

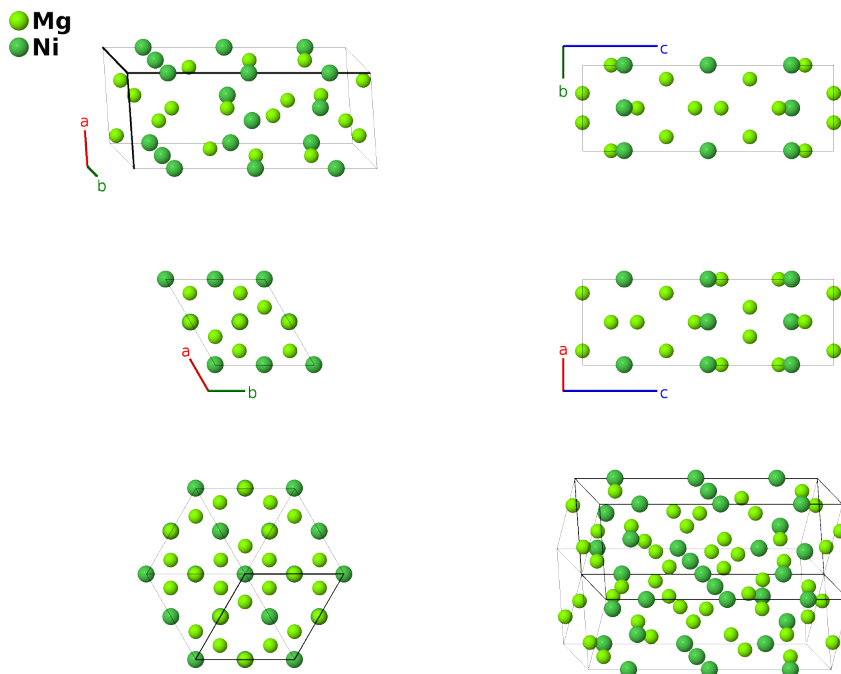
Mg₂Ni (*C_a*) Structure: A2B_hP18_180_fj_ac-001

This structure originally had the label A2B_hP18_180_fi_bd. Calls to that address will be redirected here.

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<https://aflow.org/p/TDY8>

https://aflow.org/p/A2B_hP18_180_fj_ac-001



Prototype	Mg ₂ Ni
AFLOW prototype label	A2B_hP18_180_fj_ac-001
<i>Strukturbericht</i> designation	<i>C_a</i>
ICSD	104912
Pearson symbol	hP18
Space group number	180
Space group symbol	<i>P6₂22</i>
AFLOW prototype command	<code>aflow --proto=A2B_hP18_180_fj_ac-001 --params=a, c/a, z₃, x₄</code>

Other compounds with this structure

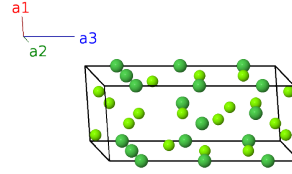
CuMg₄Ni, MoSn₂

- This structure can also be found in the enantiomorphic space group *P6₄22* #181.

- Our original (Mehl, 2017) rendering of this structure swapped the magnesium z_3 and x_4 coordinates.

Hexagonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_3 &= c\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates	=	Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	0	=	0	(3a)	Ni I
\mathbf{B}_2	$\frac{2}{3}\mathbf{a}_3$	=	$\frac{2}{3}c\hat{\mathbf{z}}$	(3a)	Ni I
\mathbf{B}_3	$\frac{1}{3}\mathbf{a}_3$	=	$\frac{1}{3}c\hat{\mathbf{z}}$	(3a)	Ni I
\mathbf{B}_4	$\frac{1}{2}\mathbf{a}_1$	=	$\frac{1}{4}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a\hat{\mathbf{y}}$	(3c)	Ni II
\mathbf{B}_5	$\frac{1}{2}\mathbf{a}_2 + \frac{2}{3}\mathbf{a}_3$	=	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a\hat{\mathbf{y}} + \frac{2}{3}c\hat{\mathbf{z}}$	(3c)	Ni II
\mathbf{B}_6	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + \frac{1}{3}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{3}c\hat{\mathbf{z}}$	(3c)	Ni II
\mathbf{B}_7	$\frac{1}{2}\mathbf{a}_1 + z_3\mathbf{a}_3$	=	$\frac{1}{4}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a\hat{\mathbf{y}} + cz_3\hat{\mathbf{z}}$	(6f)	Mg I
\mathbf{B}_8	$\frac{1}{2}\mathbf{a}_2 + (z_3 + \frac{2}{3})\mathbf{a}_3$	=	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a\hat{\mathbf{y}} + \frac{1}{3}c(3z_3 + 2)\hat{\mathbf{z}}$	(6f)	Mg I
\mathbf{B}_9	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + (z_3 + \frac{1}{3})\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} + c(z_3 + \frac{1}{3})\hat{\mathbf{z}}$	(6f)	Mg I
\mathbf{B}_{10}	$\frac{1}{2}\mathbf{a}_2 - (z_3 - \frac{2}{3})\mathbf{a}_3$	=	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a\hat{\mathbf{y}} - \frac{1}{3}c(3z_3 - 2)\hat{\mathbf{z}}$	(6f)	Mg I
\mathbf{B}_{11}	$\frac{1}{2}\mathbf{a}_1 - z_3\mathbf{a}_3$	=	$\frac{1}{4}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a\hat{\mathbf{y}} - cz_3\hat{\mathbf{z}}$	(6f)	Mg I
\mathbf{B}_{12}	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 - (z_3 - \frac{1}{3})\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} - c(z_3 - \frac{1}{3})\hat{\mathbf{z}}$	(6f)	Mg I
\mathbf{B}_{13}	$x_4\mathbf{a}_1 + 2x_4\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{3}{2}ax_4\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(6j)	Mg II
\mathbf{B}_{14}	$-2x_4\mathbf{a}_1 - x_4\mathbf{a}_2 + \frac{1}{6}\mathbf{a}_3$	=	$-\frac{3}{2}ax_4\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}} + \frac{1}{6}c\hat{\mathbf{z}}$	(6j)	Mg II
\mathbf{B}_{15}	$x_4\mathbf{a}_1 - x_4\mathbf{a}_2 + \frac{5}{6}\mathbf{a}_3$	=	$-\sqrt{3}ax_4\hat{\mathbf{y}} + \frac{5}{6}c\hat{\mathbf{z}}$	(6j)	Mg II
\mathbf{B}_{16}	$-x_4\mathbf{a}_1 - 2x_4\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$-\frac{3}{2}ax_4\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}}$	(6j)	Mg II
\mathbf{B}_{17}	$2x_4\mathbf{a}_1 + x_4\mathbf{a}_2 + \frac{1}{6}\mathbf{a}_3$	=	$\frac{3}{2}ax_4\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4\hat{\mathbf{y}} + \frac{1}{6}c\hat{\mathbf{z}}$	(6j)	Mg II
\mathbf{B}_{18}	$-x_4\mathbf{a}_1 + x_4\mathbf{a}_2 + \frac{5}{6}\mathbf{a}_3$	=	$\sqrt{3}ax_4\hat{\mathbf{y}} + \frac{5}{6}c\hat{\mathbf{z}}$	(6j)	Mg II

References

- [1] J. Schefer, P. Fischer, W. Hälg, F. Stucki, L. Schlapbach, J. J. Didisheim, K. Yvon, and A. F. Andresen, *New structure results for hydrides and deuterides of the hydrogen storage material Mg_2Ni* , *J. Less-Common Met.* **74**, 65–73 (1980), doi:10.1016/0022-5088(80)90074-0.
- [2] M. J. Mehl, D. Hicks, C. Toher, O. Levy, R. M. Hanson, G. Hart, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 1*, *Comput. Mater. Sci.* **136**, S1–S828 (2017), doi:10.1016/j.commatsci.2017.01.017.

Found in

- [1] P. Villars, *Mg_2Ni Crystal Structure* (2016). PAULING FILE in: *Inorganic Solid Phases*, SpringerMaterials (online database), Springer, Heidelberg (ed.) Springer Materials.