

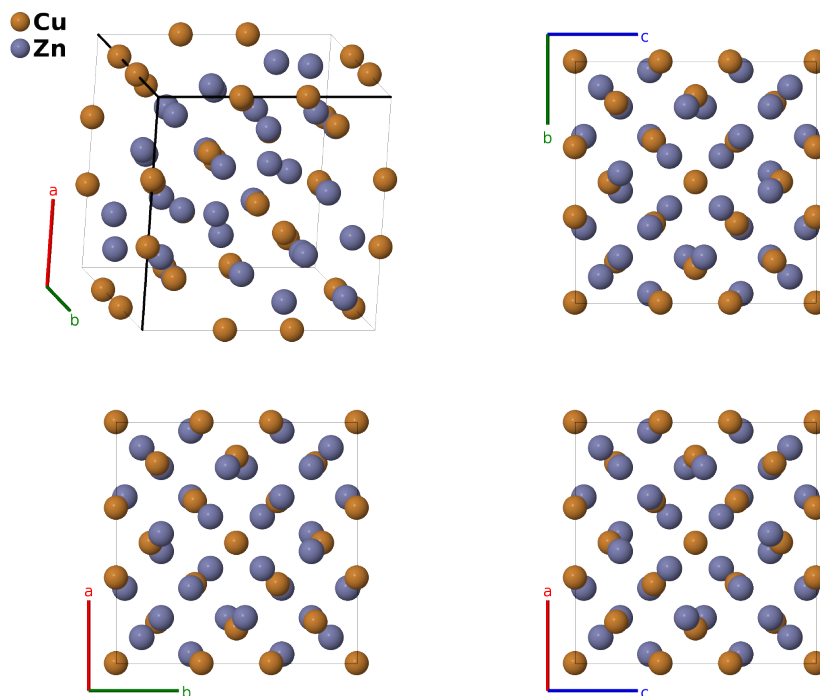
γ -Brass (Cu_5Zn_8 , $D8_2$) Structure: A5B8_cI52_217_ce_cg-001

This structure originally had the label A5B8_cI52_217_ce_cg. Calls to that address will be redirected here.

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<https://aflow.org/p/AYKC>

https://aflow.org/p/A5B8_cI52_217_ce_cg-001



Prototype	Cu_5Zn_8
AFLOW prototype label	A5B8_cI52_217_ce_cg-001
<i>Strukturbericht</i> designation	$D8_2$
Mineral name	γ -brass
ICSD	240667
Pearson symbol	cI52
Space group number	217
Space group symbol	$I\bar{4}3m$
AFLOW prototype command	<code>aflow --proto=A5B8_cI52_217_ce_cg-001 --params=a, x1, x2, x3, x4, z4</code>

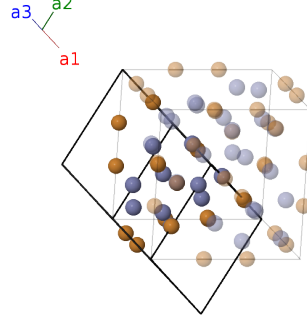
Other compounds with this structure

$\text{Cu}_x\text{Zn}_{1-x}$, $\text{Cu}_x\text{Cd}_{1-x}$, $\text{Fe}_x\text{Zn}_{1-x}$

- γ -brass comes in a variety of compositions. See the $D8_1$ and $D8_3$ structure pages for more information.
- We use the data from (Gourdon, 2007) for $\text{Cu}_{5.00}\text{Zn}_{8.00}$. At this composition the authors state that the sites are fully occupied.
- (Mizutani, 2010) classifies this as a “I-cell” γ -brass.

