

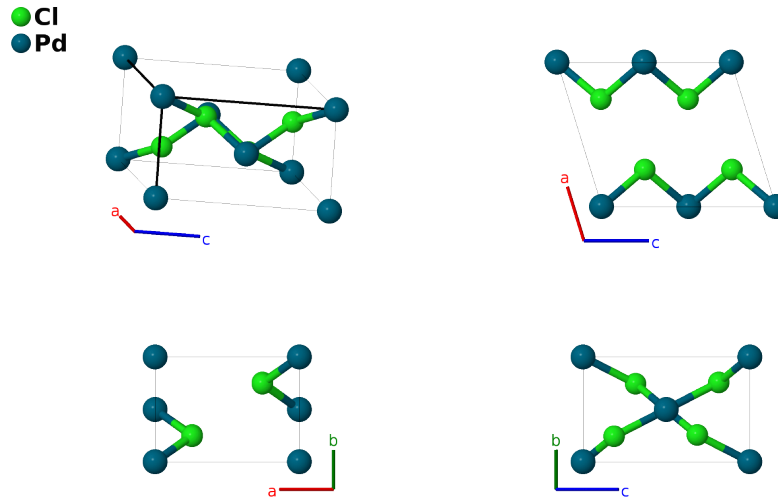
# $\gamma$ -PdCl<sub>2</sub> Structure: A2B\_mP6\_14\_e\_a-001

This structure originally had the label **A2B\_mP6\_14\_e\_a**. Calls to that address will be redirected here.

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<https://aflow.org/p/K9AG>

[https://aflow.org/p/A2B\\_mP6\\_14\\_e\\_a-001](https://aflow.org/p/A2B_mP6_14_e_a-001)

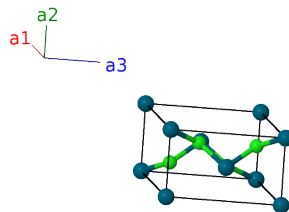


<b>Prototype</b>	Cl <sub>2</sub> Pd
<b>AFLOW prototype label</b>	A2B_mP6_14_e_a-001
<b>ICSD</b>	421221
<b>Pearson symbol</b>	mP6
<b>Space group number</b>	14
<b>Space group symbol</b>	<i>P</i> 2 <sub>1</sub> / <i>c</i>
<b>AFLOW prototype command</b>	<code>aflow --proto=A2B_mP6_14_e_a-001 --params=a, b/a, c/a, <math>\beta</math>, x<sub>2</sub>, y<sub>2</sub>, z<sub>2</sub></code>

- PdCl<sub>2</sub> is known to exist in four different structures at ambient pressure (Evers, 2010):
  - orthorhombic  $\alpha$ -PdCl<sub>2</sub>,
  - rhombohedral  $\beta$ -PdCl<sub>2</sub>,
  - monoclinic  $\gamma$ -PdCl<sub>2</sub> (this structure), and
  - monoclinic  $\delta$ -PdCl<sub>2</sub>.
- (Evers, 2010) place the palladium atoms on the (2c) Wyckoff position. We have shifted the origin so that the palladium atoms are at the (2a) position.
- Data was taken at 300K.

## Simple Monoclinic primitive vectors

$$\begin{aligned} \mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \cos \beta \hat{\mathbf{x}} + c \sin \beta \hat{\mathbf{z}} \end{aligned}$$



## Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$0$	$=$	$0$	(2a)	Pd I
$\mathbf{B}_2$	$\frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} c \cos \beta \hat{\mathbf{x}} + \frac{1}{2} b \hat{\mathbf{y}} + \frac{1}{2} c \sin \beta \hat{\mathbf{z}}$	(2a)	Pd I
$\mathbf{B}_3$	$x_2 \mathbf{a}_1 + y_2 \mathbf{a}_2 + z_2 \mathbf{a}_3$	$=$	$(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} + by_2 \hat{\mathbf{y}} + cz_2 \sin \beta \hat{\mathbf{z}}$	(4e)	Cl I
$\mathbf{B}_4$	$-x_2 \mathbf{a}_1 + (y_2 + \frac{1}{2}) \mathbf{a}_2 - (z_2 - \frac{1}{2}) \mathbf{a}_3$	$=$	$-(ax_2 + c(z_2 - \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} + b(y_2 + \frac{1}{2}) \hat{\mathbf{y}} - c(z_2 - \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	Cl I
$\mathbf{B}_5$	$-x_2 \mathbf{a}_1 - y_2 \mathbf{a}_2 - z_2 \mathbf{a}_3$	$=$	$-(ax_2 + cz_2 \cos \beta) \hat{\mathbf{x}} - by_2 \hat{\mathbf{y}} - cz_2 \sin \beta \hat{\mathbf{z}}$	(4e)	Cl I
$\mathbf{B}_6$	$x_2 \mathbf{a}_1 - (y_2 - \frac{1}{2}) \mathbf{a}_2 + (z_2 + \frac{1}{2}) \mathbf{a}_3$	$=$	$(ax_2 + c(z_2 + \frac{1}{2}) \cos \beta) \hat{\mathbf{x}} - b(y_2 - \frac{1}{2}) \hat{\mathbf{y}} + c(z_2 + \frac{1}{2}) \sin \beta \hat{\mathbf{z}}$	(4e)	Cl I

## References

- [1] J.Evers, W. Beck, M. Göbel, S. Jakob, P. Mayer, G. Oehlinger, M. Rotter, and T. Klapötke, *The Structures of  $\delta$ -PdCl<sub>2</sub> and  $\gamma$ -PdCl<sub>2</sub>: Phases with Negative Thermal Expansion in One Direction*, *Angew. Chem. Int. Ed.* **49**, 5677–5682 (2010), doi:10.1002/anie.201000680.