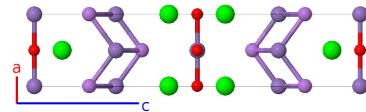
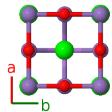
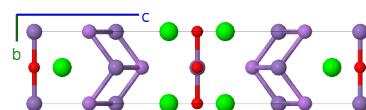
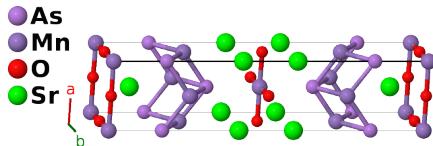


# Sr<sub>2</sub>Mn<sub>3</sub>As<sub>2</sub>O<sub>2</sub> Structure: A2B3C2D2\_tI18\_139\_e\_ad\_c\_e-001

Cite this page as: H. Eckert, S. Divilov, A. Zettel, M. J. Mehl, D. Hicks, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 4*. In preparation.

<https://aflow.org/p/JBSJ>

[https://aflow.org/p/A2B3C2D2\\_tI18\\_139\\_e\\_ad\\_c\\_e-001](https://aflow.org/p/A2B3C2D2_tI18_139_e_ad_c_e-001)



Prototype	As <sub>2</sub> Mn <sub>3</sub> O <sub>2</sub> Sr <sub>2</sub>
AFLOW prototype label	A2B3C2D2_tI18_139_e_ad_c_e-001
ICSD	81798
Pearson symbol	tI18
Space group number	139
Space group symbol	<i>I</i> 4/ <i>mmm</i>
AFLOW prototype command	aflow --proto=A2B3C2D2_tI18_139_e_ad_c_e-001 --params= <i>a</i> , <i>c/a</i> , <i>z</i> <sub>4</sub> , <i>z</i> <sub>5</sub>

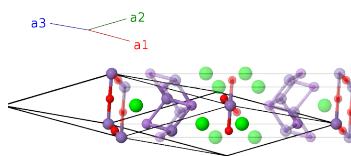
## Other compounds with this structure

La<sub>2</sub>O<sub>3</sub>Fe<sub>2</sub>S<sub>2</sub>, La<sub>2</sub>O<sub>3</sub>Fe<sub>2</sub>Se<sub>2</sub>, Sr<sub>2</sub>(Mn<sub>2</sub>Cu)As<sub>2</sub>O<sub>2</sub>, Sr<sub>2</sub>(MnZn<sub>2</sub>)As<sub>2</sub>O<sub>2</sub>, Sr<sub>2</sub>Mn<sub>3</sub>Sb<sub>2</sub>O<sub>2</sub>

- We use the data taken by (Brock, 1996) at 275K.

## Body-centered Tetragonal primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= -\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} - \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}c\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} - \frac{1}{2}c\hat{\mathbf{z}}\end{aligned}$$



## Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	= 0	= 0	(2a)	Mn I

<b>B<sub>2</sub></b>	=	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	=	$\frac{1}{2} a \hat{\mathbf{y}}$	(4c)	O I
<b>B<sub>3</sub></b>	=	$\frac{1}{2} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	=	$\frac{1}{2} a \hat{\mathbf{x}}$	(4c)	O I
<b>B<sub>4</sub></b>	=	$\frac{3}{4} \mathbf{a}_1 + \frac{1}{4} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	=	$\frac{1}{2} a \hat{\mathbf{y}} + \frac{1}{4} c \hat{\mathbf{z}}$	(4d)	Mn II
<b>B<sub>5</sub></b>	=	$\frac{1}{4} \mathbf{a}_1 + \frac{3}{4} \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	=	$\frac{1}{2} a \hat{\mathbf{x}} + \frac{1}{4} c \hat{\mathbf{z}}$	(4d)	Mn II
<b>B<sub>6</sub></b>	=	$z_4 \mathbf{a}_1 + z_4 \mathbf{a}_2$	=	$c z_4 \hat{\mathbf{z}}$	(4e)	As I
<b>B<sub>7</sub></b>	=	$-z_4 \mathbf{a}_1 - z_4 \mathbf{a}_2$	=	$-c z_4 \hat{\mathbf{z}}$	(4e)	As I
<b>B<sub>8</sub></b>	=	$z_5 \mathbf{a}_1 + z_5 \mathbf{a}_2$	=	$c z_5 \hat{\mathbf{z}}$	(4e)	Sr I
<b>B<sub>9</sub></b>	=	$-z_5 \mathbf{a}_1 - z_5 \mathbf{a}_2$	=	$-c z_5 \hat{\mathbf{z}}$	(4e)	Sr I

## References

- [1] S. L. Brock, N. P. Raju, J. E. Greedan, and S. M. Kauzlarich, *The magnetic structures of the mixed layer pnictide oxide compounds Sr<sub>2</sub>Mn<sub>3</sub>Pn<sub>2</sub>O<sub>2</sub> (Pn = As, Sb)*, J. Alloys Compd. **237**, 9–19 (1996), doi:10.1016/0925-8388(95)02066-7.

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- [1] R. Nath, V. O. Garlea, A. I. Goldman, and D. C. Johnston, *Synthesis, structure, and properties of tetragonal Sr<sub>2</sub>M<sub>3</sub>As<sub>2</sub>O<sub>2</sub> (M<sub>3</sub>=Mn<sub>3</sub>, Mn<sub>2</sub>Cu, and MnZn<sub>2</sub>) compounds containing alternating*, Phys. Rev. B **81**, 224513 (2010), doi:10.1103/PhysRevB.81.224513.