

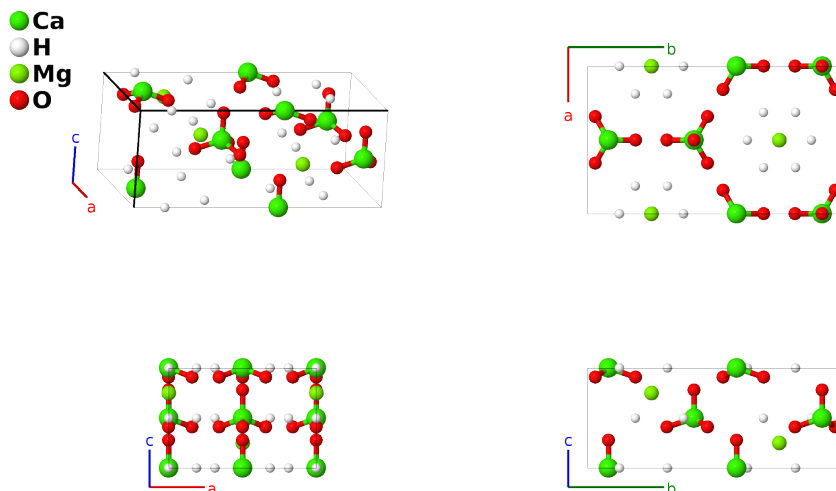
# Mg(ClO<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O (*H*<sub>411</sub>) Structure: A2B6CD8\_oP34\_31\_2a\_2a2b\_a\_4a2b-001

This structure originally had the label A2B6CD8\_oP34\_31\_2a\_2a2b\_a\_4a2b. Calls to that address will be redirected here.

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<https://aflow.org/p/ZN0T>

[https://aflow.org/p/A2B6CD8\\_oP34\\_31\\_2a\\_2a2b\\_a\\_4a2b-001](https://aflow.org/p/A2B6CD8_oP34_31_2a_2a2b_a_4a2b-001)



Prototype	Ca <sub>2</sub> (H <sub>2</sub> O) <sub>6</sub> MgO <sub>8</sub>
AFLOW prototype label	A2B6CD8_oP34_31_2a_2a2b_a_4a2b-001
<i>Strukturbericht</i> designation	<i>H</i> <sub>411</sub>
ICSD	24250
Pearson symbol	oP34
Space group number	31
Space group symbol	<i>Pmn</i> 2 <sub>1</sub>
AFLOW prototype command	<pre>aflow --proto=A2B6CD8_oP34_31_2a_2a2b_a_4a2b-001       --params=a, b/a, c/a, y<sub>1</sub>, z<sub>1</sub>, y<sub>2</sub>, z<sub>2</sub>, y<sub>3</sub>, z<sub>3</sub>, y<sub>4</sub>, z<sub>4</sub>, y<sub>5</sub>, z<sub>5</sub>, y<sub>6</sub>, z<sub>6</sub>, y<sub>7</sub>, z<sub>7</sub>, y<sub>8</sub>, z<sub>8</sub>, y<sub>9</sub>, z<sub>9</sub>, x<sub>10</sub>, y<sub>10</sub>,       z<sub>10</sub>, x<sub>11</sub>, y<sub>11</sub>, z<sub>11</sub>, x<sub>12</sub>, y<sub>12</sub>, z<sub>12</sub>, x<sub>13</sub>, y<sub>13</sub>, z<sub>13</sub></pre>

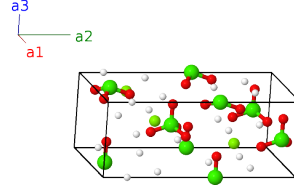
## Other compounds with this structure

Co(ClO<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Fe(ClO<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Mn(ClO<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Ni(ClO<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Zn(ClO<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Mg(BF<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Co(BF<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Fe(BF<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Mn(BF<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Ni(BF<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Zn(BF<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O

- (Gottfried, 1937) writes  $z_{13} = 0.408$  for the coordinate of the (H<sub>2</sub>O-II) molecule, but the value given by West is  $z_{13} = 0$ , which gives a symmetric arrangement of water molecules around the chlorine atom.

