

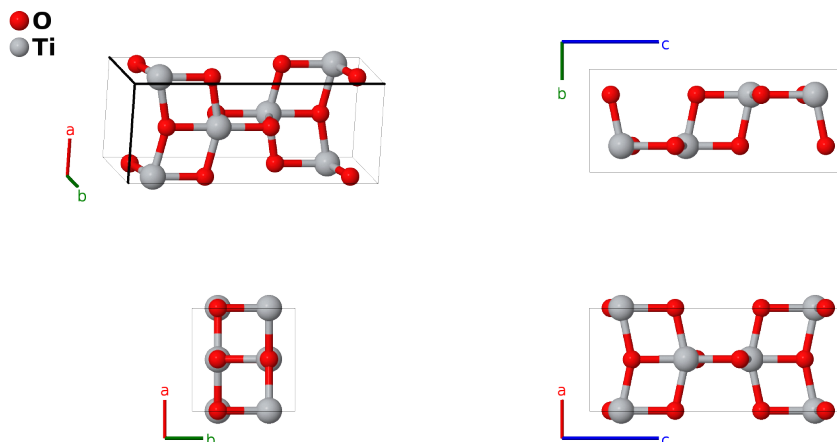
# Anatase (TiO<sub>2</sub>, C<sub>5</sub>) Structure: A2B\_tI12\_141\_e\_a-001

This structure originally had the label **A2B\_tI12\_141\_e\_a**. Calls to that address will be redirected here.

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<https://aflow.org/p/1TUW>

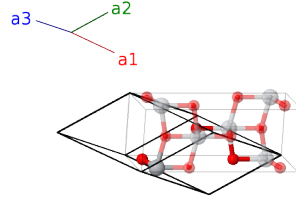
[https://aflow.org/p/A2B\\_tI12\\_141\\_e\\_a-001](https://aflow.org/p/A2B_tI12_141_e_a-001)



Prototype	O <sub>2</sub> Ti
AFLOW prototype label	A2B_tI12_141_e_a-001
<i>Strukturbericht</i> designation	C <sub>5</sub>
Mineral name	anatase
ICSD	63711
Pearson symbol	tI12
Space group number	141
Space group symbol	<i>I</i> 4 <sub>1</sub> / <i>amd</i>
AFLOW prototype command	<code>aflow --proto=A2B_tI12_141_e_a-001 --params=a, c/a, z<sub>2</sub></code>

- TiO<sub>2</sub> can also be found as rutile (C<sub>4</sub>) and brookite (C<sub>21</sub>).
- (Howard, 1991) gives the positions of the atoms in terms of setting 1 of space group *I*4<sub>1</sub>/*amd* #141. Previously, when we translated this to our standard setting 2 we entered an incorrect position for the oxygen z coordinate, giving incorrect Ti-O bond lengths and angles. This has been corrected in the current version of the CIF and the POSCAR.

**Body-centered Tetragonal primitive vectors**



$$\begin{aligned}\mathbf{a}_1 &= -\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} - \frac{1}{2}c\hat{\mathbf{z}}\end{aligned}$$

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## Basis vectors

	Lattice coordinates	=	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{7}{8}\mathbf{a}_1 + \frac{1}{8}\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	=	$\frac{3}{4}a\hat{\mathbf{y}} + \frac{1}{8}c\hat{\mathbf{z}}$	(4a)	Ti I
$\mathbf{B}_2$	$= \frac{1}{8}\mathbf{a}_1 + \frac{7}{8}\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{4}a\hat{\mathbf{y}} + \frac{3}{8}c\hat{\mathbf{z}}$	(4a)	Ti I
$\mathbf{B}_3$	$= (z_2 + \frac{1}{4})\mathbf{a}_1 + z_2\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	=	$\frac{1}{4}a\hat{\mathbf{y}} + cz_2\hat{\mathbf{z}}$	(8e)	O I
$\mathbf{B}_4$	$= z_2\mathbf{a}_1 + (z_2 + \frac{1}{4})\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{y}} + c(z_2 - \frac{1}{4})\hat{\mathbf{z}}$	(8e)	O I
$\mathbf{B}_5$	$= -(z_2 - \frac{3}{4})\mathbf{a}_1 - z_2\mathbf{a}_2 + \frac{3}{4}\mathbf{a}_3$	=	$\frac{3}{4}a\hat{\mathbf{y}} - cz_2\hat{\mathbf{z}}$	(8e)	O I
$\mathbf{B}_6$	$= -z_2\mathbf{a}_1 - (z_2 - \frac{3}{4})\mathbf{a}_2 + \frac{1}{4}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{4}a\hat{\mathbf{y}} - c(z_2 - \frac{1}{4})\hat{\mathbf{z}}$	(8e)	O I

## References

- [1] C. J. Howard, T. M. Sabine, and F. Dickson, *Structural and thermal parameters for rutile and anatase*, Acta Crystallogr. Sect. B **47**, 462–468 (1991), doi:10.1107/S010876819100335X.