

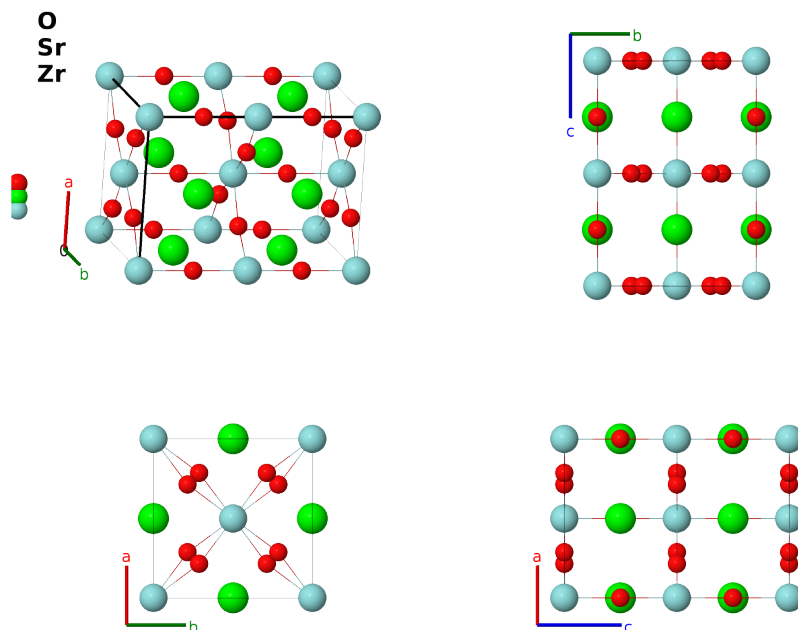
γ -SrZrO₃ Tetragonal Perovskite Structure: A3BC_tI20_140_ah_b_c-002

Cite this page as:

- H. Eckert, S. Divilov, M. J. Mehl, D. Hicks, A. C. Zettl, M. Esters, X. Campilongo, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*, Comput. Mater. Sci. **240**, 112988 (2024), doi: 10.1016/j.commatsci.2024.112988.
- D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.

<https://aflow.org/prototype-encyclopedia/AFN9>

https://aflow.org/prototype-encyclopedia/A3BC_tI20_140_ah_b_c-002



Prototype	O ₃ SrZr
AFLOW prototype label	A3BC_tI20_140_ah_b_c-002
ICSD	89356
CCDC	1649028
Pearson symbol	tI20
Space group number	140
Space group symbol	<i>I4/mcm</i>
AFLOW prototype command	<code>aflow --proto=A3BC_tI20_140_ah_b_c-002 --params=a, c/a, x₄</code>

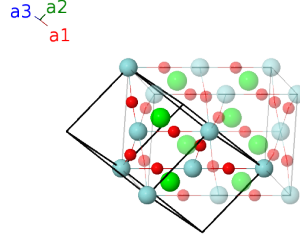
Other compounds with this structure

AsNCr₃, BaIrO₃, BaPbO₃, BaTbO₃, CaTiO₃, CeAlO₃, EuNbO₃, EuTiO₃, GeCMn₃, KMnF₃, RbCaF₃, RuZrO₃, SrHfO₃, SrMoO₃, SrRuO₃, SrSnO₃, SrTcO₃, SrTiO₃, Ba(Bi, In)O₃, Ba(Er, Mo)O₃, Ba(Ho, Mn)O₃, Sr(Mn, Nb)O₃, Sr(Mn, Ru)O₃, Sr(Mn, Sb)O₃, Sr(Ru, Ti)O₃, Sr(Ti, Zr)O₃, (Ba, Ca)SrO₃, (Ba, Sr)HfO₃, (Ca, Sr)TiO₃, (K, Sr)BiO₃, (La, Sr)MnO₃, (Pr, Sr)MnO₃