## Simple Orthorhombic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= a \hat{\mathbf{x}} \\ \mathbf{a}_2 &= b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



## Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$= y_1 \mathbf{a}_2 + z_1 \mathbf{a}_3$	$=$	$by_1 \hat{\mathbf{y}} + cz_1 \hat{\mathbf{z}}$	(2a)	Ca I
$\mathbf{B}_2$	$= \frac{1}{2} \mathbf{a}_1 - y_1 \mathbf{a}_2 + (z_1 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - by_1 \hat{\mathbf{y}} + c(z_1 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Ca I
$\mathbf{B}_3$	$= y_2 \mathbf{a}_2 + z_2 \mathbf{a}_3$	$=$	$by_2 \hat{\mathbf{y}} + cz_2 \hat{\mathbf{z}}$	(2a)	Ca II
$\mathbf{B}_4$	$= \frac{1}{2} \mathbf{a}_1 - y_2 \mathbf{a}_2 + (z_2 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - by_2 \hat{\mathbf{y}} + c(z_2 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Ca II
$\mathbf{B}_5$	$= y_3 \mathbf{a}_2 + z_3 \mathbf{a}_3$	$=$	$by_3 \hat{\mathbf{y}} + cz_3 \hat{\mathbf{z}}$	(2a)	H I
$\mathbf{B}_6$	$= \frac{1}{2} \mathbf{a}_1 - y_3 \mathbf{a}_2 + (z_3 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - by_3 \hat{\mathbf{y}} + c(z_3 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	H I
$\mathbf{B}_7$	$= y_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	$=$	$by_4 \hat{\mathbf{y}} + cz_4 \hat{\mathbf{z}}$	(2a)	H II
$\mathbf{B}_8$	$= \frac{1}{2} \mathbf{a}_1 - y_4 \mathbf{a}_2 + (z_4 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - by_4 \hat{\mathbf{y}} + c(z_4 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	H II
$\mathbf{B}_9$	$= y_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$by_5 \hat{\mathbf{y}} + cz_5 \hat{\mathbf{z}}$	(2a)	Mg I
$\mathbf{B}_{10}$	$= \frac{1}{2} \mathbf{a}_1 - y_5 \mathbf{a}_2 + (z_5 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - by_5 \hat{\mathbf{y}} + c(z_5 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	Mg I
$\mathbf{B}_{11}$	$= y_6 \mathbf{a}_2 + z_6 \mathbf{a}_3$	$=$	$by_6 \hat{\mathbf{y}} + cz_6 \hat{\mathbf{z}}$	(2a)	O I
$\mathbf{B}_{12}$	$= \frac{1}{2} \mathbf{a}_1 - y_6 \mathbf{a}_2 + (z_6 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - by_6 \hat{\mathbf{y}} + c(z_6 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	O I
$\mathbf{B}_{13}$	$= y_7 \mathbf{a}_2 + z_7 \mathbf{a}_3$	$=$	$by_7 \hat{\mathbf{y}} + cz_7 \hat{\mathbf{z}}$	(2a)	O II
$\mathbf{B}_{14}$	$= \frac{1}{2} \mathbf{a}_1 - y_7 \mathbf{a}_2 + (z_7 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - by_7 \hat{\mathbf{y}} + c(z_7 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	O II
