

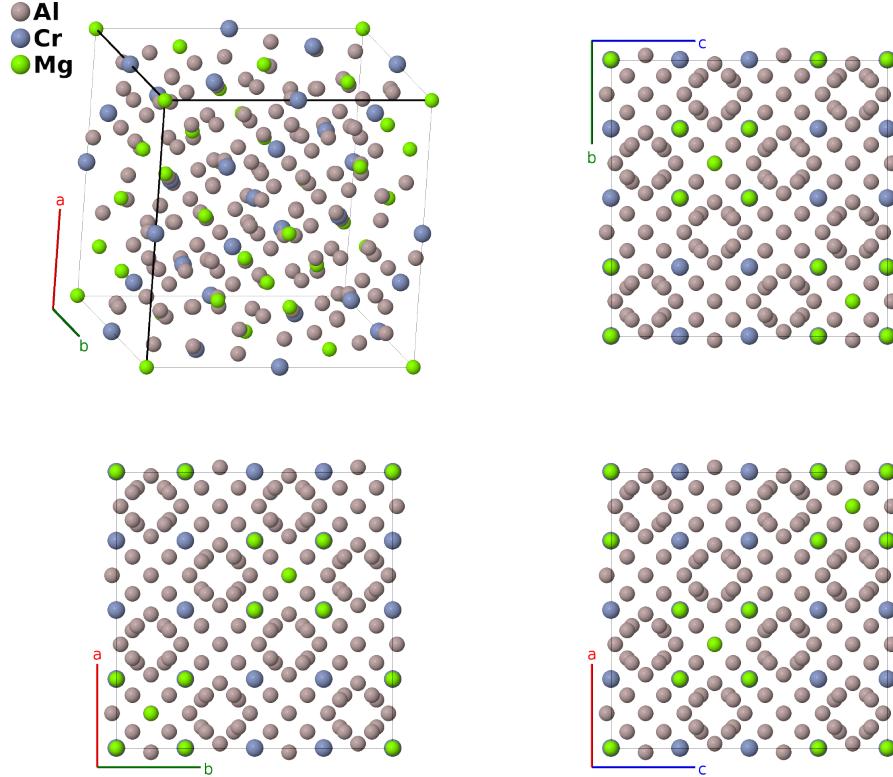
# Mg<sub>3</sub>Cr<sub>2</sub>Al<sub>18</sub> Structure: A18B2C3\_cF184\_227\_fg\_d\_ac-001

This structure originally had the label A18B2C3\_cF184\_227\_fg\_d\_ac. Calls to that address will be redirected here.

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<https://aflow.org/p/3DEV>

[https://aflow.org/p/A18B2C3\\_cF184\\_227\\_fg\\_d\\_ac-001](https://aflow.org/p/A18B2C3_cF184_227_fg_d_ac-001)



Prototype	Al <sub>18</sub> Cr <sub>2</sub> Mg <sub>3</sub>
AFLOW prototype label	A18B2C3_cF184_227_fg_d_ac-001
ICSD	57659
Pearson symbol	cF184
Space group number	227
Space group symbol	$Fd\bar{3}m$
AFLOW prototype command	aflow --proto=A18B2C3_cF184_227_fg_d_ac-001 --params=a, x <sub>4</sub> , x <sub>5</sub> , z <sub>5</sub>

