

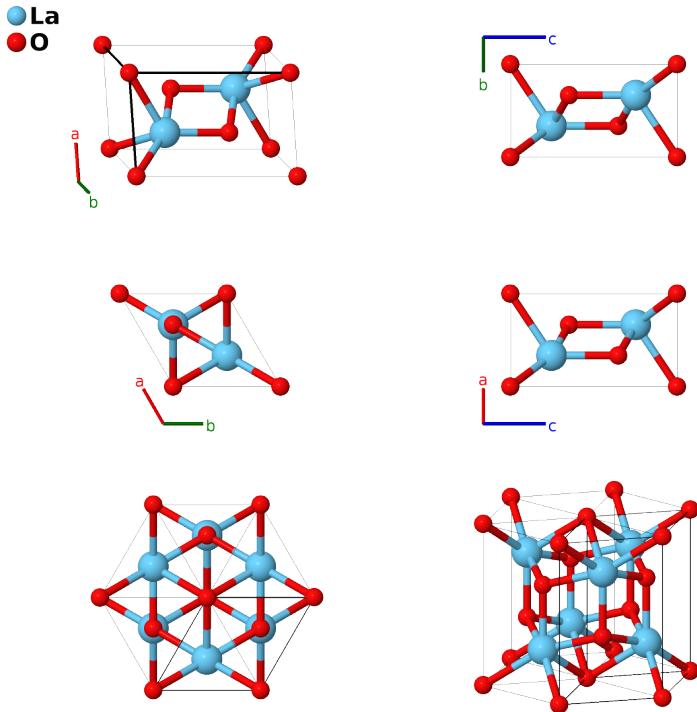
La_2O_3 ($D5_2$) Structure: A2B3_hP5_164_d_ad-001

This structure originally had the label A2B3_hP5_164_d_ad. Calls to that address will be redirected here.

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<https://aflow.org/p/D3QE>

https://aflow.org/p/A2B3_hP5_164_d_ad-001



Prototype	La_2O_3
AFLOW prototype label	A2B3_hP5_164_d_ad-001
Strukturbericht designation	$D5_2$
ICSD	100204
Pearson symbol	hP5
Space group number	164
Space group symbol	$P\bar{3}m1$
AFLOW prototype command	<code>aflow --proto=A2B3_hP5_164_d_ad-001 --params=a, c/a, z₂, z₃</code>

Other compounds with this structure

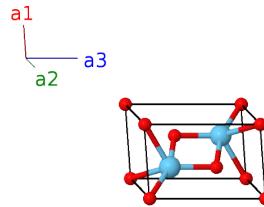
α -Bi₂Mg₃, α -Sb₂Mg₃, Ac₂O₃, As₂Mg₃, Ce₂O₃, Gd₂O₃, Nd₂O₃, Pr₂O₃, Pu₂O₃, Th₂N₃, Th₂N₂O, U₂N₃

- This structure has proved rather controversial:

- (Zachariasen 1926, 1929) originally proposed that the crystal structure of La_2O_3 and other lanthanide series oxides belong to trigonal space group $P\bar{3}21$ #150.
- This was immediately criticized by (Pauling, 1929), who suggested the trigonal space group ($P\bar{3}m1$ #164) later confirmed by (Koehler, 1953).
- Much later, (Aldebert, 1979) stated that the structure was hexagonal, space group $P6_3/mmc$ #194. This would represent a doubling of the $P\bar{3}m1$ unit cell in the z -direction, however they give structural parameters, also used by us and by (Villars, 1991, 2016), which have a density consistent with the $P\bar{3}m1$ structure, and are in reasonable agreement with lattice parameters given in (Pearson, 1958) and (Koehler, 1953). We agree with (Villars, 1991, 2016) that these are close to the correct values for the structure.
- This structure is very similar to Al_3Ni_2 ($D5_{13}$, A3B2.hP5_164.ad_d). We follow (Pearson, 1958) and assign the intermetallics as $D5_{13}$, keeping $D5_2$ for oxides and related compounds.
- The ternary $\text{Ce}_2\text{O}_2\text{S}$ structure, aka CaAl_2Si_2 , is related to this structure, the sulfur atom replacing the oxygen on the (1a) site.

Trigonal (Hexagonal) primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{2}a\hat{\mathbf{y}} \\ \mathbf{a}_3 &= c\hat{\mathbf{z}}\end{aligned}$$



Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	= 0	= 0	(1a)	O I
\mathbf{B}_2	= $\frac{1}{3}\mathbf{a}_1 + \frac{2}{3}\mathbf{a}_2 + z_2\mathbf{a}_3$	= $\frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + cz_2\hat{\mathbf{z}}$	(2d)	La I
\mathbf{B}_3	= $\frac{2}{3}\mathbf{a}_1 + \frac{1}{3}\mathbf{a}_2 - z_2\mathbf{a}_3$	= $\frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} - cz_2\hat{\mathbf{z}}$	(2d)	La I
\mathbf{B}_4	= $\frac{1}{3}\mathbf{a}_1 + \frac{2}{3}\mathbf{a}_2 + z_3\mathbf{a}_3$	= $\frac{1}{2}a\hat{\mathbf{x}} + \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} + cz_3\hat{\mathbf{z}}$	(2d)	O II
\mathbf{B}_5	= $\frac{2}{3}\mathbf{a}_1 + \frac{1}{3}\mathbf{a}_2 - z_3\mathbf{a}_3$	= $\frac{1}{2}a\hat{\mathbf{x}} - \frac{\sqrt{3}}{6}a\hat{\mathbf{y}} - cz_3\hat{\mathbf{z}}$	(2d)	O II

References

- [1] P. Aldebert and J. P. Traverse, *Etude par diffraction neutronique des structures de haute température de La_2O_3 et Nd_2O_3* , Mat. Res. Bull. **14**, 303–323 (1979), doi:10.1016/0025-5408(79)90095-3.
- [2] W. Zachariasen, *Die Kristallstruktur der -Modifikation von den Sesquioxiden der seltenen Erdmetalle. (La_2O_3 , Ce_2O_3 , Pr_2O_3 , Nd_2O_3)*, Z. phys. Chem. **123**, 134 (1926), doi:10.1515/zpch-1926-12309.
- [3] W. H. Zachariasen, *Kürzere Originalmitteilungen und Notizen*, Z. Kristallogr. **70**, 187–188 (1929), doi:10.1524/zkri.1929.70.1.187.
- [4] L. Pauling, *The Crystal Structure of the A-Modification of the Rare Earth Sesquioxides*, Z. Kristallogr. **69**, 415 (1928), doi:10.1524/zkri.1929.69.1.415.
- [5] W. C. Koehler and E. O. Wollan, *Neutron-diffraction study of the structure of the A-form of the rare earth sesquioxides*, Acta Cryst. **6**, 741–742 (1953), doi:10.1107/S0365110X53002076.
- [6] P. Villars, ed., *PAULING FILE in: Inorganic Solid Phases* (SpringerMaterials (online database), Heidelberg, 2016).

- [7] W. B. Pearson, *A Handbook of Lattice Spacings and Structures of Metals and Alloys*, no. N.R.C. No. 4303 in International Series of Monographs on Metal Physics and Physical Metallurgy (Pergamon Press, Oxford, London, Edinburgh, New York, Paris, Frankfort, 1958), 1964 reprint with corrections edn.

Found in

- [1] P. Villars and L. Calvert, *Pearson's Handbook of Crystallographic Data for Intermetallic Phases* (ASM International, Materials Park, OK, 1991), vol. IV, chap. , p. 4189.