

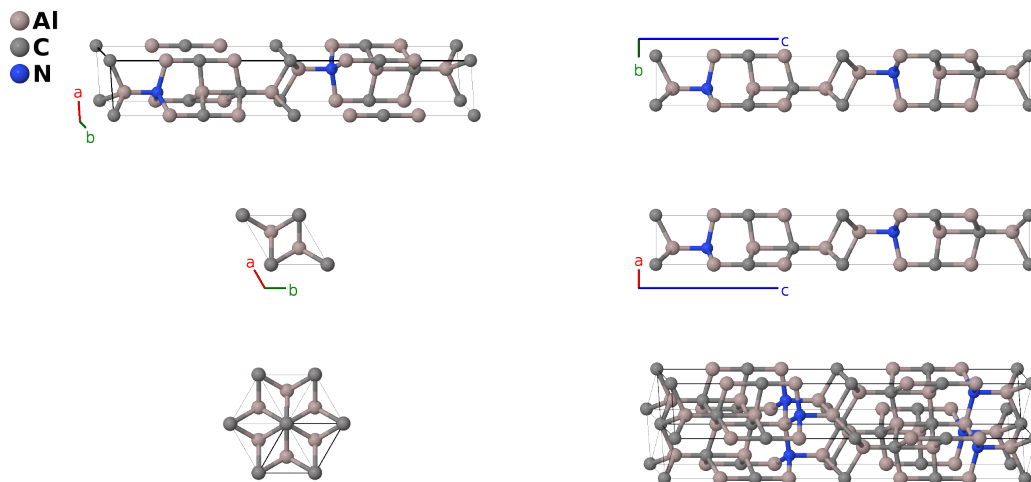
Al₅C₃N (*E*9₄) Structure: A5B3C_hP18_186_2a3b_2ab_b-001

This structure originally had the label A5B3C_hP18_186_2a3b_2ab_b. Calls to that address will be redirected here.

Cite this page as: M. J. Mehl, D. Hicks, C. Toher, O. Levy, R. M. Hanson, G. Hart, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 1*, Comput. Mater. Sci. **136**, S1-828 (2017). doi: 10.1016/j.commatsci.2017.01.017

<https://aflow.org/p/KHLK>

https://aflow.org/p/A5B3C_hP18_186_2a3b_2ab_b-001



Prototype	Al ₅ C ₃ N
AFLOW prototype label	A5B3C_hP18_186_2a3b_2ab_b-001
<i>Strukturbericht</i> designation	<i>E</i> 9 ₄
ICSD	14398
Pearson symbol	hP18
Space group number	186
Space group symbol	<i>P</i> 6 ₃ <i>m</i> c
AFLOW prototype command	<code>aflow --proto=A5B3C_hP18_186_2a3b_2ab_b-001 --params=a, c/a, z₁, z₂, z₃, z₄, z₅, z₆, z₇, z₈, z₉</code>

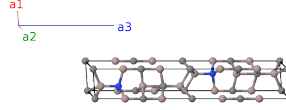
Other compounds with this structure

U₂Al₃C₄

- Space group *P*6₃*m*c #186 allows for an arbitrary placement of the origin of the *z*-axis. We set this by taking *z*₃ = 0 for the C-I atom.

Hexagonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	$z_1 \mathbf{a}_3$	=	$cz_1 \hat{\mathbf{z}}$	(2a)	Al I
\mathbf{B}_2	$(z_1 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_1 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Al I
\mathbf{B}_3	$z_2 \mathbf{a}_3$	=	$cz_2 \hat{\mathbf{z}}$	(2a)	Al II
\mathbf{B}_4	$(z_2 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_2 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Al II
\mathbf{B}_5	$z_3 \mathbf{a}_3$	=	$cz_3 \hat{\mathbf{z}}$	(2a)	C I
\mathbf{B}_6	$(z_3 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_3 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	C I
\mathbf{B}_7	$z_4 \mathbf{a}_3$	=	$cz_4 \hat{\mathbf{z}}$	(2a)	C II
\mathbf{B}_8	$(z_4 + \frac{1}{2}) \mathbf{a}_3$	=	$c(z_4 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	C II
\mathbf{B}_9	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_5 \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(2b)	Al III
\mathbf{B}_{10}	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_5 + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_5 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al III
\mathbf{B}_{11}	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_6 \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(2b)	Al IV
\mathbf{B}_{12}	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_6 + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_6 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al IV
\mathbf{B}_{13}	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_7 \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_7 \hat{\mathbf{z}}$	(2b)	Al V
\mathbf{B}_{14}	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_7 + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_7 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	Al V
\mathbf{B}_{15}	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_8 \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_8 \hat{\mathbf{z}}$	(2b)	C III
\mathbf{B}_{16}	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_8 + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_8 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	C III
\mathbf{B}_{17}	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + z_9 \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + cz_9 \hat{\mathbf{z}}$	(2b)	N I
\mathbf{B}_{18}	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + (z_9 + \frac{1}{2}) \mathbf{a}_3$	=	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + c(z_9 + \frac{1}{2}) \hat{\mathbf{z}}$	(2b)	N I

References

- [1] G. A. Jeffrey and V. Y. Wu, *The structure of the aluminum carbonitrides. II*, Acta Cryst. **20**, 538–547 (1966), doi:10.1107/S0365110X66001208.

Found in

- [1] P. Villars and K. Cenzual, *Pearson's Crystal Data – Crystal Structure Database for Inorganic Compounds* (2013). ASM International.