

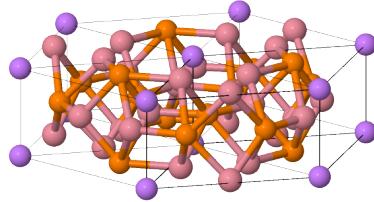
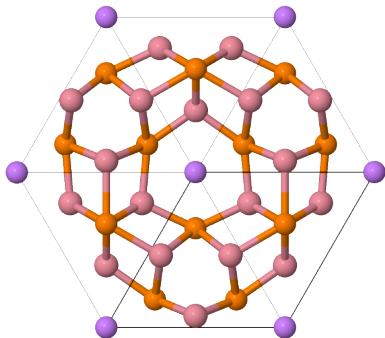
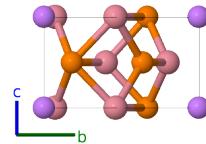
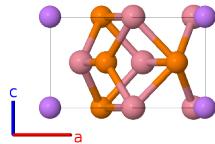
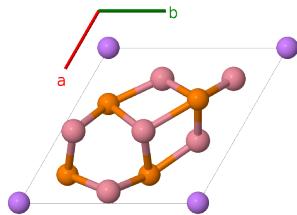
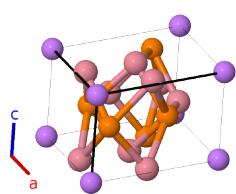
# LiCo<sub>6</sub>P<sub>4</sub> Structure: A6BC4\_hP11\_187\_jk\_a\_ck-001

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<https://aflow.org/p/NFL9>

[https://aflow.org/p/A6BC4\\_hP11\\_187\\_jk\\_a\\_ck-001](https://aflow.org/p/A6BC4_hP11_187_jk_a_ck-001)

■ Co  
■ Li  
■ P



**Prototype** Co<sub>6</sub>LiP<sub>4</sub>

**AFLOW prototype label** A6BC4\_hP11\_187\_jk\_a\_ck-001

**ICSD** 69692

**Pearson symbol** hP11

**Space group number** 187

**Space group symbol**  $P\bar{6}m2$

**AFLOW prototype command**

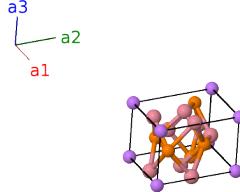
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--params=a, c/a, x3, x4, x5
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## Other compounds with this structure

CeCo<sub>6</sub>P<sub>4</sub>, CeRh<sub>6</sub>Ge<sub>4</sub>, CeRh<sub>6</sub>Ge<sub>4</sub>, DyCo<sub>6</sub>P<sub>4</sub>, DyRh<sub>6</sub>Ge<sub>4</sub>, ErCo<sub>6</sub>P<sub>4</sub>, ErRh<sub>6</sub>Ge<sub>4</sub>, EuCo<sub>6</sub>P<sub>4</sub>, EuRh<sub>6</sub>Ge<sub>4</sub>, GdCo<sub>6</sub>P<sub>4</sub>, GdRh<sub>6</sub>Ge<sub>4</sub>, HoCo<sub>6</sub>P<sub>4</sub>, HoRh<sub>6</sub>Ge<sub>4</sub>, LaCo<sub>6</sub>P<sub>4</sub>, LaRh<sub>6</sub>Ge<sub>4</sub>, LuCo<sub>6</sub>P<sub>4</sub>, LuRh<sub>6</sub>Ge<sub>4</sub>, LuRh<sub>6</sub>P<sub>4</sub>, NdCo<sub>6</sub>P<sub>4</sub>, NdRh<sub>6</sub>Ge<sub>4</sub>, PrCo<sub>6</sub>P<sub>4</sub>, PrRh<sub>6</sub>Ge<sub>4</sub>, ScRh<sub>6</sub>P<sub>4</sub>, SmCo<sub>6</sub>P<sub>4</sub>, SmRh<sub>6</sub>Ge<sub>4</sub>, TbCo<sub>6</sub>P<sub>4</sub>, TbRh<sub>6</sub>Ge<sub>4</sub>, TmCo<sub>6</sub>P<sub>4</sub>, TmRh<sub>6</sub>Ge<sub>4</sub>, YCo<sub>6</sub>P<sub>4</sub>, YRh<sub>6</sub>Ge<sub>4</sub>, YRh<sub>6</sub>P<sub>4</sub>, YbCo<sub>6</sub>P<sub>4</sub>, YbRh<sub>6</sub>Ge<sub>4</sub>

## Hexagonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_3 &= c\hat{\mathbf{z}}\end{aligned}$$



## Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	= 0	=	0	(1a)	Li I
$\mathbf{B}_2$	= $\frac{1}{3}\mathbf{a}_1 + \frac{2}{3}\mathbf{a}_2$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a\hat{\mathbf{y}}$	(1c)	P I
$\mathbf{B}_3$	= $x_3\mathbf{a}_1 - x_3\mathbf{a}_2$	=	$-\sqrt{3}ax_3\hat{\mathbf{y}}$	(3j)	Co I
$\mathbf{B}_4$	= $x_3\mathbf{a}_1 + 2x_3\mathbf{a}_2$	=	$\frac{3}{2}ax_3\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3\hat{\mathbf{y}}$	(3j)	Co I
$\mathbf{B}_5$	= $-2x_3\mathbf{a}_1 - x_3\mathbf{a}_2$	=	$-\frac{3}{2}ax_3\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_3\hat{\mathbf{y}}$	(3j)	Co I
$\mathbf{B}_6$	= $x_4\mathbf{a}_1 - x_4\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$-\sqrt{3}ax_4\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	Co II
$\mathbf{B}_7$	= $x_4\mathbf{a}_1 + 2x_4\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{3}{2}ax_4\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	Co II
$\mathbf{B}_8$	= $-2x_4\mathbf{a}_1 - x_4\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$-\frac{3}{2}ax_4\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	Co II
$\mathbf{B}_9$	= $x_5\mathbf{a}_1 - x_5\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$-\sqrt{3}ax_5\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	P II
$\mathbf{B}_{10}$	= $x_5\mathbf{a}_1 + 2x_5\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{3}{2}ax_5\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_5\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	P II
$\mathbf{B}_{11}$	= $-2x_5\mathbf{a}_1 - x_5\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$-\frac{3}{2}ax_5\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_5\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(3k)	P II

## References

- [1] R. Buschmann and H.-U. Schuster, *Darstellung und Kristallstruktur der Verbindung LiCo<sub>6</sub>P<sub>4</sub>*, Z. Naturforsch. B **46**, 699–701 (1991), doi:10.1515/znb-1991-0525.
- [1] S. F. Matar, A. Al-Alam, N. Ouaini, and R. Pöttgen, *Ab initio investigations of the electronic structures and chemical bonding in LiCo<sub>6</sub>P<sub>4</sub> and Li<sub>2</sub>Co<sub>12</sub>P<sub>7</sub>*, J. Solid State Chem. **202**, 227–233 (2013), doi:10.1016/j.jssc.2013.03.032.