

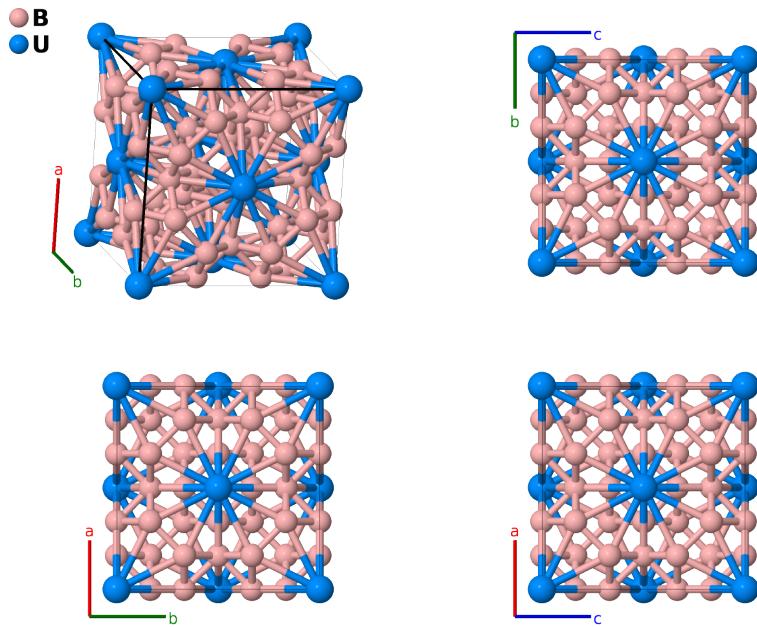
# UB<sub>12</sub> ( $D2_f$ ) Structure: A12B\_cF52\_225\_h\_b-001

This structure originally had the label A12B\_cF52\_225\_i\_a. Calls to that address will be redirected here.

Cite this page as: M. J. Mehl, D. Hicks, C. Toher, O. Levy, R. M. Hanson, G. Hart, and S. Curtarolo, *The AFLOW Library of Crystallographic Prototypes: Part 1*, Comput. Mater. Sci. **136**, S1-828 (2017). doi: 10.1016/j.commatsci.2017.01.017

<https://aflow.org/p/JTKK>

[https://aflow.org/p/A12B\\_cF52\\_225\\_h\\_b-001](https://aflow.org/p/A12B_cF52_225_h_b-001)



<b>Prototype</b>	B <sub>12</sub> U
<b>AFLOW prototype label</b>	A12B_cF52_225_h_b-001
<b>Strukturbericht designation</b>	$D2_f$
<b>ICSD</b>	24705
<b>Pearson symbol</b>	cF52
<b>Space group number</b>	225
<b>Space group symbol</b>	$Fm\bar{3}m$
<b>AFLOW prototype command</b>	aflow --proto=A12B_cF52_225_h_b-001 --params=a,y2

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## Other compounds with this structure

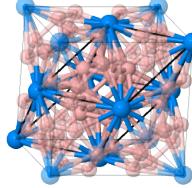
DyB<sub>12</sub>, ErB<sub>12</sub>, LuB<sub>12</sub>, ThB<sub>12</sub>, TmB<sub>12</sub>, YB<sub>12</sub>, YbB<sub>12</sub>, ZrB<sub>12</sub>, (Th<sub>0.93</sub>Zr<sub>0.07</sub>)B<sub>12</sub>

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## Face-centered Cubic primitive vectors

$$\begin{aligned}
 \mathbf{a}_1 &= \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}} \\
 \mathbf{a}_2 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{z}} \\
 \mathbf{a}_3 &= \frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}}
 \end{aligned}$$

$\text{a}_3/2$   
 $\text{a}_1$



## Basis vectors

	Lattice coordinates	Cartesian coordinates	Wyckoff position	Atom type
$\mathbf{B}_1$ =	$\frac{1}{2}\mathbf{a}_1 + \frac{1}{2}\mathbf{a}_2 + \frac{1}{2}\mathbf{a}_3$	$\frac{1}{2}a\hat{\mathbf{x}} + \frac{1}{2}a\hat{\mathbf{y}} + \frac{1}{2}a\hat{\mathbf{z}}$	(4b)	U I
$\mathbf{B}_2$ =	$2y_2\mathbf{a}_1$	$ay_2\hat{\mathbf{y}} + ay_2\hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_3$ =	$2y_2\mathbf{a}_2 - 2y_2\mathbf{a}_3$	$-ay_2\hat{\mathbf{y}} + ay_2\hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_4$ =	$-2y_2\mathbf{a}_2 + 2y_2\mathbf{a}_3$	$ay_2\hat{\mathbf{y}} - ay_2\hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_5$ =	$-2y_2\mathbf{a}_1$	$-ay_2\hat{\mathbf{y}} - ay_2\hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_6$ =	$2y_2\mathbf{a}_2$	$ay_2\hat{\mathbf{x}} + ay_2\hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_7$ =	$-2y_2\mathbf{a}_1 + 2y_2\mathbf{a}_3$	$ay_2\hat{\mathbf{x}} - ay_2\hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_8$ =	$2y_2\mathbf{a}_1 - 2y_2\mathbf{a}_3$	$-ay_2\hat{\mathbf{x}} + ay_2\hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_9$ =	$-2y_2\mathbf{a}_2$	$-ay_2\hat{\mathbf{x}} - ay_2\hat{\mathbf{z}}$	(48h)	B I
$\mathbf{B}_{10}$ =	$2y_2\mathbf{a}_3$	$ay_2\hat{\mathbf{x}} + ay_2\hat{\mathbf{y}}$	(48h)	B I
$\mathbf{B}_{11}$ =	$2y_2\mathbf{a}_1 - 2y_2\mathbf{a}_2$	$-ay_2\hat{\mathbf{x}} + ay_2\hat{\mathbf{y}}$	(48h)	B I
$\mathbf{B}_{12}$ =	$-2y_2\mathbf{a}_1 + 2y_2\mathbf{a}_2$	$ay_2\hat{\mathbf{x}} - ay_2\hat{\mathbf{y}}$	(48h)	B I
$\mathbf{B}_{13}$ =	$-2y_2\mathbf{a}_3$	$-ay_2\hat{\mathbf{x}} - ay_2\hat{\mathbf{y}}$	(48h)	B I

## References

- [1] P. Blum and F. Bertaut, *Contribution à l'Étude des Borures à Teneur Élevée en Bore*, Acta Cryst. **7**, 81–86 (1954), doi:10.1107/S0365110X54000151.

## Found in

- [1] W. B. Pearson, *The Crystal Chemistry and Physics of Metals and Alloys* (Wiley Interscience, New York, London, Sydney, Tornoto, 1972).