Body-centered Cubic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= -\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{2}a \hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{2}a \hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}} - \frac{1}{2}a \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	$= 2x_1 \mathbf{a}_1 + 2x_1 \mathbf{a}_2 + 2x_1 \mathbf{a}_3$	$=$	$ax_1 \hat{\mathbf{x}} + ax_1 \hat{\mathbf{y}} + ax_1 \hat{\mathbf{z}}$	(8c)	Cu I
\mathbf{B}_2	$= -2x_1 \mathbf{a}_3$	$=$	$-ax_1 \hat{\mathbf{x}} - ax_1 \hat{\mathbf{y}} + ax_1 \hat{\mathbf{z}}$	(8c)	Cu I
\mathbf{B}_3	$= -2x_1 \mathbf{a}_2$	$=$	$-ax_1 \hat{\mathbf{x}} + ax_1 \hat{\mathbf{y}} - ax_1 \hat{\mathbf{z}}$	(8c)	Cu I
\mathbf{B}_4	$= -2x_1 \mathbf{a}_1$	$=$	$ax_1 \hat{\mathbf{x}} - ax_1 \hat{\mathbf{y}} - ax_1 \hat{\mathbf{z}}$	(8c)	Cu I
\mathbf{B}_5	$= 2x_2 \mathbf{a}_1 + 2x_2 \mathbf{a}_2 + 2x_2 \mathbf{a}_3$	$=$	$ax_2 \hat{\mathbf{x}} + ax_2 \hat{\mathbf{y}} + ax_2 \hat{\mathbf{z}}$	(8c)	Zn I
\mathbf{B}_6	$= -2x_2 \mathbf{a}_3$	$=$	$-ax_2 \hat{\mathbf{x}} - ax_2 \hat{\mathbf{y}} + ax_2 \hat{\mathbf{z}}$	(8c)	Zn I
\mathbf{B}_7	$= -2x_2 \mathbf{a}_2$	$=$	$-ax_2 \hat{\mathbf{x}} + ax_2 \hat{\mathbf{y}} - ax_2 \hat{\mathbf{z}}$	(8c)	Zn I
\mathbf{B}_8	$= -2x_2 \mathbf{a}_1$	$=$	$ax_2 \hat{\mathbf{x}} - ax_2 \hat{\mathbf{y}} - ax_2 \hat{\mathbf{z}}$	(8c)	Zn I
\mathbf{B}_9	$= x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	$=$	$ax_3 \hat{\mathbf{x}}$	(12e)	Cu II
\mathbf{B}_{10}	$= -x_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	$=$	$-ax_3 \hat{\mathbf{x}}$	(12e)	Cu II
\mathbf{B}_{11}	$= x_3 \mathbf{a}_1 + x_3 \mathbf{a}_3$	$=$	$ax_3 \hat{\mathbf{y}}$	(12e)	Cu II
\mathbf{B}_{12}	$= -x_3 \mathbf{a}_1 - x_3 \mathbf{a}_3$	$=$	$-ax_3 \hat{\mathbf{y}}$	(12e)	Cu II
\mathbf{B}_{13}	$= x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2$	$=$	$ax_3 \hat{\mathbf{z}}$	(12e)	Cu II
\mathbf{B}_{14}	$= -x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2$	$=$	$-ax_3 \hat{\mathbf{z}}$	(12e)	Cu II
\mathbf{B}_{15}	$= (x_4 + z_4) \mathbf{a}_1 + (x_4 + z_4) \mathbf{a}_2 + 2x_4 \mathbf{a}_3$	$=$	$ax_4 \hat{\mathbf{x}} + ax_4 \hat{\mathbf{y}} + az_4 \hat{\mathbf{z}}$	(24g)	Zn II
\mathbf{B}_{16}	$= -(x_4 - z_4) \mathbf{a}_1 - (x_4 - z_4) \mathbf{a}_2 - 2x_4 \mathbf{a}_3$	$=$	$-ax_4 \hat{\mathbf{x}} - ax_4 \hat{\mathbf{y}} + az_4 \hat{\mathbf{z}}$	(24g)	Zn II
\mathbf{B}_{17}	$= (x_4 - z_4) \mathbf{a}_1 - (x_4 + z_4) \mathbf{a}_2$	$=$	$-ax_4 \hat{\mathbf{x}} + ax_4 \hat{\mathbf{y}} - az_4 \hat{\mathbf{z}}$	(24g)	Zn II
\mathbf{B}_{18}	$= -(x_4 + z_4) \mathbf{a}_1 + (x_4 - z_4) \mathbf{a}_2$	$=$	$ax_4 \hat{\mathbf{x}} - ax_4 \hat{\mathbf{y}} - az_4 \hat{\mathbf{z}}$	(24g)	Zn II
\mathbf{B}_{19}	$= 2x_4 \mathbf{a}_1 + (x_4 + z_4) \mathbf{a}_2 + (x_4 + z_4) \mathbf{a}_3$	$=$	$az_4 \hat{\mathbf{x}} + ax_4 \hat{\mathbf{y}} + ax_4 \hat{\mathbf{z}}$	(24g)	Zn II
\mathbf{B}_{20}	$= -2x_4 \mathbf{a}_1 - (x_4 - z_4) \mathbf{a}_2 - (x_4 - z_4) \mathbf{a}_3$	$=$	$az_4 \hat{\mathbf{x}} - ax_4 \hat{\mathbf{y}} - ax_4 \hat{\mathbf{z}}$	(24g)	Zn II
\mathbf{B}_{21}	$= (x_4 - z_4) \mathbf{a}_2 - (x_4 + z_4) \mathbf{a}_3$	$=$	$-az_4 \hat{\mathbf{x}} - ax_4 \hat{\mathbf{y}} + ax_4 \hat{\mathbf{z}}$	(24g)	Zn II

