

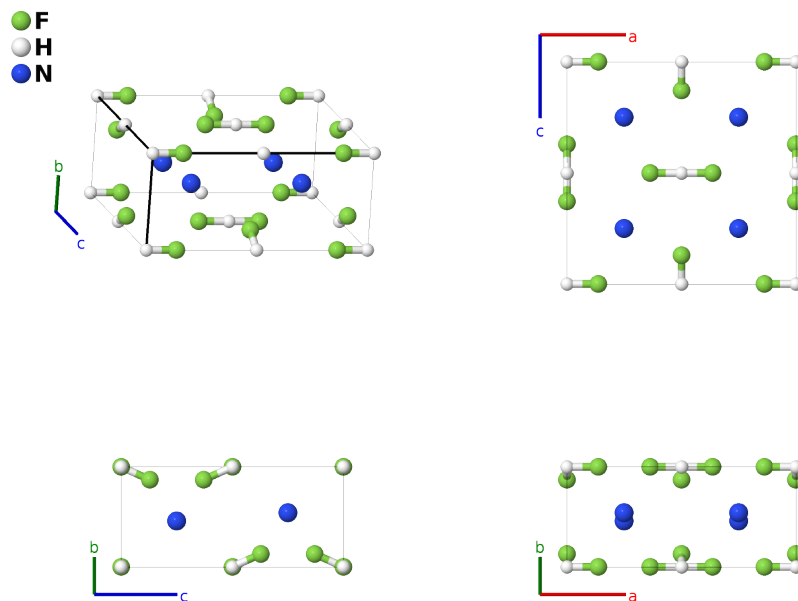
NH₄HF₂ (*F*5₈) Structure: A2BC_oP16_53_eh_ab_g-001

This structure originally had the label A2BC_oP16_53_eh_ab_g. Calls to that address will be redirected here.

Cite this page as: D. Hicks, M. J. Mehl, M. Esters, C. Oses, O. Levy, G. L. W. Hart, C. Toher, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 3*, Comput. Mater. Sci. **199**, 110450 (2021), doi: 10.1016/j.commatsci.2021.110450.

<https://aflow.org/p/DCYY>

https://aflow.org/p/A2BC_oP16_53_eh_ab_g-001



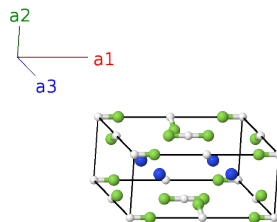
Prototype	F ₂ H ₅ N
AFLOW prototype label	A2BC_oP16_53_eh_ab_g-001
<i>Strukturbericht</i> designation	<i>F</i> 5 ₈
ICSD	28893
Pearson symbol	oP16
Space group number	53
Space group symbol	<i>Pmna</i>
AFLOW prototype command	<code>aflow --proto=A2BC_oP16_53_eh_ab_g-001 --params=a, b/a, c/a, x₃, y₄, y₅, z₅</code>

- This structure was first investigated by (Pauling, 1933) and assigned *Strukturbericht* designation *F*5₈ by (Gottfried, 1937). It was reinvestigated by (Rogers, 1940). Neither paper notes the positions of the hydrogen atoms, but under the assumption that the structure is similar to KHF₂ (*F*5₂), (Downs, 2003) puts some of them between pairs of fluorine atoms. The remaining hydrogen atoms are part of the NH₄ radical.

- The crystal structure was given in the $Pman$ setting of space group #53. We used FINDSYM to change it to the standard $Pmna$ structure.

Simple Orthorhombic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	$=$	0	$=$	0	(2a) H I
\mathbf{B}_2	$=$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} + \frac{1}{2} c \hat{\mathbf{z}}$	(2a) H I
\mathbf{B}_3	$=$	$\frac{1}{2} \mathbf{a}_1$	$=$	$\frac{1}{2} a \hat{\mathbf{x}}$	(2b) H II
\mathbf{B}_4	$=$	$\frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2} c \hat{\mathbf{z}}$	(2b) H II
\mathbf{B}_5	$=$	$x_3 \mathbf{a}_1$	$=$	$a x_3 \hat{\mathbf{x}}$	(4e) F I
\mathbf{B}_6	$=$	$-(x_3 - \frac{1}{2}) \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	$=$	$-a (x_3 - \frac{1}{2}) \hat{\mathbf{x}} + \frac{1}{2} c \hat{\mathbf{z}}$	(4e) F I
\mathbf{B}_7	$=$	$-x_3 \mathbf{a}_1$	$=$	$-a x_3 \hat{\mathbf{x}}$	(4e) F I
\mathbf{B}_8	$=$	$(x_3 + \frac{1}{2}) \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	$=$	$a (x_3 + \frac{1}{2}) \hat{\mathbf{x}} + \frac{1}{2} c \hat{\mathbf{z}}$	(4e) F I
\mathbf{B}_9	$=$	$\frac{1}{4} \mathbf{a}_1 + y_4 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} + b y_4 \hat{\mathbf{y}} + \frac{1}{4} c \hat{\mathbf{z}}$	(4g) NH I
\mathbf{B}_{10}	$=$	$\frac{1}{4} \mathbf{a}_1 - y_4 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{1}{4} a \hat{\mathbf{x}} - b y_4 \hat{\mathbf{y}} + \frac{3}{4} c \hat{\mathbf{z}}$	(4g) NH I
\mathbf{B}_{11}	$=$	$\frac{3}{4} \mathbf{a}_1 - y_4 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} - b y_4 \hat{\mathbf{y}} + \frac{3}{4} c \hat{\mathbf{z}}$	(4g) NH I
\mathbf{B}_{12}	$=$	$\frac{3}{4} \mathbf{a}_1 + y_4 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{3}{4} a \hat{\mathbf{x}} + b y_4 \hat{\mathbf{y}} + \frac{1}{4} c \hat{\mathbf{z}}$	(4g) NH I
\mathbf{B}_{13}	$=$	$y_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$b y_5 \hat{\mathbf{y}} + c z_5 \hat{\mathbf{z}}$	(4h) F II
\mathbf{B}_{14}	$=$	$\frac{1}{2} \mathbf{a}_1 - y_5 \mathbf{a}_2 + (z_5 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - b y_5 \hat{\mathbf{y}} + c (z_5 + \frac{1}{2}) \hat{\mathbf{z}}$	(4h) F II
\mathbf{B}_{15}	$=$	$\frac{1}{2} \mathbf{a}_1 + y_5 \mathbf{a}_2 - (z_5 - \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} + b y_5 \hat{\mathbf{y}} - c (z_5 - \frac{1}{2}) \hat{\mathbf{z}}$	(4h) F II
\mathbf{B}_{16}	$=$	$-y_5 \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$-b y_5 \hat{\mathbf{y}} - c z_5 \hat{\mathbf{z}}$	(4h) F II

References

- [1] M. T. Rogers and L. Helmholz, *A Redetermination of the Parameters in Ammonium Bifluoride*, J. Am. Chem. Soc. **62**, 1533–1536 (1940), doi:10.1021/ja01863a057.
- [2] L. Pauling, *The Crystal Structure of Ammonium Hydrogen Fluoride, NH_4HF_2* , Z. Kristallogr. **85**, 380–391 (1933), doi:10.1524/zkri.1933.85.1.380.
- [3] C. Gottfried and F. Schosberger, eds., *Strukturbericht Band III 1933-1935* (Akademische Verlagsgesellschaft M. B. H., Leipzig, 1937).

Found in

- [1] R. T. Downs and M. Hall-Wallace, *The American Mineralogist Crystal Structure Database*, Am. Mineral. **88**, 247–250 (2003).