [1] G. A. Jeffrey and V. Y. Wu, *The structures of the aluminum carbonitrides*, Acta Cryst. **16**, 559–566 (1963), doi:10.1107/S0365110X63001468.
- [2] G. A. Jeffrey and V. Y. Wu, *The structure of the aluminum carbonitrides. II*, Acta Cryst. **20**, 538–547 (1966), doi:10.1107/S0365110X66001208.

Found in

- [1] D. Urushihara, M. Kaga, T. Asaka, H. Nakano, and K. Fukuda, *Synthesis and structural characterization of Al₇C₃N₃-homeotypic aluminum silicon oxycarbonitride, (Al_{7-x}Si_x)(O_yC_zN_{6-y-z})* ($x \sim 1.2$, $y \sim 1.0$ and $z \sim 3.5$), J. Solid State Chem. **184**, 2278–2284 (2011), doi:10.1016/j.jssc.2011.06.030.