

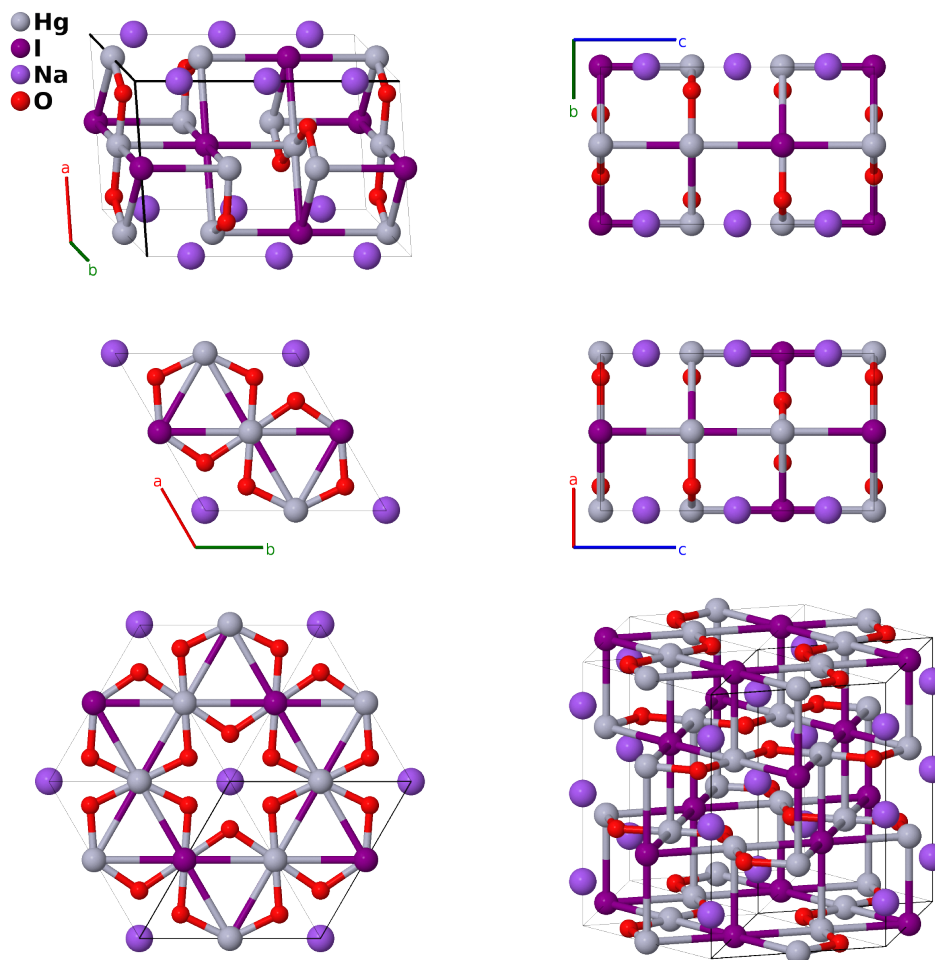
Hg₂O₂NaI Structure: A2BCD2_hP18_180_f_c_b_i-001

This structure originally had the label A2BCD2_hP18_180_f_c_b_i. Calls to that address will be redirected here.

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<https://aflow.org/p/TQ0B>

https://aflow.org/p/A2BCD2_hP18_180_f_c_b_i-001

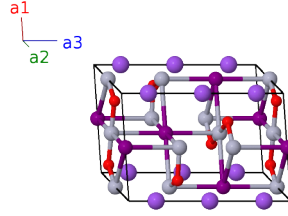


Prototype	Hg ₂ INaO ₂
AFLOW prototype label	A2BCD2_hP18_180_f_c_b_i-001
ICSD	14125
Pearson symbol	hP18
Space group number	180
Space group symbol	$P6_22$
AFLOW prototype command	<code>aflow --proto=A2BCD2_hP18_180_f_c_b_i-001 --params=a, c/a, z₃, x₄</code>

- This structure can also be found in the enantiomorphic space group $P6_422$ #181.

Hexagonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	$= \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{2}c \hat{\mathbf{z}}$	(3b)	Na I
\mathbf{B}_2	$= \frac{1}{6} \mathbf{a}_3$	$=$	$\frac{1}{6}c \hat{\mathbf{z}}$	(3b)	Na I
\mathbf{B}_3	$= \frac{5}{6} \mathbf{a}_3$	$=$	$\frac{5}{6}c \hat{\mathbf{z}}$	(3b)	Na I
\mathbf{B}_4	$= \frac{1}{2} \mathbf{a}_1$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}}$	(3c)	I I
\mathbf{B}_5	$= \frac{1}{2} \mathbf{a}_2 + \frac{2}{3} \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + \frac{2}{3}c \hat{\mathbf{z}}$	(3c)	I I
\mathbf{B}_6	$= \frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{1}{3} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{3}c \hat{\mathbf{z}}$	(3c)	I I
\mathbf{B}_7	$= \frac{1}{2} \mathbf{a}_1 + z_3 \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(6f)	Hg I
\mathbf{B}_8	$= \frac{1}{2} \mathbf{a}_2 + (z_3 + \frac{2}{3}) \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} + \frac{1}{3}c(3z_3 + 2) \hat{\mathbf{z}}$	(6f)	Hg I
\mathbf{B}_9	$= \frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + (z_3 + \frac{1}{3}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + c(z_3 + \frac{1}{3}) \hat{\mathbf{z}}$	(6f)	Hg I
\mathbf{B}_{10}	$= \frac{1}{2} \mathbf{a}_2 - (z_3 - \frac{2}{3}) \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} - \frac{1}{3}c(3z_3 - 2) \hat{\mathbf{z}}$	(6f)	Hg I
\mathbf{B}_{11}	$= \frac{1}{2} \mathbf{a}_1 - z_3 \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} - \frac{\sqrt{3}}{4}a \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(6f)	Hg I
\mathbf{B}_{12}	$= \frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 - (z_3 - \frac{1}{3}) \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} - c(z_3 - \frac{1}{3}) \hat{\mathbf{z}}$	(6f)	Hg I
\mathbf{B}_{13}	$= x_4 \mathbf{a}_1 + 2x_4 \mathbf{a}_2$	$=$	$\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}}$	(6i)	O I
\mathbf{B}_{14}	$= -2x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{2}{3} \mathbf{a}_3$	$=$	$-\frac{3}{2}ax_4 \hat{\mathbf{x}} + \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{2}{3}c \hat{\mathbf{z}}$	(6i)	O I
\mathbf{B}_{15}	$= x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + \frac{1}{3} \mathbf{a}_3$	$=$	$-\sqrt{3}ax_4 \hat{\mathbf{y}} + \frac{1}{3}c \hat{\mathbf{z}}$	(6i)	O I
\mathbf{B}_{16}	$= -x_4 \mathbf{a}_1 - 2x_4 \mathbf{a}_2$	$=$	$-\frac{3}{2}ax_4 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}}$	(6i)	O I
\mathbf{B}_{17}	$= 2x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + \frac{2}{3} \mathbf{a}_3$	$=$	$\frac{3}{2}ax_4 \hat{\mathbf{x}} - \frac{\sqrt{3}}{2}ax_4 \hat{\mathbf{y}} + \frac{2}{3}c \hat{\mathbf{z}}$	(6i)	O I
\mathbf{B}_{18}	$= -x_4 \mathbf{a}_1 + x_4 \mathbf{a}_2 + \frac{1}{3} \mathbf{a}_3$	$=$	$\sqrt{3}ax_4 \hat{\mathbf{y}} + \frac{1}{3}c \hat{\mathbf{z}}$	(6i)	O I

References

- [1] K. Aurivillius, *Least-Squares Refinement of the Crystal Structures of Orthorhombic HgO and of Hg₂O₂NaI*, Acta Chemica Scand. **18**, 1305–1306 (1964), doi:10.3891/acta.chem.scand.18-1305.