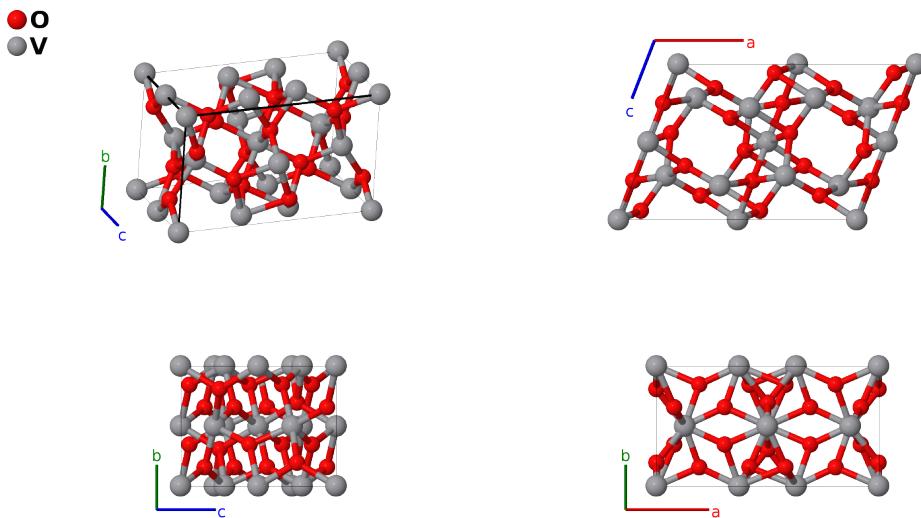


Oxyvanite (V_3O_5) Structure: A5B3_mC32_15_e2f_af-003

Cite this page as: H. Eckert, S. Divilov, A. Zettel, M. J. Mehl, D. Hicks, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*. In preparation.

<https://aflow.org/p/C32G>

https://aflow.org/p/A5B3_mC32_15_e2f_af-003



Prototype	O_5V_3
AFLOW prototype label	A5B3_mC32_15_e2f_af-003
Mineral name	oxyvanite
ICSD	164872
Pearson symbol	mC32
Space group number	15
Space group symbol	$C2/c$
AFLOW prototype command	<code>aflow --proto=A5B3_mC32_15_e2f_af-003 --params=a, b/a, c/a, β, y_2, x_3, y_3, z_3, x_4, y_4, z_4, x_5, y_5, z_5</code>

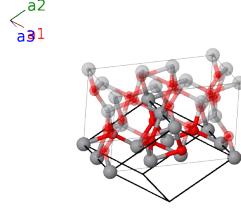
Other compounds with this structure

V_2TiO_5 (berdesinskiite)

- We use the data from Sample I-2-7 of (Armbruster, 2009), taken at room temperature. This structure is metastable up to 423-433K, after which it becomes the stable phase. The ground state of V_3O_5 has a monoclinic cell twice as large as this one, with space group $P2/c$ #13. V_3O_5 can also be found in the metastable form of anosovite, which has the Fe_2TiO_5 ($E4_1$) structure (Weber, 2012).
- In the sample studied by (Armbruster, 2009), the V (4a) site actually has the composition $\text{V}_{0.59}\text{Ti}_{0.41}$, while the V (8f) site is actually $\text{V}_{0.835}\text{Cr}_{0.165}$.

