

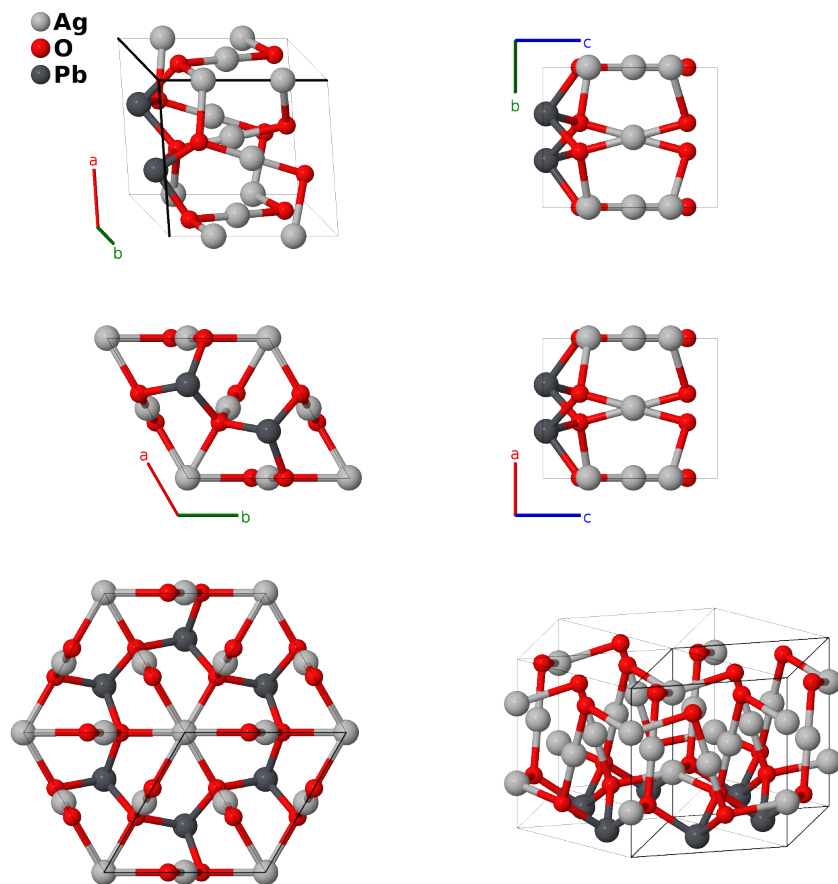
Ag₅Pb₂O₆ Structure: A5B6C2_hP13_157_2ac_2c_b-001

This structure originally had the label A5B6C2_hP13_157_2ac_2c_b. Calls to that address will be redirected here.

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<https://aflow.org/p/1YM0>

https://aflow.org/p/A5B6C2_hP13_157_2ac_2c_b-001

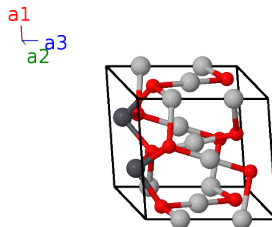


Prototype	Ag ₅ O ₆ Pb ₂
AFLOW prototype label	A5B6C2_hP13_157_2ac_2c_b-001
ICSD	24038
Pearson symbol	hP13
Space group number	157
Space group symbol	<i>P</i> 31 <i>m</i>
AFLOW prototype command	<code>aflow --proto=A5B6C2_hP13_157_2ac_2c_b-001 --params=a, c/a, z₁, z₂, z₃, x₄, z₄, x₅, z₅, x₆, z₆</code>

- The original reference (Byström, 1950) lists this structure as $\text{Ag}_5\text{Pb}_2\text{O}_6$, while (Villars, 1985) lists it as Ag_2PbO_3 . (Byström, 1950) provides three Wyckoff positions for Ag: (1a), (1a), and (3c), while (Villars, 1985) only provides two: (1a) and (3c), giving rise to the stoichiometry discrepancy. While both descriptions yield space group $P31m$ #157, we use the structure and coordinates provided by the original reference (Byström, 1950).

Trigonal (Hexagonal) primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	$= z_1 \mathbf{a}_3$	$=$	$c z_1 \hat{\mathbf{z}}$	(1a)	Ag I
\mathbf{B}_2	$= z_2 \mathbf{a}_3$	$=$	$c z_2 \hat{\mathbf{z}}$	(1a)	Ag II
\mathbf{B}_3	$= \frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c z_3 \hat{\mathbf{z}}$	(2b)	Pb I
\mathbf{B}_4	$= \frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c z_3 \hat{\mathbf{z}}$	(2b)	Pb I
\mathbf{B}_5	$= x_4 \mathbf{a}_1 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a x_4 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a x_4 \hat{\mathbf{y}} + c z_4 \hat{\mathbf{z}}$	(3c)	Ag III
\mathbf{B}_6	$= x_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$\frac{1}{2}a x_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a x_4 \hat{\mathbf{y}} + c z_4 \hat{\mathbf{z}}$	(3c)	Ag III
\mathbf{B}_7	$= -x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$-a x_4 \hat{\mathbf{x}} + c z_4 \hat{\mathbf{z}}$	(3c)	Ag III
\mathbf{B}_8	$= x_5 \mathbf{a}_1 + z_5 \mathbf{a}_3$	$=$	$\frac{1}{2}a x_5 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a x_5 \hat{\mathbf{y}} + c z_5 \hat{\mathbf{z}}$	(3c)	O I
\mathbf{B}_9	$= x_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$\frac{1}{2}a x_5 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a x_5 \hat{\mathbf{y}} + c z_5 \hat{\mathbf{z}}$	(3c)	O I
\mathbf{B}_{10}	$= -x_5 \mathbf{a}_1 - x_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$-a x_5 \hat{\mathbf{x}} + c z_5 \hat{\mathbf{z}}$	(3c)	O I
\mathbf{B}_{11}	$= x_6 \mathbf{a}_1 + z_6 \mathbf{a}_3$	$=$	$\frac{1}{2}a x_6 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a x_6 \hat{\mathbf{y}} + c z_6 \hat{\mathbf{z}}$	(3c)	O II
\mathbf{B}_{12}	$= x_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$\frac{1}{2}a x_6 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a x_6 \hat{\mathbf{y}} + c z_6 \hat{\mathbf{z}}$	(3c)	O II
\mathbf{B}_{13}	$= -x_6 \mathbf{a}_1 - x_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$-a x_6 \hat{\mathbf{x}} + c z_6 \hat{\mathbf{z}}$	(3c)	O II

References

- [1] A. Byström and L. Evers, *The Crystal Structures of Ag_2PbO_3 and $\text{Ag}_5\text{Pb}_2\text{O}_6$* , Acta Chem. Scand. **4**, 613–627 (1950), doi:10.3891/acta.chem.scand.04-0613.

Found in

- [1] P. Villars and L. D. Calvert, eds., *Pearson's Handbook of Crystallographic Data for Intermetallic Phases*, vol. 1 (American Society of Metals, Materials Park, Ohio, 1985).