

CaC₂-I ($C11_a$) Structure:

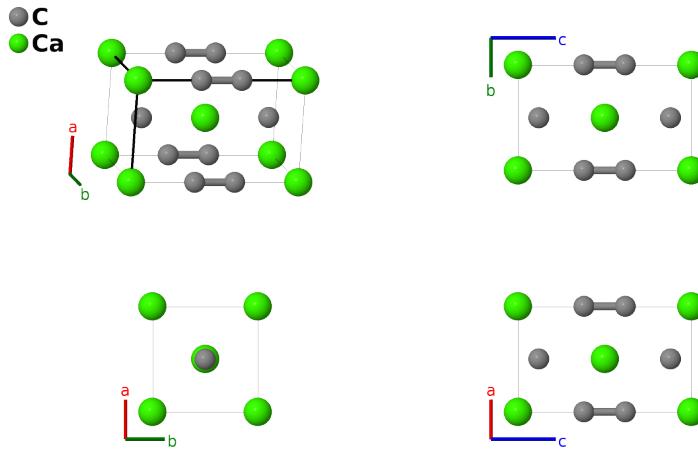
A2B_tI6_139_e_a-001

This structure originally had the label A2B_tI6_139_e_a. Calls to that address will be redirected here.

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<https://aflow.org/p/RDQF>

https://aflow.org/p/A2B_tI6_139_e_a-001



Prototype C₂Ca

AFLOW prototype label A2B_tI6_139_e_a-001

Strukturbericht designation $C11_a$

ICSD 56164

Pearson symbol tI6

Space group number 139

Space group symbol $I4/mmm$

AFLOW prototype command `aflow --proto=A2B_tI6_139_e_a-001
--params=a, c/a, z2`

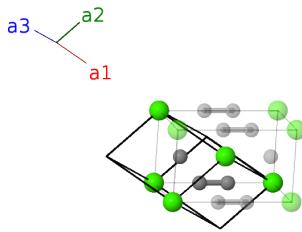
Other compounds with this structure

BaC₂, BaO₂, CeC₂, CsO₂, DyC₂, ErC₂, EuC₂, GdC₂, HoC₂, KO₂, LaC₂, LuC₂, NdC₂, PrC₂, PuC₂, RbO₂, SmC₂, SrC₂, SrO₂, TbC₂, TmC₂, UC₂, YC₂, YbC₂

- (Ewald, 1931) designated both CaC₂ and MoSi₂ as *Strukturbericht C11*. (Smithells, 1955) separated them into $C11_a$ (CaC₂) and $C11_b$ (MoSi₂, AB₂.tI6_139_a-e).
- This is the stable room-temperature structure for CaC₂. At low temperatures it transforms into the ThC₂ (C_g) structure. There is also a meta-stable room temperature structure CaC₂-III (A2B_mC12_12_2i_i).
- (v. Stackelberg, 1930) describes this structure in a face-centered-tetragonal setting. We follow (Smithells, 1955) and place it in the equivalent body-centered tetragonal setting. (Stackelberg, 1930) gives $a = 5.84\text{\AA}$ for the face-centered lattice. If we divide this by $\sqrt{2}$ we get $a = 3.87\text{\AA}$ for the body-centered lattice. The ICSD entry gives $a = 3.83$, which we attribute to an arithmetic error.

Body-centered Tetragonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= -\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} - \frac{1}{2}c\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates	=	Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	0	=	0	(2a)	Ca I
\mathbf{B}_2	$z_2 \mathbf{a}_1 + z_2 \mathbf{a}_2$	=	$cz_2 \hat{\mathbf{z}}$	(4e)	C I
\mathbf{B}_3	$-z_2 \mathbf{a}_1 - z_2 \mathbf{a}_2$	=	$-cz_2 \hat{\mathbf{z}}$	(4e)	C I

References

- [1] M. von Stackelberg, *Die Krystallstruktur des CaC₂*, Naturwissenschaften **18**, 305–306 (1930), doi:10.1007/BF01.
- [2] P. P. Ewald and C. Hermann, eds., *Strukturbericht 1913-1928* (Akademische Verlagsgesellschaft M. B. H., Leipzig, 1931).
- [3] C. J. Smithells, *Metals Reference Book* (Butterworths Scientific, London, 1955), second edn.

Found in

- [1] N.-G. Vannerberg, *The Crystal Structure of Calcium Carbide III*, Acta Chem. Scand. **15**, 769–774 (1961), doi:10.3891/acta.chem.scand.15-0769.