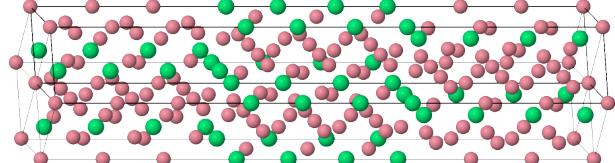
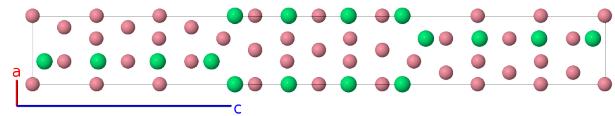
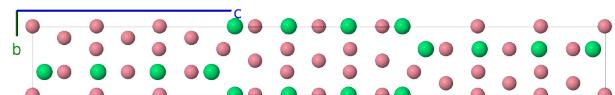
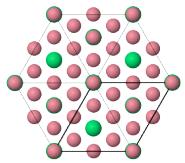
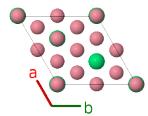
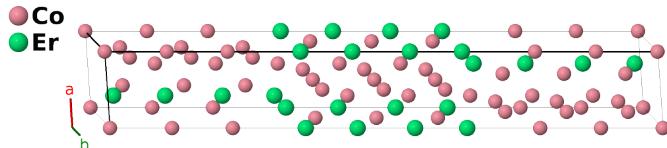


Er₂Co₇ Structure: A7B2_hR18_166_a2cdh_2c-001

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<https://aflow.org/p/VGKE>

https://aflow.org/p/A7B2_hR18_166_a2cdh_2c-001



Prototype Co₇Er₂

AFLOW prototype label A7B2_hR18_166_a2cdh_2c-001

ICSD 102366

Pearson symbol hR18

Space group number 166

Space group symbol $R\bar{3}m$

AFLOW prototype command `aflow --proto=A7B2_hR18_166_a2cdh_2c-001
--params=a, c/a, x2, x3, x4, x5, x7, z7`

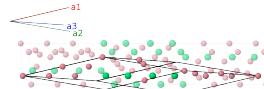
Other compounds with this structure

Dy₂Co₇, Er₂Ni₇, Gd₂Co₇, Gd₂Ni₇, Ho₂Co₇, La₂Ni₇, Lu₂Co₇, Tb₂Co₇, Th₂Fe₇, Th₂Ni₇, Tm₂Co₇, Y₂Co₇, Y₂Ni₇

- The ICSD lists Gd₂Co₇ as the prototype, but (Ostertag, 1967) only gives atomic coordinates for Er₂Co₇, so we will use that instead.
- Hexagonal settings of this structure can be obtained with the option `--hex`.