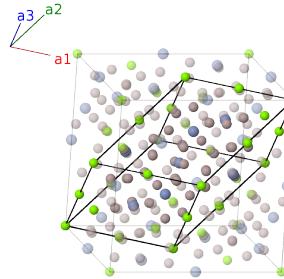
## Other compounds with this structure

LaCr<sub>2</sub>Al<sub>20</sub>, CeCr<sub>2</sub>Al<sub>20</sub>, PrCr<sub>2</sub>Al<sub>20</sub>, SmCr<sub>2</sub>Al<sub>20</sub>, YbCr<sub>2</sub>Al<sub>20</sub>, CeTi<sub>2</sub>Al<sub>20</sub>, PrTi<sub>2</sub>Al<sub>20</sub>, SmTi<sub>2</sub>Al<sub>20</sub>, YbTi<sub>2</sub>Al<sub>20</sub>, CeV<sub>2</sub>Al<sub>20</sub>, GdV<sub>2</sub>Al<sub>20</sub>, LaV<sub>2</sub>Al<sub>20</sub>, PrV<sub>2</sub>Al<sub>20</sub>, SmV<sub>2</sub>Al<sub>20</sub>, CeNi<sub>2</sub>Cd<sub>20</sub>, GdNi<sub>2</sub>Cd<sub>20</sub>, LaNi<sub>2</sub>Cd<sub>20</sub>, NdNi<sub>2</sub>Cd<sub>20</sub>, PrNi<sub>2</sub>Cd<sub>20</sub>, SmNi<sub>2</sub>Cd<sub>20</sub>, YNi<sub>2</sub>Cd<sub>20</sub>, CePd<sub>2</sub>Cd<sub>20</sub>, PrPd<sub>2</sub>Cd<sub>20</sub>, SmPd<sub>2</sub>Cd<sub>20</sub>, UOs<sub>2</sub>Zn<sub>20</sub>

- (Samson, 1958) gives the atomic coordinates in terms of Setting 1 of space group  $F\bar{4}dm$  #227. We have shifted this to the standard Setting 2, where the inversion site of the lattice is at the origin.
- If the (8a), (16c) and (16d) sites are occupied by the same type of atom this becomes the  $Zn_{22}Zr$  structure.
- In the ternary compounds  $LnMX_{20}$ , the rare earth (Ln) metal occupies the (8a) site, the transition metal (M) the (16d) site, and  $X=Al,Cd,Zn$  occupies the (16c), (48gf) and (96g) sites. These compounds are sometimes listed under the  $CeCr_2Al_{20}$  prototype.

### Face-centered Cubic primitive vectors

$$\begin{aligned}\mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}} \\ \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{z}} \\ \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}}\end{aligned}$$



