

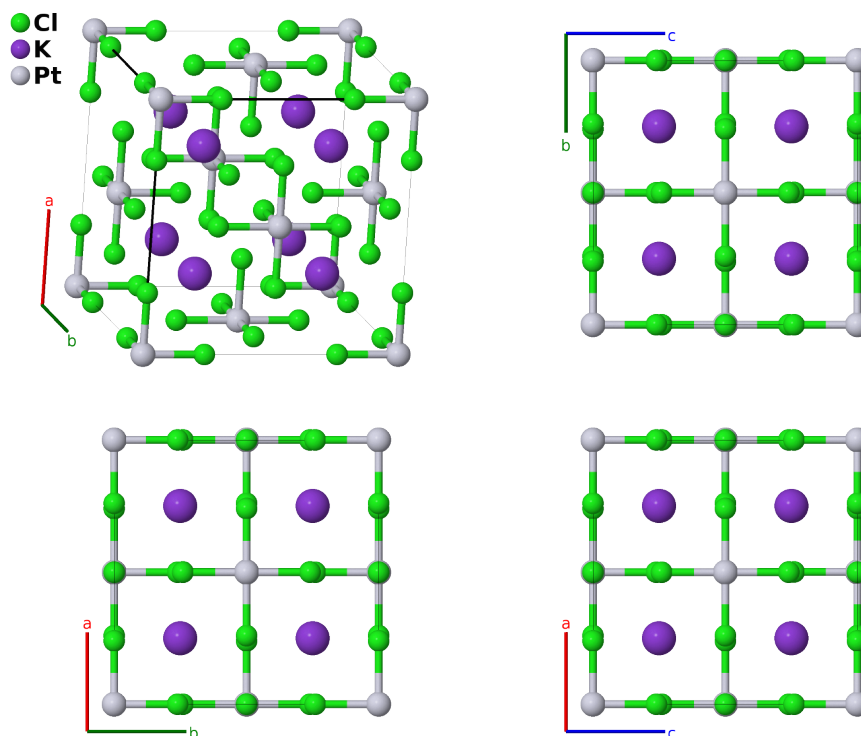
K_2PtCl_6 ($J1_1$) Structure: A6B2C_cF36_225_e_c_a-001

This structure originally had the label A6B2C_cF36_225_e_c_a. Calls to that address will be redirected here.

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<https://afLOW.org/p/L6LF>

https://afLOW.org/p/A6B2C_cF36_225_e_c_a-001



Prototype	Cl_6K_2Pt
AFLOW prototype label	A6B2C_cF36_225_e_c_a-001
<i>Strukturbericht</i> designation	$J1_1$
ICSD	31114
Pearson symbol	cF36
Space group number	225
Space group symbol	$Fm\bar{3}m$
AFLOW prototype command	<code>afLOW --proto=A6B2C_cF36_225_e_c_a-001 --params=a, x3</code>

Other compounds with this structure

Ba_2RuH_6 , Ca_2FeH_6 , Ca_2OsH_6 , Ca_2RuH_6 , Cs_2CoF_6 , Cs_2CsF_6 , Cs_2GeCl_6 , Cs_2GeF_6 , Cs_2MnCl_6 , Cs_2NbI_6 , Cs_2PbCl_6 , Cs_2PdBr_6 , Cs_2PtCl_6 , Cs_2SnBr_6 , Cs_2SnCl_6 , Cs_2SnI_6 , Cs_2TeCl_6 , Cs_2TiCl_6 , Cs_2TiCl_6 , Cs_2ZrCl_6 , Gd_2MnGa_6 , K_2MnCl_6 , K_2OsBr_2 , K_2OsCl_6 ,

K₂PtCl₆, K₂PtCl₆, K₂ReCl₆, K₂SnBr₆, K₂SnCl₆, K₂SnI₆, K₂TeBr₆, Mg₂FeH₆, Mg₂OsH₆, Mg₂RuH₆, Rb₂MnCl₆, Rb₂PbCl₆, Rb₂PdCl₆, Rb₂PtCl₆, Rb₂SeCl₆, Rb₂SnBr₆, Rb₂SnCl₆, Rb₂SnI₆, Rb₂TaBr₆, Rb₂TeCl₆, Rb₂ZrCl₆, Sr₂FeH₆, Sr₂OsH₆, Sr₂RuH₆, Tl₂SnCl₆, (NH₄)₂PbCl₆, (NH₄)₂PtCl₆, (NH₄)₂SbCl₆, (NH₄)₂SiF₆, (NH₄)₂SnCl₆, Br₂Ni(NH₃)₆, Cl₂Co(NH₃)₆, Cs₂Ni(NH₃)₆, I₂Co(NH₃)₆, I₂Ni(NH₃)₆

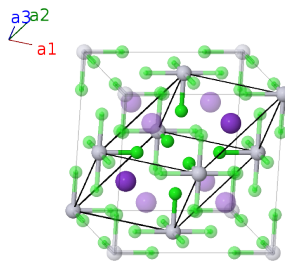
- (Ewald, 1931) originally listed this as *Strukturbericht H61* (*H6₁* in later notation). (Hermann, 1937) changed this to *I1₁*, and (Gottfried, 1937) changed it to *J1₁*.
- (Douglas, 2006), Table 6.6, provides an extensive list of compounds with this structure. Most have the formula A₂MX₆, where A is an alkali metal, M is a metal, and X is a halide. An ammonium ion (NH₄⁺) can also substitute for the alkali.

Face-centered Cubic primitive vectors

$$\mathbf{a}_1 = \frac{1}{2}a \hat{y} + \frac{1}{2}a \hat{z}$$

$$\mathbf{a}_2 = \frac{1}{2}a \hat{x} + \frac{1}{2}a \hat{z}$$

$$\mathbf{a}_3 = \frac{1}{2}a \hat{x} + \frac{1}{2}a \hat{y}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	=	0	=	0	(4a) Pt I
\mathbf{B}_2	=	$\frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	=	$\frac{1}{4}a \hat{x} + \frac{1}{4}a \hat{y} + \frac{1}{4}a \hat{z}$	(8c) K I
\mathbf{B}_3	=	$\frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	=	$\frac{3}{4}a \hat{x} + \frac{3}{4}a \hat{y} + \frac{3}{4}a \hat{z}$	(8c) K I
\mathbf{B}_4	=	$-x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	=	$ax_3 \hat{x}$	(24e) Cl I
\mathbf{B}_5	=	$x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	=	$-ax_3 \hat{x}$	(24e) Cl I
\mathbf{B}_6	=	$x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	=	$ax_3 \hat{y}$	(24e) Cl I
\mathbf{B}_7	=	$-x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	=	$-ax_3 \hat{y}$	(24e) Cl I
\mathbf{B}_8	=	$x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	=	$ax_3 \hat{z}$	(24e) Cl I
\mathbf{B}_9	=	$-x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	=	$-ax_3 \hat{z}$	(24e) Cl I

References

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- [4] P. P. Ewald and C. Hermann, eds., *Strukturbericht 1913-1928* (Akademische Verlagsgesellschaft M. B. H., Leipzig, 1931).
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Found in

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