$\mathbf{B}_{15}$	$= y_8 \mathbf{a}_2 + z_8 \mathbf{a}_3$	$=$	$by_8 \hat{\mathbf{y}} + cz_8 \hat{\mathbf{z}}$	(2a)	O III
$\mathbf{B}_{16}$	$= \frac{1}{2} \mathbf{a}_1 - y_8 \mathbf{a}_2 + (z_8 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - by_8 \hat{\mathbf{y}} + c(z_8 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	O III
$\mathbf{B}_{17}$	$= y_9 \mathbf{a}_2 + z_9 \mathbf{a}_3$	$=$	$by_9 \hat{\mathbf{y}} + cz_9 \hat{\mathbf{z}}$	(2a)	O IV
$\mathbf{B}_{18}$	$= \frac{1}{2} \mathbf{a}_1 - y_9 \mathbf{a}_2 + (z_9 + \frac{1}{2}) \mathbf{a}_3$	$=$	$\frac{1}{2} a \hat{\mathbf{x}} - by_9 \hat{\mathbf{y}} + c(z_9 + \frac{1}{2}) \hat{\mathbf{z}}$	(2a)	O IV
$\mathbf{B}_{19}$	$= x_{10} \mathbf{a}_1 + y_{10} \mathbf{a}_2 + z_{10} \mathbf{a}_3$	$=$	$ax_{10} \hat{\mathbf{x}} + by_{10} \hat{\mathbf{y}} + cz_{10} \hat{\mathbf{z}}$	(4b)	H III
$\mathbf{B}_{20}$	$= -(x_{10} - \frac{1}{2}) \mathbf{a}_1 - y_{10} \mathbf{a}_2 + (z_{10} + \frac{1}{2}) \mathbf{a}_3$	$=$	$-a(x_{10} - \frac{1}{2}) \hat{\mathbf{x}} - by_{10} \hat{\mathbf{y}} + c(z_{10} + \frac{1}{2}) \hat{\mathbf{z}}$	(4b)	H III
$\mathbf{B}_{21}$	$= (x_{10} + \frac{1}{2}) \mathbf{a}_1 - y_{10} \mathbf{a}_2 + (z_{10} + \frac{1}{2}) \mathbf{a}_3$	$=$	$a(x_{10} + \frac{1}{2}) \hat{\mathbf{x}} - by_{10} \hat{\mathbf{y}} + c(z_{10} + \frac{1}{2}) \hat{\mathbf{z}}$	(4b)	H III
$\mathbf{B}_{22}$	$= -x_{10} \mathbf{a}_1 + y_{10} \mathbf{a}_2 + z_{10} \mathbf{a}_3$	$=$	$-ax_{10} \hat{\mathbf{x}} + by_{10} \hat{\mathbf{y}} + cz_{10} \hat{\mathbf{z}}$	(4b)	H III
$\mathbf{B}_{23}$	$= x_{11} \mathbf{a}_1 + y_{11} \mathbf{a}_2 + z_{11} \mathbf{a}_3$	$=$	$ax_{11} \hat{\mathbf{x}} + by_{11} \hat{\mathbf{y}} + cz_{11} \hat{\mathbf{z}}$	(4b)	H IV
$\mathbf{B}_{24}$	$= -(x_{11} - \frac{1}{2}) \mathbf{a}_1 - y_{11} \mathbf{a}_2 + (z_{11} + \frac{1}{2}) \mathbf{a}_3$	$=$	$-a(x_{11} - \frac{1}{2}) \hat{\mathbf{x}} - by_{11} \hat{\mathbf{y}} + c(z_{11} + \frac{1}{2}) \hat{\mathbf{z}}$	(4b)	H IV
$\mathbf{B}_{25}$	$= (x_{11} + \frac{1}{2}) \mathbf{a}_1 - y_{11} \mathbf{a}_2 + (z_{11} + \frac{1}{2}) \mathbf{a}_3$	$=$	$a(x_{11} + \frac{1}{2}) \hat{\mathbf{x}} - by_{11} \hat{\mathbf{y}} + c(z_{11} + \frac{1}{2}) \hat{\mathbf{z}}$	(4b)	H IV
$\mathbf{B}_{26}$	$= -x_{11} \mathbf{a}_1 + y_{11} \mathbf{a}_2 + z_{11} \mathbf{a}_3$	$=$	$-ax_{11} \hat{\mathbf{x}} + by_{11} \hat{\mathbf{y}} + cz_{11} \hat{\mathbf{z}}$	(4b)	H IV
$\mathbf{B}_{27}$	$= x_{12} \mathbf{a}_1 + y_{12} \mathbf{a}_2 + z_{12} \mathbf{a}_3$	$=$	$ax_{12} \hat{\mathbf{x}} + by_{12} \hat{\mathbf{y}} + cz_{12} \hat{\mathbf{z}}$	(4b)	O V