### Basis vectors

	Lattice coordinates		Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$	$\frac{1}{8}\mathbf{a}_1 + \frac{1}{8}\mathbf{a}_2 + \frac{1}{8}\mathbf{a}_3$	=	$\frac{1}{8}a\hat{\mathbf{x}} + \frac{1}{8}a\hat{\mathbf{y}} + \frac{1}{8}a\hat{\mathbf{z}}$	(8a)	Mg I
$\mathbf{B}_2$	$\frac{7}{8}\mathbf{a}_1 + \frac{7}{8}\mathbf{a}_2 + \frac{7}{8}\mathbf{a}_3$	=	$\frac{7}{8}a\hat{\mathbf{x}} + \frac{7}{8}a\hat{\mathbf{y}} + \frac{7}{8}a\hat{\mathbf{z}}$	(8a)	Mg I
$\mathbf{B}_3$	0	=	0	(16c)	Mg II
$\mathbf{B}_4$	$\frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{y}}$	(16c)	Mg II
$\mathbf{B}_5$	$\frac{1}{2}\mathbf{a}_2$	=	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{z}}$	(16c)	Mg II
$\mathbf{B}_6$	$\frac{1}{2}\mathbf{a}_1$	=	$\frac{1}{4}a\hat{\mathbf{y}} + \frac{1}{4}a\hat{\mathbf{z}}$	(16c)	Mg II
$\mathbf{B}_7$	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}}$	(16d)	Cr I
$\mathbf{B}_8$	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2$	=	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}}$	(16d)	Cr I
$\mathbf{B}_9$	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{4}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{4}a\hat{\mathbf{z}}$	(16d)	Cr I
$\mathbf{B}_{10}$	$\frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	=	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{4}a\hat{\mathbf{y}} + \frac{1}{4}a\hat{\mathbf{z}}$	(16d)	Cr I
$\mathbf{B}_{11}$	$-(x_4 - \frac{1}{4})\mathbf{a}_1 + x_4\mathbf{a}_2 + x_4\mathbf{a}_3$	=	$ax_4\hat{\mathbf{x}} + \frac{1}{8}a\hat{\mathbf{y}} + \frac{1}{8}a\hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{12}$	$x_4\mathbf{a}_1 - (x_4 - \frac{1}{4})\mathbf{a}_2 - (x_4 - \frac{1}{4})\mathbf{a}_3$	=	$-a(x_4 - \frac{1}{4})\hat{\mathbf{x}} + \frac{1}{8}a\hat{\mathbf{y}} + \frac{1}{8}a\hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{13}$	$x_4\mathbf{a}_1 - (x_4 - \frac{1}{4})\mathbf{a}_2 + x_4\mathbf{a}_3$	=	$\frac{1}{8}a\hat{\mathbf{x}} + ax_4\hat{\mathbf{y}} + \frac{1}{8}a\hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{14}$	$-(x_4 - \frac{1}{4})\mathbf{a}_1 + x_4\mathbf{a}_2 - (x_4 - \frac{1}{4})\mathbf{a}_3$	=	$\frac{1}{8}a\hat{\mathbf{x}} - a(x_4 - \frac{1}{4})\hat{\mathbf{y}} + \frac{1}{8}a\hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{15}$	$x_4\mathbf{a}_1 + x_4\mathbf{a}_2 - (x_4 - \frac{1}{4})\mathbf{a}_3$	=	$\frac{1}{8}a\hat{\mathbf{x}} + \frac{1}{8}a\hat{\mathbf{y}} + ax_4\hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{16}$	$-(x_4 - \frac{1}{4})\mathbf{a}_1 - (x_4 - \frac{1}{4})\mathbf{a}_2 + x_4\mathbf{a}_3$	=	$\frac{1}{8}a\hat{\mathbf{x}} + \frac{1}{8}a\hat{\mathbf{y}} - a(x_4 - \frac{1}{4})\hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{17}$	$(x_4 + \frac{3}{4})\mathbf{a}_1 - x_4\mathbf{a}_2 + (x_4 + \frac{3}{4})\mathbf{a}_3$	=	$\frac{3}{8}a\hat{\mathbf{x}} + a(x_4 + \frac{3}{4})\hat{\mathbf{y}} + \frac{3}{8}a\hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{18}$	$-x_4\mathbf{a}_1 + (x_4 + \frac{3}{4})\mathbf{a}_2 - x_4\mathbf{a}_3$	=	$\frac{3}{8}a\hat{\mathbf{x}} - ax_4\hat{\mathbf{y}} + \frac{3}{8}a\hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{19}$	$-x_4\mathbf{a}_1 + (x_4 + \frac{3}{4})\mathbf{a}_2 + (x_4 + \frac{3}{4})\mathbf{a}_3$	=	$a(x_4 + \frac{3}{4})\hat{\mathbf{x}} + \frac{3}{8}a\hat{\mathbf{y}} + \frac{3}{8}a\hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{20}$	$(x_4 + \frac{3}{4})\mathbf{a}_1 - x_4\mathbf{a}_2 - x_4\mathbf{a}_3$	=	$-ax_4\hat{\mathbf{x}} + \frac{3}{8}a\hat{\mathbf{y}} + \frac{3}{8}a\hat{\mathbf{z}}$	(48f)	Al I