- As with most perovskites, SrZrO_3 undergoes a number of phase transitions. The temperature-driven phase transitions are summarized by (Kumar, 2017).
 - We have arbitrarily labeled the phases from low- to high-temperature by Greek letters.
 - Below 970K α - SrZrO_3 is in the orthorhombic CaTiO_3 ($Pnma$) structure.
 - In the range 970-1100K, the system is in either the
 - * Base-centered orthorhombic β - SrZrO_3 phase in the $Cmcm$ lueshite structure, or
 - * The body-centered orthorhombic β' - SrZrO_3 phase in the orthorhombic BaPbO_3 structure.
- The former phase seems to be favored by experiment, but it does not share any subgroups with the $Pnma$ structure, so a direct phase transition is difficult. On the other hand the later structure does not show up in the experimental results in the ICSD, only in a theoretical calculation (ICSD 89356) by (Liu, 2012). (Kumar, 2017) mostly favor the first option, while (Fujimori, 2004) favor the second, albeit with no experimental data for the $Imma$ phase.
- Between 1100-1440K the system is in the tetragonal γ - SrZrO_3 phase. (this structure)
 - Above 1440K the system is in the cubic perovskite phase ($E2_1$).
- Here we use the data taken by (Kennedy, 1999) at 1023K.
 - The current structure can be a high-temperature (*e.g.* SrZrO_3) or high-pressure (BaIrO_3) phase, depending on the compound.
 - We attempted to list the chemical formulas for the similar structures in the form ABC_3 , where “A” is the (4b) Wyckoff position, (corresponding to the Ca site in cubic perovskite), “B” (4c) (Ti site) and “C” is (4a)&(8h) (the oxygen site).
 - In addition, both the “A” (4b) and “B” (4c) sites are often alloyed, and even the (4a) and (8h) sites can be a mixture of oxygen and nitrogen.

Body-centered Tetragonal primitive vectors

$$\begin{aligned}
\mathbf{a}_1 &= -\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}} \\
\mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}} \\
\mathbf{a}_3 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}} - \frac{1}{2}c \hat{\mathbf{z}}
\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2$	$=$	$\frac{1}{4}c \hat{\mathbf{z}}$	(4a)	O I
$\mathbf{B}_2$	$= \frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2$	$=$	$\frac{3}{4}c \hat{\mathbf{z}}$	(4a)	O I
$\mathbf{B}_3$	$= \frac{3}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(4b)	Sr I
$\mathbf{B}_4$	$= \frac{1}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{4}c \hat{\mathbf{z}}$	(4b)	Sr I
$\mathbf{B}_5$	$= 0$	$=$	0	(4c)	Zr I
$\mathbf{B}_6$	$= \frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{2}c \hat{\mathbf{z}}$	(4c)	Zr I
$\mathbf{B}_7$	$= \left(x_4 + \frac{1}{2}\right) \mathbf{a}_1 + x_4 \mathbf{a}_2 + \left(2x_4 + \frac{1}{2}\right) \mathbf{a}_3$	$=$	$ax_4 \hat{\mathbf{x}} + a \left(x_4 + \frac{1}{2}\right) \hat{\mathbf{y}}$	(8h)	O II
$\mathbf{B}_8$	$= -\left(x_4 - \frac{1}{2}\right) \mathbf{a}_1 - x_4 \mathbf{a}_2 - \left(2x_4 - \frac{1}{2}\right) \mathbf{a}_3$	$=$	$-ax_4 \hat{\mathbf{x}} - a \left(x_4 - \frac{1}{2}\right) \hat{\mathbf{y}}$	(8h)	O II

$$\mathbf{B}_9 = x_4 \mathbf{a}_1 - \left(x_4 - \frac{1}{2}\right) \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3 = -a \left(x_4 - \frac{1}{2}\right) \hat{\mathbf{x}} + ax_4 \hat{\mathbf{y}} \quad (8h) \quad \text{O II}$$

$$\mathbf{B}_{10} = -x_4 \mathbf{a}_1 + \left(x_4 + \frac{1}{2}\right) \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3 = a \left(x_4 + \frac{1}{2}\right) \hat{\mathbf{x}} - ax_4 \hat{\mathbf{y}} \quad (8h) \quad \text{O II}$$

References

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Found in

- [1] A. Kumar, S. Kumari, H. Borkar, R. S. K., and J. F. Scott, *Experimental verification of the ab initio phase transition sequence in SrZrO₃ and comparisons with SrHfO₃ and SrSnO₃*, npj Comput. Mater. **3**, 2 (2017), doi:10.1038/s41524-016-0002-y.

First cited in

N. Anderson, M. J. Mehl, H. Eckert, S. Divilov, X. Campilongo, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 5*. Submitted to Computational Materials Science (2026).