

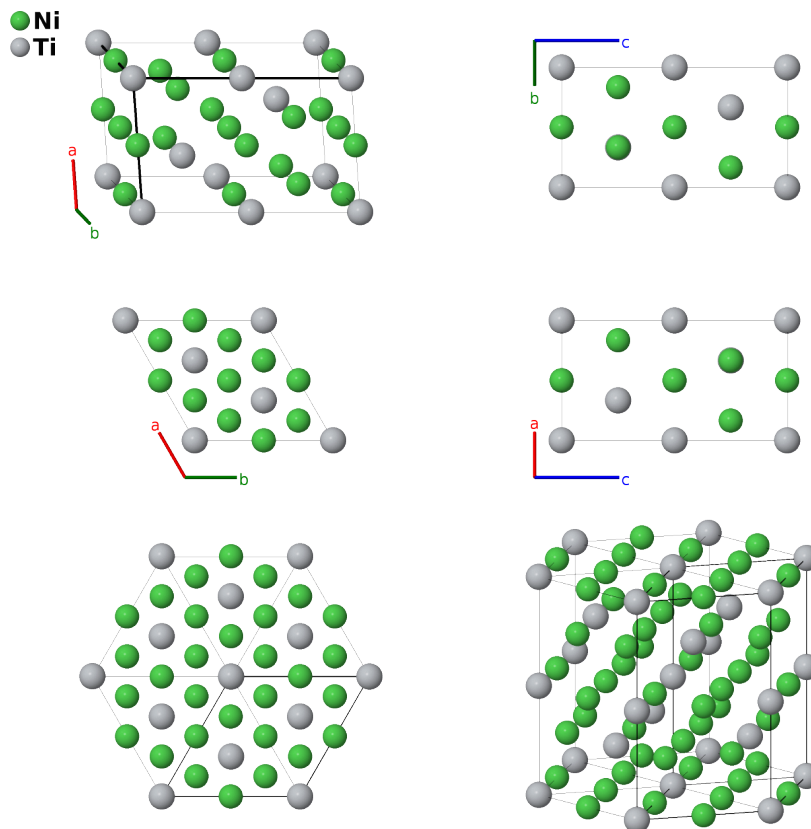
# Ni<sub>3</sub>Ti (*D*0<sub>24</sub>) Structure: A3B\_hP16\_194\_gh\_ac-001

This structure originally had the label A3B\_hP16\_194\_gh\_ac. Calls to that address will be redirected here.

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<https://aflow.org/p/7D2A>

[https://aflow.org/p/A3B\\_hP16\\_194\\_gh\\_ac-001](https://aflow.org/p/A3B_hP16_194_gh_ac-001)



Prototype	Ni <sub>3</sub> Ti
AFLOW prototype label	A3B_hP16_194_gh_ac-001
<i>Strukturbericht</i> designation	<i>D</i> 0 <sub>24</sub>
ICSD	30216
Pearson symbol	hP16
Space group number	194
Space group symbol	<i>P</i> 6 <sub>3</sub> / <i>mmc</i>
AFLOW prototype command	<code>aflow --proto=A3B_hP16_194_gh_ac-001 --params=a, c/a, x<sub>4</sub></code>

## Other compounds with this structure

Al<sub>3</sub>Dy, Al<sub>3</sub>Ru, Co<sub>3</sub>Ti, Ni<sub>3</sub>Ti, Pd<sub>3</sub>Hf, Pd<sub>3</sub>Np, Pd<sub>3</sub>Th, Pd<sub>3</sub>Ti, Pd<sub>3</sub>Zr, Pt<sub>3</sub>Hf, Pt<sub>3</sub>Zr

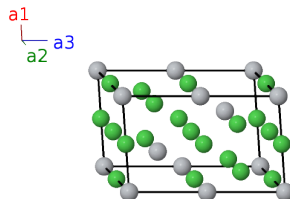
- The internal coordinate  $x_4$  was not determined in any reference we could find. We follow (Villars, 2016) and the ICSD and set  $x_4 = -1/6$ , which places the niobium atoms in line with the Ti atoms in the  $z = 1/4$  and  $z = 3/4$  planes. This is not required by symmetry, and it is likely that the actual value of  $x_4$  will be close, but not equal to,  $-1/6$ .

## Hexagonal primitive vectors

$$\mathbf{a}_1 = \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}}$$

$$\mathbf{a}_2 = \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}}$$

$$\mathbf{a}_3 = c \hat{\mathbf{z}}$$



## Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$0$	$=$	$0$	(2a)	Ti I
$\mathbf{B}_2$	$\frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \hat{\mathbf{z}}$	(2a)	Ti I
$\mathbf{B}_3$	$\frac{1}{3} \mathbf{a}_1 + \frac{2}{3} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(2c)	Ti II
$\mathbf{B}_4$	$\frac{2}{3} \mathbf{a}_1 + \frac{1}{3} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(2c)	Ti II
$\mathbf{B}_5$	$\frac{1}{2} \mathbf{a}_1$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}}$	(6g)	Ni I
$\mathbf{B}_6$	$\frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}}$	(6g)	Ni I
$\mathbf{B}_7$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{2}a \hat{\mathbf{x}}$	(6g)	Ni I
$\mathbf{B}_8$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(6g)	Ni I
$\mathbf{B}_9$	$\frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(6g)	Ni I
$\mathbf{B}_{10}$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}c \hat{\mathbf{z}}$	(6g)	Ni I
$\mathbf{B}_{11}$	$x_4 \mathbf{a}_1 + 2x_4 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(6h)	Ni II
$\mathbf{B}_{12}$	$-2x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(6h)	Ni II
$\mathbf{B}_{13}$	$x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$-\sqrt{3}ax_4 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(6h)	Ni II
$\mathbf{B}_{14}$	$-x_4 \mathbf{a}_1 - 2x_4 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_4 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(6h)	Ni II
$\mathbf{B}_{15}$	$2x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_4 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(6h)	Ni II
$\mathbf{B}_{16}$	$-x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\sqrt{3}ax_4 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(6h)	Ni II

## References

- [1] F. Laves and H. J. Wallbaum, *Die Kristallstruktur von Ni<sub>3</sub>Ti und Si<sub>2</sub>Ti (Zwei neue Typen.)*, Z. Kristallogr. **101**, 78–93 (1939), doi:10.1524/zkri.1939.101.1.78.

## Found in

- [1] P. Villars, ed., *PAULING FILE in: Inorganic Solid Phases* (SpringerMaterials (online database), Heidelberg, 2016).