$\mathbf{B}_{21}$	$-x_4 \mathbf{a}_1 - x_4 \mathbf{a}_2 + (x_4 + \frac{3}{4}) \mathbf{a}_3$	$=$	$\frac{3}{8}a \hat{\mathbf{x}} + \frac{3}{8}a \hat{\mathbf{y}} - ax_4 \hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{22}$	$(x_4 + \frac{3}{4}) \mathbf{a}_1 + (x_4 + \frac{3}{4}) \mathbf{a}_2 - x_4 \mathbf{a}_3$	$=$	$\frac{3}{8}a \hat{\mathbf{x}} + \frac{3}{8}a \hat{\mathbf{y}} + a(x_4 + \frac{3}{4}) \hat{\mathbf{z}}$	(48f)	Al I
$\mathbf{B}_{23}$	$z_5 \mathbf{a}_1 + z_5 \mathbf{a}_2 + (2x_5 - z_5) \mathbf{a}_3$	$=$	$ax_5 \hat{\mathbf{x}} + ax_5 \hat{\mathbf{y}} + az_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{24}$	$z_5 \mathbf{a}_1 + z_5 \mathbf{a}_2 - (2x_5 + z_5 - \frac{1}{2}) \mathbf{a}_3$	$=$	$-a(x_5 - \frac{1}{4}) \hat{\mathbf{x}} - a(x_5 - \frac{1}{4}) \hat{\mathbf{y}} + az_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{25}$	$(2x_5 - z_5) \mathbf{a}_1 - (2x_5 + z_5 - \frac{1}{2}) \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$-a(x_5 - \frac{1}{4}) \hat{\mathbf{x}} + ax_5 \hat{\mathbf{y}} - a(z_5 - \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{26}$	$-(2x_5 + z_5 - \frac{1}{2}) \mathbf{a}_1 + (2x_5 - z_5) \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$ax_5 \hat{\mathbf{x}} - a(x_5 - \frac{1}{4}) \hat{\mathbf{y}} - a(z_5 - \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{27}$	$(2x_5 - z_5) \mathbf{a}_1 + z_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$az_5 \hat{\mathbf{x}} + ax_5 \hat{\mathbf{y}} + ax_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{28}$	$-(2x_5 + z_5 - \frac{1}{2}) \mathbf{a}_1 + z_5 \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$az_5 \hat{\mathbf{x}} - a(x_5 - \frac{1}{4}) \hat{\mathbf{y}} - a(x_5 - \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{29}$	$z_5 \mathbf{a}_1 + (2x_5 - z_5) \mathbf{a}_2 - (2x_5 + z_5 - \frac{1}{2}) \mathbf{a}_3$	$=$	$-a(z_5 - \frac{1}{4}) \hat{\mathbf{x}} - a(x_5 - \frac{1}{4}) \hat{\mathbf{y}} + ax_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{30}$	$z_5 \mathbf{a}_1 - (2x_5 + z_5 - \frac{1}{2}) \mathbf{a}_2 + (2x_5 - z_5) \mathbf{a}_3$	$=$	$-a(z_5 - \frac{1}{4}) \hat{\mathbf{x}} + ax_5 \hat{\mathbf{y}} - a(x_5 - \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{31}$	$z_5 \mathbf{a}_1 + (2x_5 - z_5) \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$ax_5 \hat{\mathbf{x}} + az_5 \hat{\mathbf{y}} + ax_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{32}$	$z_5 \mathbf{a}_1 - (2x_5 + z_5 - \frac{1}{2}) \mathbf{a}_2 + z_5 \mathbf{a}_3$	$=$	$-a(x_5 - \frac{1}{4}) \hat{\mathbf{x}} + az_5 \hat{\mathbf{y}} - a(x_5 - \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{33}$	$-(2x_5 + z_5 - \frac{1}{2}) \mathbf{a}_1 + z_5 \mathbf{a}_2 + (2x_5 - z_5) \mathbf{a}_3$	$=$	$ax_5 \hat{\mathbf{x}} - a(z_5 - \frac{1}{4}) \hat{\mathbf{y}} - a(x_5 - \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{34}$	$(2x_5 - z_5) \mathbf{a}_1 + z_5 \mathbf{a}_2 - (2x_5 + z_5 - \frac{1}{2}) \mathbf{a}_3$	$=$	$-a(x_5 - \frac{1}{4}) \hat{\mathbf{x}} - a(z_5 - \frac{1}{4}) \hat{\mathbf{y}} + ax_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{35}$	$-z_5 \mathbf{a}_1 - z_5 \mathbf{a}_2 + (2x_5 + z_5 + \frac{1}{2}) \mathbf{a}_3$	$=$	$a(x_5 + \frac{1}{4}) \hat{\mathbf{x}} + a(x_5 + \frac{1}{4}) \hat{\mathbf{y}} - az_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{36}$	$-z_5 \mathbf{a}_1 - z_5 \mathbf{a}_2 - (2x_5 - z_5) \mathbf{a}_3$	$=$	$-ax_5 \hat{\mathbf{x}} - ax_5 \hat{\mathbf{y}} - az_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{37}$	$-(2x_5 - z_5) \mathbf{a}_1 + (2x_5 + z_5 + \frac{1}{2}) \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$a(x_5 + \frac{1}{4}) \hat{\mathbf{x}} - ax_5 \hat{\mathbf{y}} + a(z_5 + \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{38}$	$(2x_5 + z_5 + \frac{1}{2}) \mathbf{a}_1 - (2x_5 - z_5) \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$-ax_5 \hat{\mathbf{x}} + a(x_5 + \frac{1}{4}) \hat{\mathbf{y}} + a(z_5 + \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{39}$	$-(2x_5 - z_5) \mathbf{a}_1 - z_5 \mathbf{a}_2 + (2x_5 + z_5 + \frac{1}{2}) \mathbf{a}_3$	$=$	$a(x_5 + \frac{1}{4}) \hat{\mathbf{x}} + a(z_5 + \frac{1}{4}) \hat{\mathbf{y}} - ax_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{40}$	$(2x_5 + z_5 + \frac{1}{2}) \mathbf{a}_1 - z_5 \mathbf{a}_2 - (2x_5 - z_5) \mathbf{a}_3$	$=$	$-ax_5 \hat{\mathbf{x}} + a(z_5 + \frac{1}{4}) \hat{\mathbf{y}} + a(x_5 + \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{41}$	$-z_5 \mathbf{a}_1 - (2x_5 - z_5) \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$-ax_5 \hat{\mathbf{x}} - az_5 \hat{\mathbf{y}} - ax_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{42}$	$-z_5 \mathbf{a}_1 + (2x_5 + z_5 + \frac{1}{2}) \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$a(x_5 + \frac{1}{4}) \hat{\mathbf{x}} - az_5 \hat{\mathbf{y}} + a(x_5 + \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{43}$	$-z_5 \mathbf{a}_1 - (2x_5 - z_5) \mathbf{a}_2 + (2x_5 + z_5 + \frac{1}{2}) \mathbf{a}_3$	$=$	$a(z_5 + \frac{1}{4}) \hat{\mathbf{x}} + a(x_5 + \frac{1}{4}) \hat{\mathbf{y}} - ax_5 \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{44}$	$-z_5 \mathbf{a}_1 + (2x_5 + z_5 + \frac{1}{2}) \mathbf{a}_2 - (2x_5 - z_5) \mathbf{a}_3$	$=$	$a(z_5 + \frac{1}{4}) \hat{\mathbf{x}} - ax_5 \hat{\mathbf{y}} + a(x_5 + \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{45}$	$(2x_5 + z_5 + \frac{1}{2}) \mathbf{a}_1 - z_5 \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$-az_5 \hat{\mathbf{x}} + a(x_5 + \frac{1}{4}) \hat{\mathbf{y}} + a(x_5 + \frac{1}{4}) \hat{\mathbf{z}}$	(96g)	Al II
$\mathbf{B}_{46}$	$-(2x_5 - z_5) \mathbf{a}_1 - z_5 \mathbf{a}_2 - z_5 \mathbf{a}_3$	$=$	$-az_5 \hat{\mathbf{x}} - ax_5 \hat{\mathbf{y}} - ax_5 \hat{\mathbf{z}}$	(96g)	Al II

## References

- [1] S. Samson, *The Crystal Structure of the Intermetallic Compound  $Mg_3Cr_2Al_{18}$* , Acta Cryst. **11** (1958), doi:10.1107/S0365110X58002425.

## Found in

- [1] S. Samson, *The Crystal Structure of the Intermetallic Compound ZrZn<sub>22</sub>*, Acta Cryst. **14** (1961), doi:10.1107/S0365110X61003600.