

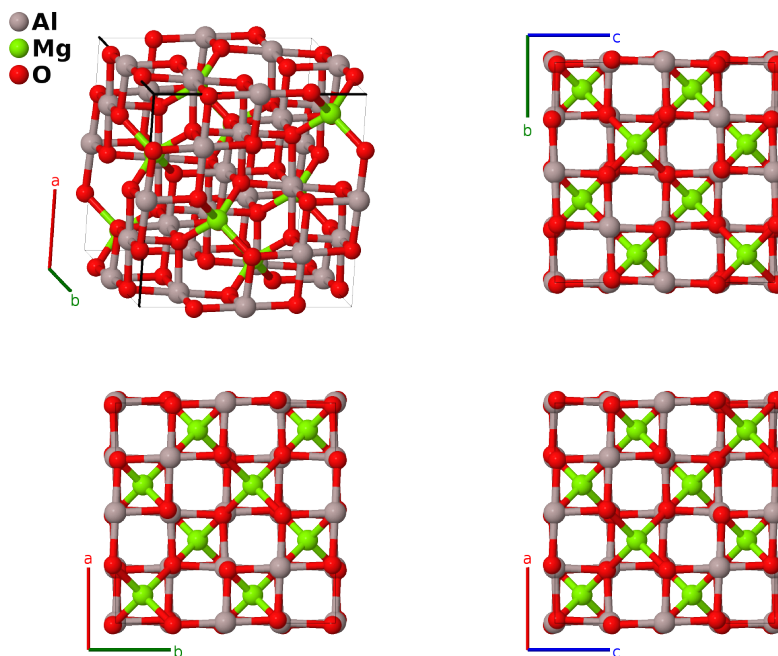
# Spinel ( $\text{Al}_2\text{MgO}_4$ , $H1_1$ ) Structure: A2BC4\_cF56\_227\_c\_b\_e-001

This structure originally had the label A2BC4\_cF56\_227\_d\_a\_e. Calls to that address will be redirected here.

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<https://aflow.org/p/MZYE>

[https://aflow.org/p/A2BC4\\_cF56\\_227\\_c\\_b\\_e-001](https://aflow.org/p/A2BC4_cF56_227_c_b_e-001)



Prototype	$\text{Al}_2\text{MgO}_4$
AFLOW prototype label	A2BC4_cF56_227_c_b_e-001
<i>Strukturbericht</i> designation	$H1_1$
Mineral name	spinel
ICSD	26485
Pearson symbol	cF56
Space group number	227
Space group symbol	$Fd\bar{3}m$
AFLOW prototype command	<code>aflow --proto=A2BC4_cF56_227_c_b_e-001 --params=a, x3</code>

## Other compounds with this structure

$\text{Al}_2\text{CdS}_4$ ,  $\text{Al}_2\text{CrS}_4$ ,  $\text{Al}_2\text{ZnSe}_4$ ,  $\text{Cd}_2\text{VS}_4$ ,  $\text{Co}_2\text{NiS}_4$ ,  $\text{Co}_3\text{Se}_4$ ,  $\text{Cr}_2\text{CdTe}_4$ ,  $\text{Cr}_2\text{CoS}_4$ ,  $\text{Cr}_2\text{CuS}_4$ ,  $\text{Cr}_2\text{CuSe}_4$ ,  $\text{Cr}_2\text{CoO}_4$ ,  $\text{Cr}_2\text{FeO}_4$ ,  $\text{Cr}_2\text{FeS}_4$ ,  $\text{Cr}_2\text{HgS}_4$ ,  $\text{Cr}_2\text{MnS}_4$ ,  $\text{Cr}_2\text{SeZn}_4$ ,  $\text{Cr}_2\text{ZrCd}_4$ ,  $\text{Cr}_2\text{ZrSe}_4$ ,  $\text{Ga}_2\text{CoS}_4$ ,  $\text{Fe}_2\text{NiO}_4$ ,  $\text{In}_2\text{CaS}_4$ ,  $\text{In}_2\text{CdSe}_4$ ,  $\text{In}_2\text{CrS}_4$ ,  $\text{In}_2\text{FeS}_4$ ,  $\text{In}_2\text{HgS}_4$ ,  $\text{In}_2\text{MgS}_4$ ,  $\text{In}_2\text{MnS}_4$ ,  $\text{In}_2\text{NiS}_4$ ,  $\text{In}_2\text{ZrCd}_4$ ,  $\text{Lu}_2\text{FeS}_4$ ,  $\text{Lu}_2\text{MgS}_4$ ,  $\text{Lu}_2\text{MnS}_4$ ,  $\text{Mg}_2\text{GeO}_4$ ,  $\text{Mn}_2\text{ZnTe}_4$ ,  $\text{Ni}_2\text{FeS}_4$ ,  $\text{Sc}_2\text{FeS}_4$ ,  $\text{Sc}_2\text{MnS}_4$ ,  $\text{Ti}_2\text{CuS}_4$ ,  $\text{V}_2\text{CuS}_4$ ,  $\text{V}_2\text{ZnO}_4$ ,  $\text{Yb}_2\text{FeS}_4$ ,  $\text{Yb}_2\text{MnS}_4$ ,  $\text{Co}_3\text{O}_4$ ,  $\text{Co}_3\text{S}_4$ ,  $\text{Fe}_3\text{O}_4$ ,  $\text{Fe}_3\text{S}_4$  (greigite),  $\text{Ni}_3\text{S}_4$

- An *inverse spinel* has four Al atoms on the (8a) sites and (Al,Mg) alloyed on the (16d) sites.
- The binary  $D7_2$  and ternary  $H1_1$  spinel structures are for all intents and purposes identical. We could use  $D7_2$  for the binary spinels and  $H1_1$  for the ternaries, but historically this has not been the case. We dual-list this structure only to keep the historical record intact.
- (Hahn, 1955) has an extensive list of ternary spinels and inverse spinels.

