

Rb₃Mn₂Cl₇ Structure:

A7B2C3_tI24_139_aeg_e_be-001

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<https://aflow.org/prototype-encyclopedia/7JA2>

https://aflow.org/prototype-encyclopedia/A7B2C3_tI24_139_aeg_e_be-001

Prototype	Cl ₇ Mn ₂ Rb ₃
AFLOW prototype label	A7B2C3_tI24_139_aeg_e_be-001
ICSD	9491
CCDC	1594478
Pearson symbol	tI24
Space group number	139
Space group symbol	<i>I</i> 4/ <i>mmm</i>
AFLOW prototype command	<code>aflow --proto=A7B2C3_tI24_139_aeg_e_be-001 --params=a, c/a, z₃, z₄, z₅, z₆</code>

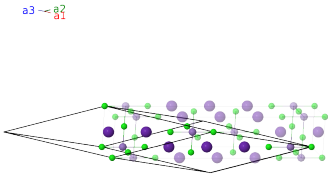
Other compounds with this structure

Cs₃Ca₂Cl₇, Cs₃Mg₂Cl₇, K₃Mn₂Cl₇, Rb₃Mn₂Br₇

- Quaternary compounds with this structure are listed under the Sr₃Sc₂O₅Cl₂ prototype. The ICSD list some, but not all, of those structures under the Rb₃Mn₂Cl₇ structure type.

Body-centered Tetragonal primitive vectors

$$\begin{aligned} \mathbf{a}_1 &= -\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} - \frac{1}{2}c\hat{\mathbf{z}} \end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	=	0	=	0	(2a) Cl I

$\mathbf{B}_2$	$=$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{2} c \hat{\mathbf{z}}$	(2b)	Rb I
$\mathbf{B}_3$	$=$	$z_3 \mathbf{a}_1 + z_3 \mathbf{a}_2$	$=$	$cz_3 \hat{\mathbf{z}}$	(4e)	Cl II
$\mathbf{B}_4$	$=$	$-z_3 \mathbf{a}_1 - z_3 \mathbf{a}_2$	$=$	$-cz_3 \hat{\mathbf{z}}$	(4e)	Cl II
$\mathbf{B}_5$	$=$	$z_4 \mathbf{a}_1 + z_4 \mathbf{a}_2$	$=$	$cz_4 \hat{\mathbf{z}}$	(4e)	Mn I
$\mathbf{B}_6$	$=$	$-z_4 \mathbf{a}_1 - z_4 \mathbf{a}_2$	$=$	$-cz_4 \hat{\mathbf{z}}$	(4e)	Mn I
$\mathbf{B}_7$	$=$	$z_5 \mathbf{a}_1 + z_5 \mathbf{a}_2$	$=$	$cz_5 \hat{\mathbf{z}}$	(4e)	Rb II
$\mathbf{B}_8$	$=$	$-z_5 \mathbf{a}_1 - z_5 \mathbf{a}_2$	$=$	$-cz_5 \hat{\mathbf{z}}$	(4e)	Rb II
$\mathbf{B}_9$	$=$	$\left(z_6 + \frac{1}{2}\right) \mathbf{a}_1 + z_6 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(8g)	Cl III
$\mathbf{B}_{10}$	$=$	$z_6 \mathbf{a}_1 + \left(z_6 + \frac{1}{2}\right) \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} + cz_6 \hat{\mathbf{z}}$	(8g)	Cl III
$\mathbf{B}_{11}$	$=$	$-\left(z_6 - \frac{1}{2}\right) \mathbf{a}_1 - z_6 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{y}} - cz_6 \hat{\mathbf{z}}$	(8g)	Cl III
$\mathbf{B}_{12}$	$=$	$-z_6 \mathbf{a}_1 - \left(z_6 - \frac{1}{2}\right) \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - cz_6 \hat{\mathbf{z}}$	(8g)	Cl III

References

- [1] M. Amit, A. Horowitz, and J. Makovsky, *Crystal Structure of Rb_2CoCl_4 and $Rb_3Mn_2Cl_7$* , Israel J. Chem. **10**, 715–719 (1972), doi:10.1002/ijch.197200070.

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Cl
Mn
Rb