Rhombohedral primitive vectors

$$\begin{aligned}
\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + \frac{1}{3}c\hat{\mathbf{z}} \\
\mathbf{a}_2 &= \frac{1}{\sqrt{3}}a\hat{\mathbf{y}} + \frac{1}{3}c\hat{\mathbf{z}} \\
\mathbf{a}_3 &= -\frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + \frac{1}{3}c\hat{\mathbf{z}}
\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	= 0	=	0	(1a)	Co I
\mathbf{B}_2	= $x_2 \mathbf{a}_1 + x_2 \mathbf{a}_2 + x_2 \mathbf{a}_3$	=	$cx_2 \hat{\mathbf{z}}$	(2c)	Co II
\mathbf{B}_3	= $-x_2 \mathbf{a}_1 - x_2 \mathbf{a}_2 - x_2 \mathbf{a}_3$	=	$-cx_2 \hat{\mathbf{z}}$	(2c)	Co II
\mathbf{B}_4	= $x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	=	$cx_3 \hat{\mathbf{z}}$	(2c)	Co III
\mathbf{B}_5	= $-x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	=	$-cx_3 \hat{\mathbf{z}}$	(2c)	Co III
\mathbf{B}_6	= $x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + x_4 \mathbf{a}_3$	=	$cx_4 \hat{\mathbf{z}}$	(2c)	Er I
\mathbf{B}_7	= $-x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 - x_4 \mathbf{a}_3$	=	$-cx_4 \hat{\mathbf{z}}$	(2c)	Er I
\mathbf{B}_8	= $x_5 \mathbf{a}_1 + x_5 \mathbf{a}_2 + x_5 \mathbf{a}_3$	=	$cx_5 \hat{\mathbf{z}}$	(2c)	Er II
\mathbf{B}_9	= $-x_5 \mathbf{a}_1 - x_5 \mathbf{a}_2 - x_5 \mathbf{a}_3$	=	$-cx_5 \hat{\mathbf{z}}$	(2c)	Er II
\mathbf{B}_{10}	= $\frac{1}{2} \mathbf{a}_1$	=	$\frac{1}{4}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{12}a\hat{\mathbf{y}} + \frac{1}{6}c\hat{\mathbf{z}}$	(3d)	Co IV
\mathbf{B}_{11}	= $\frac{1}{2} \mathbf{a}_2$	=	$\frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + \frac{1}{6}c\hat{\mathbf{z}}$	(3d)	Co IV
\mathbf{B}_{12}	= $\frac{1}{2} \mathbf{a}_3$	=	$-\frac{1}{4}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{12}a\hat{\mathbf{y}} + \frac{1}{6}c\hat{\mathbf{z}}$	(3d)	Co IV
\mathbf{B}_{13}	= $x_7 \mathbf{a}_1 + x_7 \mathbf{a}_2 + z_7 \mathbf{a}_3$	=	$\frac{1}{2}a(x_7 - z_7)\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a(x_7 - z_7)\hat{\mathbf{y}} + \frac{1}{3}c(2x_7 + z_7)\hat{\mathbf{z}}$	(6h)	Co V
\mathbf{B}_{14}	= $z_7 \mathbf{a}_1 + x_7 \mathbf{a}_2 + x_7 \mathbf{a}_3$	=	$-\frac{1}{2}a(x_7 - z_7)\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a(x_7 - z_7)\hat{\mathbf{y}} + \frac{1}{3}c(2x_7 + z_7)\hat{\mathbf{z}}$	(6h)	Co V
\mathbf{B}_{15}	= $x_7 \mathbf{a}_1 + z_7 \mathbf{a}_2 + x_7 \mathbf{a}_3$	=	$-\frac{1}{\sqrt{3}}a(x_7 - z_7)\hat{\mathbf{y}} + \frac{1}{3}c(2x_7 + z_7)\hat{\mathbf{z}}$	(6h)	Co V
\mathbf{B}_{16}	= $-z_7 \mathbf{a}_1 - x_7 \mathbf{a}_2 - x_7 \mathbf{a}_3$	=	$\frac{1}{2}a(x_7 - z_7)\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a(x_7 - z_7)\hat{\mathbf{y}} - \frac{1}{3}c(2x_7 + z_7)\hat{\mathbf{z}}$	(6h)	Co V
\mathbf{B}_{17}	= $-x_7 \mathbf{a}_1 - x_7 \mathbf{a}_2 - z_7 \mathbf{a}_3$	=	$-\frac{1}{2}a(x_7 - z_7)\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a(x_7 - z_7)\hat{\mathbf{y}} - \frac{1}{3}c(2x_7 + z_7)\hat{\mathbf{z}}$	(6h)	Co V
\mathbf{B}_{18}	= $-x_7 \mathbf{a}_1 - z_7 \mathbf{a}_2 - x_7 \mathbf{a}_3$	=	$\frac{1}{\sqrt{3}}a(x_7 - z_7)\hat{\mathbf{y}} - \frac{1}{3}c(2x_7 + z_7)\hat{\mathbf{z}}$	(6h)	Co V

References

- [1] W. Ostertag, *The crystal structure of Er_2Co_7 and other rare earth-cobalt compounds R_2Co_7 ($R = Gd, Tb, Dy, Ho, Tm, Lu, Y$)*, J. Less-Common Met. **13**, 385–390 (1967), doi:10.1016/0022-5088(67)90032-X.