

Na₂SO₄ (III) Structure:

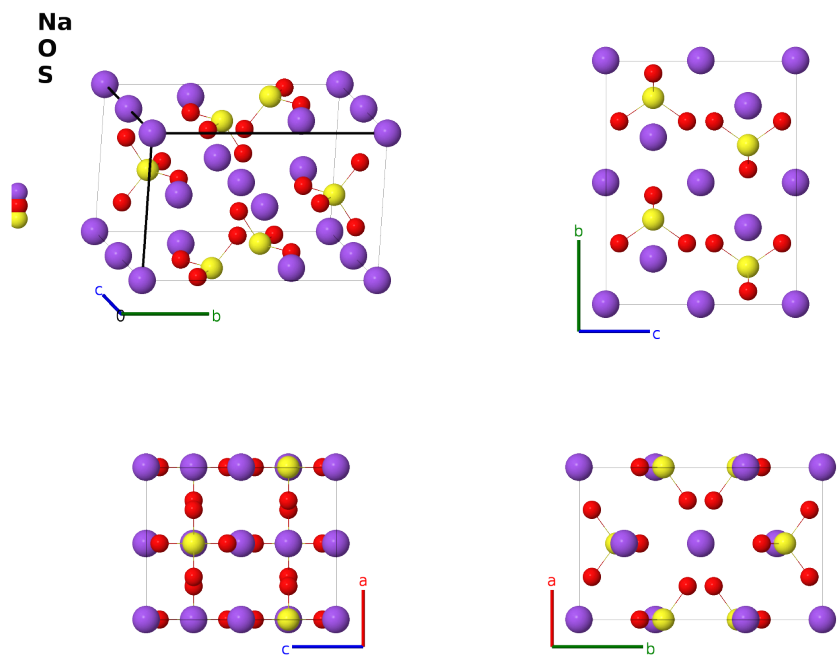
A2B4C_oC28_63_ac_fg_c-001

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<https://aflow.org/prototype-encyclopedia/0R2K>

https://aflow.org/prototype-encyclopedia/A2B4C_oC28_63_ac_fg_c-001



Prototype	Na ₂ O ₄ S
AFLOW prototype label	A2B4C_oC28_63_ac_fg_c-001
ICSD	31687
CCDC	1605685
Pearson symbol	oC28
Space group number	63
Space group symbol	<i>Cmcm</i>
AFLOW prototype command	<code>aflow --proto=A2B4C_oC28_63_ac_fg_c-001</code> <code>--params=a, b/a, c/a, y₂, y₃, y₄, z₄, x₅, y₅</code>

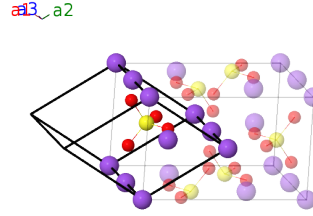
Other compounds with this structure

Na₂CrO₄

- Na_2SO_4 forms in many anhydroous phases (Nord, 1973), which we will reference as we update this site. For now we have
 - Na_2SO_4 III (this structure) and
 - Thenardite (Na_2SO_4 V, $H1_7$).
- (Frevel, 1940) put this structure in the $Pbnn$ setting of space group #52, but AFLOW shows that the positions of the atoms are consistent with the higher symmetry $Cmcm$ #63 space group. Transforming from $Pbnn$ to $Cmcm$ halves the size of the unit cell.
- The date for this structure was taken at room temperature.
- The lattice constants in the ICSD differ slightly (less than 1%) from the published values in (Frevel, 1940). Ordinarily we would use the published values, but since there is a space group shift here we decided to use the ICSD values to avoid confusion.

Base-centered Orthorhombic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a \hat{\mathbf{x}} - \frac{1}{2}b \hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}b \hat{\mathbf{y}} \\ \mathbf{a}_3 &= c \hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	=	0	=	0	(4a) Na I
$\mathbf{B}_2$	=	$\frac{1}{2} \mathbf{a}_3$	=	$\frac{1}{2}c \hat{\mathbf{z}}$	(4a) Na I
$\mathbf{B}_3$	=	$-y_2 \mathbf{a}_1 + y_2 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	=	$by_2 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(4c) Na II
$\mathbf{B}_4$	=	$y_2 \mathbf{a}_1 - y_2 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	=	$-by_2 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(4c) Na II
$\mathbf{B}_5$	=	$-y_3 \mathbf{a}_1 + y_3 \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	=	$by_3 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(4c) S I
$\mathbf{B}_6$	=	$y_3 \mathbf{a}_1 - y_3 \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	=	$-by_3 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(4c) S I
$\mathbf{B}_7$	=	$-y_4 \mathbf{a}_1 + y_4 \mathbf{a}_2 + z_4 \mathbf{a}_3$	=	$by_4 \hat{\mathbf{y}} + cz_4 \hat{\mathbf{z}}$	(8f) O I
$\mathbf{B}_8$	=	$y_4 \mathbf{a}_1 - y_4 \mathbf{a}_2 + (z_4 + \frac{1}{2}) \mathbf{a}_3$	=	$-by_4 \hat{\mathbf{y}} + c(z_4 + \frac{1}{2}) \hat{\mathbf{z}}$	(8f) O I
$\mathbf{B}_9$	=	$-y_4 \mathbf{a}_1 + y_4 \mathbf{a}_2 - (z_4 - \frac{1}{2}) \mathbf{a}_3$	=	$by_4 \hat{\mathbf{y}} - c(z_4 - \frac{1}{2}) \hat{\mathbf{z}}$	(8f) O I
$\mathbf{B}_{10}$	=	$y_4 \mathbf{a}_1 - y_4 \mathbf{a}_2 - z_4 \mathbf{a}_3$	=	$-by_4 \hat{\mathbf{y}} - cz_4 \hat{\mathbf{z}}$	(8f) O I
$\mathbf{B}_{11}$	=	$(x_5 - y_5) \mathbf{a}_1 + (x_5 + y_5) \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	=	$ax_5 \hat{\mathbf{x}} + by_5 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(8g) O II
$\mathbf{B}_{12}$	=	$-(x_5 - y_5) \mathbf{a}_1 - (x_5 + y_5) \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	=	$-ax_5 \hat{\mathbf{x}} - by_5 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(8g) O II
$\mathbf{B}_{13}$	=	$-(x_5 + y_5) \mathbf{a}_1 - (x_5 - y_5) \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	=	$-ax_5 \hat{\mathbf{x}} + by_5 \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(8g) O II
$\mathbf{B}_{14}$	=	$(x_5 + y_5) \mathbf{a}_1 + (x_5 - y_5) \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	=	$ax_5 \hat{\mathbf{x}} - by_5 \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(8g) O II

References

- [1] L. F. Frevel, *The Crystal Structure of Sodium Sulfate III*, J. Chem. Phys. **8**, 290 (1940), doi:10.1063/1.1750647.
- [2] A. C. Nord, *Refinement of the Crystal Structure of Thenardite Na_2SO_4 (V)*, Acta Chem. Scand. **27**, 814–822 (1973).

Found in

[1] *Inorganic Crystal Structure Database*. Entry 31687 [Na₂SO₄ (III)].

First cited in

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