

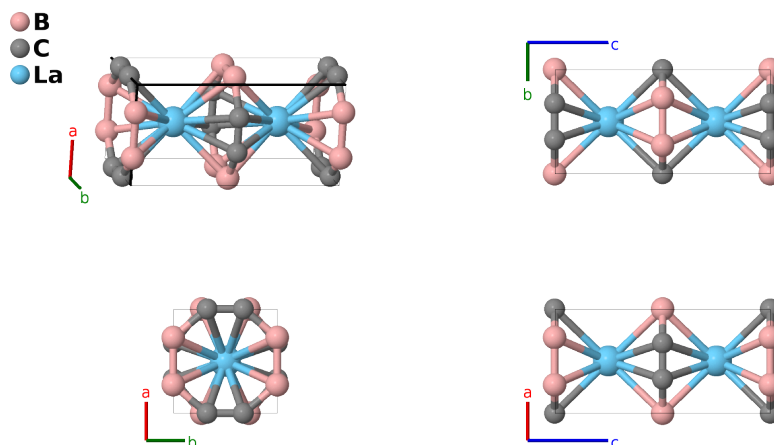
LaB₂C₂ Structure:

A2B2C_tP10_131_j_1_f-001

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<https://aflow.org/p/XLRT>

https://aflow.org/p/A2B2C_tP10_131_j_1_f-001



Prototype	B ₂ C ₂ La
AFLOW prototype label	A2B2C_tP10_131_j_1_f-001
ICSD	23300
Pearson symbol	tP10
Space group number	131
Space group symbol	<i>P</i> 4 ₂ / <i>mmc</i>
AFLOW prototype command	<code>aflow --proto=A2B2C_tP10_131_j_1_f-001 --params=a, c/a, x₂, x₃</code>

Other compounds with this structure

CaB₂C₂, CeB₂C₂, DyB₂C₂, EuB₂C₂, GdB₂C₂, HoB₂C₂, LuB₂C₂, NdB₂C₂, PrB₂C₂, SmB₂C₂, TbB₂C₂, TmB₂C₂, YB₂C₂, YbB₂C₂

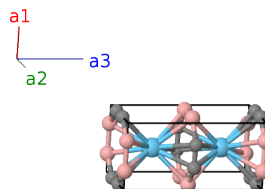
- (Bauer, 1980) placed this structure in space group $P\bar{4}2c$ #112, with lanthanum atoms at the (2a) Wyckoff site, boron at (4h), and carbon at (4i). (Cenzual, 1991) showed that these Wyckoff positions imply an inversion site not present in $P\bar{4}2c$, and so the true space group is $P4_2/mmc$ #131.

Simple Tetragonal primitive vectors

$$\mathbf{a}_1 = a \hat{x}$$

$$\mathbf{a}_2 = a \hat{y}$$

$$\mathbf{a}_3 = c \hat{z}$$



Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
\mathbf{B}_1	$=$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{1}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}} + \frac{1}{4}c \hat{\mathbf{z}}$	(2f) La I
\mathbf{B}_2	$=$	$\frac{1}{2} \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_2 + \frac{3}{4} \mathbf{a}_3$	$=$	$\frac{1}{2}a \hat{\mathbf{x}} + \frac{1}{2}a \hat{\mathbf{y}} + \frac{3}{4}c \hat{\mathbf{z}}$	(2f) La I
\mathbf{B}_3	$=$	$x_2 \mathbf{a}_1$	$=$	$ax_2 \hat{\mathbf{x}}$	(4j) B I
\mathbf{B}_4	$=$	$-x_2 \mathbf{a}_1$	$=$	$-ax_2 \hat{\mathbf{x}}$	(4j) B I
\mathbf{B}_5	$=$	$x_2 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$ax_2 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(4j) B I
\mathbf{B}_6	$=$	$-x_2 \mathbf{a}_2 + \frac{1}{2} \mathbf{a}_3$	$=$	$-ax_2 \hat{\mathbf{y}} + \frac{1}{2}c \hat{\mathbf{z}}$	(4j) B I
\mathbf{B}_7	$=$	$x_3 \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	$=$	$ax_3 \hat{\mathbf{x}} + \frac{1}{2}c \hat{\mathbf{z}}$	(4l) C I
\mathbf{B}_8	$=$	$-x_3 \mathbf{a}_1 + \frac{1}{2} \mathbf{a}_3$	$=$	$-ax_3 \hat{\mathbf{x}} + \frac{1}{2}c \hat{\mathbf{z}}$	(4l) C I
\mathbf{B}_9	$=$	$x_3 \mathbf{a}_2$	$=$	$ax_3 \hat{\mathbf{y}}$	(4l) C I
\mathbf{B}_{10}	$=$	$-x_3 \mathbf{a}_2$	$=$	$-ax_3 \hat{\mathbf{y}}$	(4l) C I

References

- [1] J. Bauer and O. Bars, *The ordering of boron and carbon atoms in the LaB_2C_2 structure*, Acta Crystallogr. Sect. B **36**, 1540–1544 (1980), doi:10.1107/S0567740880006541.

Found in

- [1] K. Cenzual, L. M. Gelato, M. Penzo, and E. Parthé, *Inorganic structure types with revised space groups. I*, Acta Crystallogr. Sect. B **47**, 433–439 (1991), doi:10.1107/S0108768191000903.