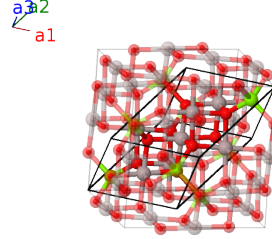
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### Face-centered Cubic primitive vectors

$$\mathbf{a}_1 = \frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{2}a \hat{\mathbf{z}}$$

$$\mathbf{a}_2 = \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{z}}$$

$$\mathbf{a}_3 = \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}}$$




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### Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= \frac{3}{8} \mathbf{a}_1 + \frac{3}{8} \mathbf{a}_2 + \frac{3}{8} \mathbf{a}_3$	$=$	$\frac{3}{8}a \hat{\mathbf{x}} + \frac{3}{8}a \hat{\mathbf{y}} + \frac{3}{8}a \hat{\mathbf{z}}$	(8b)	Mg I
$\mathbf{B}_2$	$= \frac{5}{8} \mathbf{a}_1 + \frac{5}{8} \mathbf{a}_2 + \frac{5}{8} \mathbf{a}_3$	$=$	$\frac{5}{8}a \hat{\mathbf{x}} + \frac{5}{8}a \hat{\mathbf{y}} + \frac{5}{8}a \hat{\mathbf{z}}$	(8b)	Mg I
$\mathbf{B}_3$	$= 0$	$=$	$0$	(16c)	Al I
$\mathbf{B}_4$	$= \frac{1}{2} \mathbf{a}_3$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{1}{4}a \hat{\mathbf{y}}$	(16c)	Al I
$\mathbf{B}_5$	$= \frac{1}{2} \mathbf{a}_2$	$=$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{1}{4}a \hat{\mathbf{z}}$	(16c)	Al I
$\mathbf{B}_6$	$= \frac{1}{2} \mathbf{a}_1$	$=$	$\frac{1}{4}a \hat{\mathbf{y}} + \frac{1}{4}a \hat{\mathbf{z}}$	(16c)	Al I
$\mathbf{B}_7$	$= x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	$=$	$ax_3 \hat{\mathbf{x}} + ax_3 \hat{\mathbf{y}} + ax_3 \hat{\mathbf{z}}$	(32e)	O I
$\mathbf{B}_8$	$= x_3 \mathbf{a}_1 + x_3 \mathbf{a}_2 - (3x_3 - \frac{1}{2}) \mathbf{a}_3$	$=$	$-a(x_3 - \frac{1}{4}) \hat{\mathbf{x}} - a(x_3 - \frac{1}{4}) \hat{\mathbf{y}} + ax_3 \hat{\mathbf{z}}$	(32e)	O I
$\mathbf{B}_9$	$= x_3 \mathbf{a}_1 - (3x_3 - \frac{1}{2}) \mathbf{a}_2 + x_3 \mathbf{a}_3$	$=$	$-a(x_3 - \frac{1}{4}) \hat{\mathbf{x}} + ax_3 \hat{\mathbf{y}} - a(x_3 - \frac{1}{4}) \hat{\mathbf{z}}$	(32e)	O I
$\mathbf{B}_{10}$	$= -(3x_3 - \frac{1}{2}) \mathbf{a}_1 + x_3 \mathbf{a}_2 + x_3 \mathbf{a}_3$	$=$	$ax_3 \hat{\mathbf{x}} - a(x_3 - \frac{1}{4}) \hat{\mathbf{y}} - a(x_3 - \frac{1}{4}) \hat{\mathbf{z}}$	(32e)	O I
$\mathbf{B}_{11}$	$= -x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 + (3x_3 + \frac{1}{2}) \mathbf{a}_3$	$=$	$a(x_3 + \frac{1}{4}) \hat{\mathbf{x}} + a(x_3 + \frac{1}{4}) \hat{\mathbf{y}} - ax_3 \hat{\mathbf{z}}$	(32e)	O I
$\mathbf{B}_{12}$	$= -x_3 \mathbf{a}_1 - x_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	$=$	$-ax_3 \hat{\mathbf{x}} - ax_3 \hat{\mathbf{y}} - ax_3 \hat{\mathbf{z}}$	(32e)	O I
$\mathbf{B}_{13}$	$= -x_3 \mathbf{a}_1 + (3x_3 + \frac{1}{2}) \mathbf{a}_2 - x_3 \mathbf{a}_3$	$=$	$a(x_3 + \frac{1}{4}) \hat{\mathbf{x}} - ax_3 \hat{\mathbf{y}} + a(x_3 + \frac{1}{4}) \hat{\mathbf{z}}$	(32e)	O I
$\mathbf{B}_{14}$	$= (3x_3 + \frac{1}{2}) \mathbf{a}_1 - x_3 \mathbf{a}_2 - x_3 \mathbf{a}_3$	$=$	$-ax_3 \hat{\mathbf{x}} + a(x_3 + \frac{1}{4}) \hat{\mathbf{y}} + a(x_3 + \frac{1}{4}) \hat{\mathbf{z}}$	(32e)	O I

### References

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- [2] H. Hahn, G. Frank, W. Klingler, A. D. Störger, and G. Störger, *Chalkogenide. VI. Über Ternäre Chalkogenide des Aluminiums, Galliums und Indiums mit Zink, Cadmium und Quecksilber*, Z. Anorganische und Allgemeine Chemie **279**, 241–270 (1955), doi:10.1002/zaac.19552790502.

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- [1] R. J. Hill, J. R. Craig, and G. V. Gibbs, *Systematics of the Spinel Structure Type*, Phys. Chem. Miner. **4**, 317–339 (1979).