

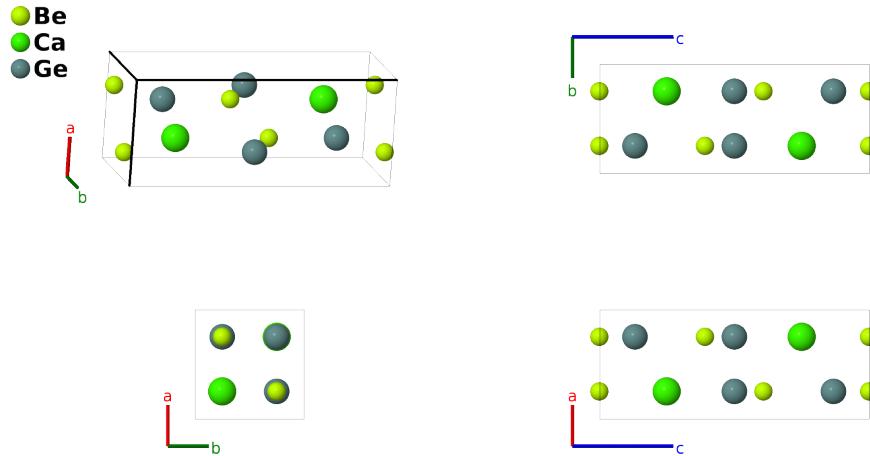
CaBe₂Ge₂ Structure: A2BC2_tP10_129_ac_c_bc-001

This structure originally had the label A2BC2_tP10_129_ac_c_bc. Calls to that address will be redirected here.

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<https://aflow.org/p/H0MM>

https://aflow.org/p/A2BC2_tP10_129_ac_c_bc-001



Prototype	Be ₂ CaGe ₂
AFLOW prototype label	A2BC2_tP10_129_ac_c_bc-001
ICSD	25337
Pearson symbol	tP10
Space group number	129
Space group symbol	$P4/nmm$
AFLOW prototype command	<code>aflow --proto=A2BC2_tP10_129_ac_c_bc-001 --params=a, c/a, z₃, z₄, z₅</code>

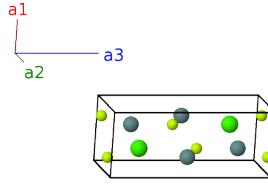
Other compounds with this structure

BaAu₂Sn₂, BaMg₂Pb₂, BaPd₂Sb₂, BaZn₂Sn₂, CeCu₂Sn₂, CeRh₂As₂, CeRh₂P₂, EuAu₂Al₂, EuPd₂Sb₂, EuPt₂Ge₂, HoPt₂Si₂, LaCu₂Sn₂, LaPt₂Bi₂, LaPt₂Ge₂, LaPt₂Si₂, LaRh₂As₂, LaRh₂P₂, LiPd₂Bi₂, NdRh₂As₂, NdRh₂P₂, PrRh₂As₂, PrRh₂P₂, SrAu₂Sn₂, SrCu₂Sn₂, SrPd₂Sb₂, SrPt₂As₂, ThIr₂Si₂, ThPt₂Si₂, UIr₂Si₂, UPt₂Si₂

- This is a ternary form of the $D1_3$ ($BaAl_4$) structure. The atomic positions are approximately the same as in the conventional cell of $BaAl_4$, but the distribution of the atoms on those sites and the resulting relaxation leads to a different structure.
- Space group $P4/nmm$ #129 has two settings, but both have the same z -axis origin, so either setting will do here. We chose our standard setting 2.

Simple Tetragonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= a \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1 =$	$\frac{3}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2$	$\frac{3}{4}a \hat{\mathbf{x}} + \frac{1}{4}a \hat{\mathbf{y}}$	(2a)	Be I
$\mathbf{B}_2 =$	$\frac{1}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{3}{4}a \hat{\mathbf{y}}$	(2a)	Be I
$\mathbf{B}_3 =$	$\frac{3}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$\frac{3}{4}a \hat{\mathbf{x}} + \frac{1}{4}a \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(2b)	Ge I
$\mathbf{B}_4 =$	$\frac{1}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{3}{4}a \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(2b)	Ge I
$\mathbf{B}_5 =$	$\frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_3 \mathbf{a}_3$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{1}{4}a \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(2c)	Be II
$\mathbf{B}_6 =$	$\frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_3 \mathbf{a}_3$	$\frac{3}{4}a \hat{\mathbf{x}} + \frac{3}{4}a \hat{\mathbf{y}} - cz_3 \hat{\mathbf{z}}$	(2c)	Be II
$\mathbf{B}_7 =$	$\frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_4 \mathbf{a}_3$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{1}{4}a \hat{\mathbf{y}} + cz_4 \hat{\mathbf{z}}$	(2c)	Ca I
$\mathbf{B}_8 =$	$\frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_4 \mathbf{a}_3$	$\frac{3}{4}a \hat{\mathbf{x}} + \frac{3}{4}a \hat{\mathbf{y}} - cz_4 \hat{\mathbf{z}}$	(2c)	Ca I
$\mathbf{B}_9 =$	$\frac{1}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + z_5 \mathbf{a}_3$	$\frac{1}{4}a \hat{\mathbf{x}} + \frac{1}{4}a \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(2c)	Ge II
$\mathbf{B}_{10} =$	$\frac{3}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 - z_5 \mathbf{a}_3$	$\frac{3}{4}a \hat{\mathbf{x}} + \frac{3}{4}a \hat{\mathbf{y}} - cz_5 \hat{\mathbf{z}}$	(2c)	Ge II

References

- [1] B. Eisenmann, N. May, W. Müller, and H. Schäfer, *Eine neue strukturelle Variante des BaAl₄-Typs: Der CaBe₂Ge₂-Typ*, Z. Naturforsch. B **27** (1972), doi:10.1515/znb-1972-1008. 1155-1157.