Base-centered Monoclinic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}b\hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}b\hat{\mathbf{y}} \\ \mathbf{a}_3 &= c\cos\beta\hat{\mathbf{x}} + c\sin\beta\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	= 0	= 0	(4a)	V I
\mathbf{B}_2	= $\frac{1}{2}\mathbf{a}_3$	= $\frac{1}{2}c\cos\beta\hat{\mathbf{x}} + \frac{1}{2}c\sin\beta\hat{\mathbf{z}}$	(4a)	V I
\mathbf{B}_3	= $-y_2\mathbf{a}_1 + y_2\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	= $\frac{1}{4}c\cos\beta\hat{\mathbf{x}} + by_2\hat{\mathbf{y}} + \frac{1}{4}c\sin\beta\hat{\mathbf{z}}$	(4e)	O I
\mathbf{B}_4	= $y_2\mathbf{a}_1 - y_2\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	= $\frac{3}{4}c\cos\beta\hat{\mathbf{x}} - by_2\hat{\mathbf{y}} + \frac{3}{4}c\sin\beta\hat{\mathbf{z}}$	(4e)	O I
\mathbf{B}_5	= $(x_3 - y_3)\mathbf{a}_1 + (x_3 + y_3)\mathbf{a}_2 + z_3\mathbf{a}_3$	= $(ax_3 + cz_3\cos\beta)\hat{\mathbf{x}} + by_3\hat{\mathbf{y}} + cz_3\sin\beta\hat{\mathbf{z}}$	(8f)	O II
\mathbf{B}_6	= $-(x_3 + y_3)\mathbf{a}_1 - (x_3 - y_3)\mathbf{a}_2 - (z_3 - \frac{1}{2})\mathbf{a}_3$	= $-(ax_3 + c(z_3 - \frac{1}{2})\cos\beta)\hat{\mathbf{x}} + by_3\hat{\mathbf{y}} - c(z_3 - \frac{1}{2})\sin\beta\hat{\mathbf{z}}$	(8f)	O II
\mathbf{B}_7	= $-(x_3 - y_3)\mathbf{a}_1 - (x_3 + y_3)\mathbf{a}_2 - z_3\mathbf{a}_3$	= $-(ax_3 + cz_3\cos\beta)\hat{\mathbf{x}} - by_3\hat{\mathbf{y}} - cz_3\sin\beta\hat{\mathbf{z}}$	(8f)	O II
\mathbf{B}_8	= $(x_3 + y_3)\mathbf{a}_1 + (x_3 - y_3)\mathbf{a}_2 + (z_3 + \frac{1}{2})\mathbf{a}_3$	= $(ax_3 + c(z_3 + \frac{1}{2})\cos\beta)\hat{\mathbf{x}} - by_3\hat{\mathbf{y}} + c(z_3 + \frac{1}{2})\sin\beta\hat{\mathbf{z}}$	(8f)	O II
\mathbf{B}_9	= $(x_4 - y_4)\mathbf{a}_1 + (x_4 + y_4)\mathbf{a}_2 + z_4\mathbf{a}_3$	= $(ax_4 + cz_4\cos\beta)\hat{\mathbf{x}} + by_4\hat{\mathbf{y}} + cz_4\sin\beta\hat{\mathbf{z}}$	(8f)	O III
\mathbf{B}_{10}	= $-(x_4 + y_4)\mathbf{a}_1 - (x_4 - y_4)\mathbf{a}_2 - (z_4 - \frac{1}{2})\mathbf{a}_3$	= $-(ax_4 + c(z_4 - \frac{1}{2})\cos\beta)\hat{\mathbf{x}} + by_4\hat{\mathbf{y}} - c(z_4 - \frac{1}{2})\sin\beta\hat{\mathbf{z}}$	(8f)	O III
\mathbf{B}_{11}	= $-(x_4 - y_4)\mathbf{a}_1 - (x_4 + y_4)\mathbf{a}_2 - z_4\mathbf{a}_3$	= $-(ax_4 + cz_4\cos\beta)\hat{\mathbf{x}} - by_4\hat{\mathbf{y}} - cz_4\sin\beta\hat{\mathbf{z}}$	(8f)	O III
\mathbf{B}_{12}	= $(x_4 + y_4)\mathbf{a}_1 + (x_4 - y_4)\mathbf{a}_2 + (z_4 + \frac{1}{2})\mathbf{a}_3$	= $(ax_4 + c(z_4 + \frac{1}{2})\cos\beta)\hat{\mathbf{x}} - by_4\hat{\mathbf{y}} + c(z_4 + \frac{1}{2})\sin\beta\hat{\mathbf{z}}$	(8f)	O III
\mathbf{B}_{13}	= $(x_5 - y_5)\mathbf{a}_1 + (x_5 + y_5)\mathbf{a}_2 + z_5\mathbf{a}_3$	= $(ax_5 + cz_5\cos\beta)\hat{\mathbf{x}} + by_5\hat{\mathbf{y}} + cz_5\sin\beta\hat{\mathbf{z}}$	(8f)	V II
\mathbf{B}_{14}	= $-(x_5 + y_5)\mathbf{a}_1 - (x_5 - y_5)\mathbf{a}_2 - (z_5 - \frac{1}{2})\mathbf{a}_3$	= $-(ax_5 + c(z_5 - \frac{1}{2})\cos\beta)\hat{\mathbf{x}} + by_5\hat{\mathbf{y}} - c(z_5 - \frac{1}{2})\sin\beta\hat{\mathbf{z}}$	(8f)	V II
\mathbf{B}_{15}	= $-(x_5 - y_5)\mathbf{a}_1 - (x_5 + y_5)\mathbf{a}_2 - z_5\mathbf{a}_3$	= $-(ax_5 + cz_5\cos\beta)\hat{\mathbf{x}} - by_5\hat{\mathbf{y}} - cz_5\sin\beta\hat{\mathbf{z}}$	(8f)	V II
\mathbf{B}_{16}	= $(x_5 + y_5)\mathbf{a}_1 + (x_5 - y_5)\mathbf{a}_2 + (z_5 + \frac{1}{2})\mathbf{a}_3$	= $(ax_5 + c(z_5 + \frac{1}{2})\cos\beta)\hat{\mathbf{x}} - by_5\hat{\mathbf{y}} + c(z_5 + \frac{1}{2})\sin\beta\hat{\mathbf{z}}$	(8f)	V II

References

- [1] T. Armbruster, E. V. Galuskin, L. Z. Reznitsky, and E. V. Sklyarov, *X-ray structural investigation of the oxyvanite (V_3O_5) - berdesinskiite (V_2TiO_5) series: V^{4+} substituting for octahedrally coordinated Ti^{4+}* , Eur. J. Mineral. **21**, 885–891 (2009), doi:10.1127/0935-1221/2009/0021-1951.

- [2] D. Weber, C. Wessel, C. Reimann, C. Schwickert, A. Müller, T. Ressler, R. Pöttgen, T. Bredow, R. Dronskowski, and M. Lerch, *Anosovite-Type V₃O₅: A New Binary Oxide of Vanadium*, Inorg. Chem. **51**, 8524–8529 (2012), doi:10.1021/ic301096d.