$$\begin{aligned}
\mathbf{B}_{28} &= -\left(x_{12} - \frac{1}{2}\right) \mathbf{a}_1 - y_{12} \mathbf{a}_2 + \left(z_{12} + \frac{1}{2}\right) \mathbf{a}_3 &= -a\left(x_{12} - \frac{1}{2}\right) \hat{\mathbf{x}} - by_{12} \hat{\mathbf{y}} + c\left(z_{12} + \frac{1}{2}\right) \hat{\mathbf{z}} &(4b) & \text{O V} \\
\mathbf{B}_{29} &= \left(x_{12} + \frac{1}{2}\right) \mathbf{a}_1 - y_{12} \mathbf{a}_2 + \left(z_{12} + \frac{1}{2}\right) \mathbf{a}_3 &= a\left(x_{12} + \frac{1}{2}\right) \hat{\mathbf{x}} - by_{12} \hat{\mathbf{y}} + c\left(z_{12} + \frac{1}{2}\right) \hat{\mathbf{z}} &(4b) & \text{O V} \\
\mathbf{B}_{30} &= -x_{12} \mathbf{a}_1 + y_{12} \mathbf{a}_2 + z_{12} \mathbf{a}_3 &= -ax_{12} \hat{\mathbf{x}} + by_{12} \hat{\mathbf{y}} + cz_{12} \hat{\mathbf{z}} &(4b) & \text{O V} \\
\mathbf{B}_{31} &= x_{13} \mathbf{a}_1 + y_{13} \mathbf{a}_2 + z_{13} \mathbf{a}_3 &= ax_{13} \hat{\mathbf{x}} + by_{13} \hat{\mathbf{y}} + cz_{13} \hat{\mathbf{z}} &(4b) & \text{O VI} \\
\mathbf{B}_{32} &= -\left(x_{13} - \frac{1}{2}\right) \mathbf{a}_1 - y_{13} \mathbf{a}_2 + \left(z_{13} + \frac{1}{2}\right) \mathbf{a}_3 &= -a\left(x_{13} - \frac{1}{2}\right) \hat{\mathbf{x}} - by_{13} \hat{\mathbf{y}} + c\left(z_{13} + \frac{1}{2}\right) \hat{\mathbf{z}} &(4b) & \text{O VI} \\
\mathbf{B}_{33} &= \left(x_{13} + \frac{1}{2}\right) \mathbf{a}_1 - y_{13} \mathbf{a}_2 + \left(z_{13} + \frac{1}{2}\right) \mathbf{a}_3 &= a\left(x_{13} + \frac{1}{2}\right) \hat{\mathbf{x}} - by_{13} \hat{\mathbf{y}} + c\left(z_{13} + \frac{1}{2}\right) \hat{\mathbf{z}} &(4b) & \text{O VI} \\
\mathbf{B}_{34} &= -x_{13} \mathbf{a}_1 + y_{13} \mathbf{a}_2 + z_{13} \mathbf{a}_3 &= -ax_{13} \hat{\mathbf{x}} + by_{13} \hat{\mathbf{y}} + cz_{13} \hat{\mathbf{z}} &(4b) & \text{O VI}
\end{aligned}$$

## References

- [1] C. D. West, *Crystal Structures of Hydrated Compounds II. Structure Type  $Mg(ClO_4)_2 \cdot 6H_2O$* , Z. Krystallogr. **91**, 480–493 (1935), doi:10.1524/zkri.1935.91.1.480.
- [2] C. Gottfried and F. Schossberger, eds., *Strukturbericht Band III 1933-1935* (Akademische Verlagsgesellschaft M. B. H., Leipzig, 1937).

## Found in

- [1] K. Robertson and D. Bish, *Stability of phases in the  $Mg(ClO_4)_2 \cdot nH_2O$  system and implications for perchlorate occurrences on Mars*, J. Geophys. Res. **116**, E07006 (2011), doi:10.1029/2010JE003754.