$$\mathbf{B}_{22} = -(x_4 + z_4) \mathbf{a}_2 + (x_4 - z_4) \mathbf{a}_3 = -az_4 \hat{\mathbf{x}} + ax_4 \hat{\mathbf{y}} - ax_4 \hat{\mathbf{z}} \quad (24g) \quad \text{Zn II}$$

$$\mathbf{B}_{23} = (x_4 + z_4) \mathbf{a}_1 + 2x_4 \mathbf{a}_2 + (x_4 + z_4) \mathbf{a}_3 = ax_4 \hat{\mathbf{x}} + az_4 \hat{\mathbf{y}} + ax_4 \hat{\mathbf{z}} \quad (24g) \quad \text{Zn II}$$

$$\mathbf{B}_{24} = -(x_4 - z_4) \mathbf{a}_1 - 2x_4 \mathbf{a}_2 - (x_4 - z_4) \mathbf{a}_3 = -ax_4 \hat{\mathbf{x}} + az_4 \hat{\mathbf{y}} - ax_4 \hat{\mathbf{z}} \quad (24g) \quad \text{Zn II}$$

$$\mathbf{B}_{25} = -(x_4 + z_4) \mathbf{a}_1 + (x_4 - z_4) \mathbf{a}_3 = ax_4 \hat{\mathbf{x}} - az_4 \hat{\mathbf{y}} - ax_4 \hat{\mathbf{z}} \quad (24g) \quad \text{Zn II}$$

$$\mathbf{B}_{26} = (x_4 - z_4) \mathbf{a}_1 - (x_4 + z_4) \mathbf{a}_3 = -ax_4 \hat{\mathbf{x}} - az_4 \hat{\mathbf{y}} + ax_4 \hat{\mathbf{z}} \quad (24g) \quad \text{Zn II}$$

References

- [1] O. Gourdon, D. Gout, D. J. Williams, T. Proffen, S. Hobbs, and G. J. Miller, *Atomic Distributions in the γ -Brass Structure of the Cu-Zn System: A Structural and Theoretical Study*, *Inorg. Chem.* **46**, 251–260 (2007), doi:10.1021/ic0616380.
- [2] U. Mizutani, *Hume-Rothery Rules for Structurally Complex Alloy Phases* (CRC Press, Boca Raton, London, New York, 2010).